

Detection of uranium contamination in the process of dismantling and decommissioning nuclear facilities

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- Photonic Systems Laboratory (DRT/LETI/DOPT/SISP/LASP)
- Laboratory for System Design and Modelling (DES/DER/SESI/LEMS)

Summary

Context and objectives

- Nuclear decommissioning in France
- Gaseous Diffusion Plant (UDG – Tricastin Nuclear Site)
- Technical challenges and chosen detection methods

Contamination detection modelling

- Beta particles detection
- Gamma rays detection
- Alpha particles detection
- Estimation of acquisition times and conclusions

In-situ measurements at the UDG enrichment facility

- Detectors and inspected zones
- Method for surface activity estimation
- Results and comparison between methods

Conclusions and perspectives





1 ■ Context and objectives



Nuclear decommissioning in France

35 facilities awaiting or undergoing decommissioning in France

Reasons for shutdown :

- Aging components (facilities built in the 1960s with an estimated 40 years of service life)
- Political decisions (nuclear power phase-out)

A lengthy process :

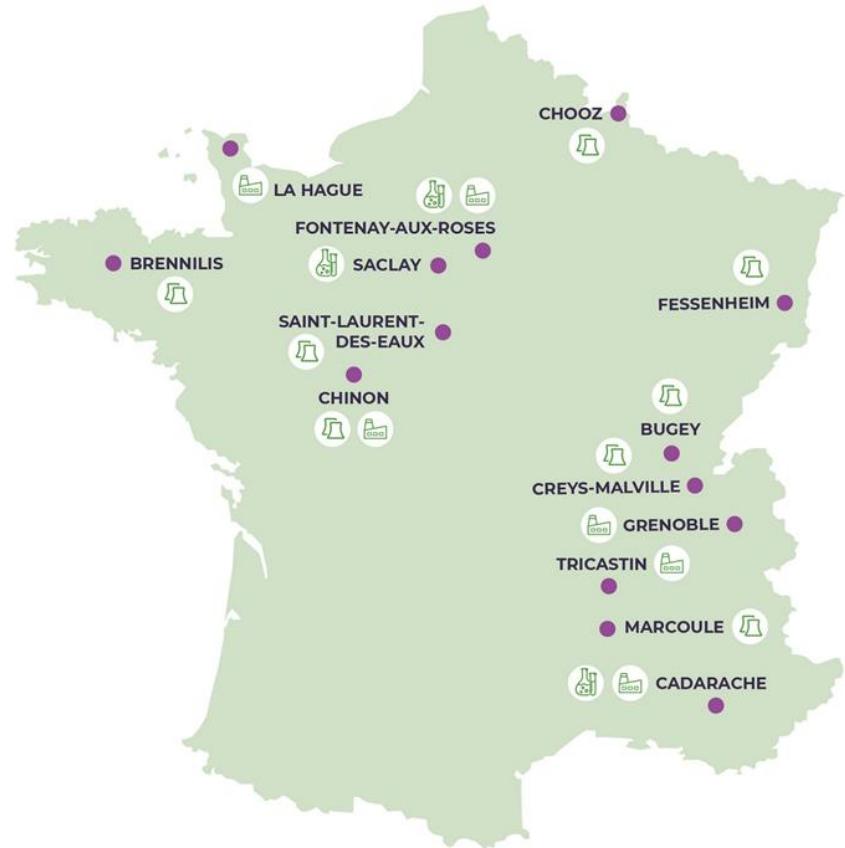
- 15 years to dismantle a Pressurized Water Reactor (PWR)
- 10 years for an enrichment facility

A costly procedure :

- 350-400 million euros to dismantle a PWR

Technical challenges may extend decommissioning time depending on the facility

A critical need to develop new technologies aiming to reduce times and costs of the decommissioning process arises



Reactor Factory Laboratory and research reactor

Map of shutdown nuclear facilities in France
(source : French Nuclear Safety Authority)

The Gaseous Diffusion Plant (UDG) - Tricastin Nuclear Site

Legacy uranium enrichment facility

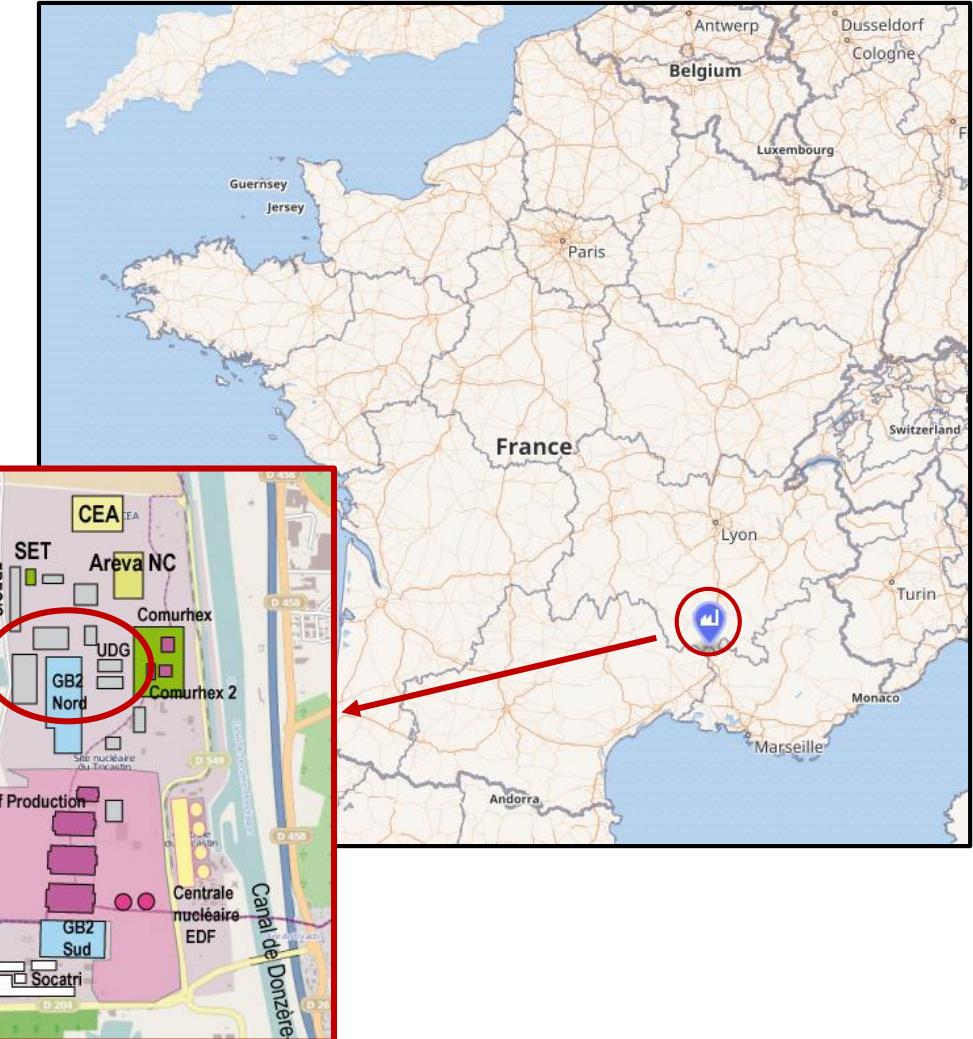
→ increase of ^{235}U proportion via gaseous diffusion process

Characteristics of the UDG enrichment facility

- Lifetime = 40 years
- Total surface $\approx 700\,000 \text{ m}^2$
- 4 plants
 - Low enrichment plant (LP) $> 0,8 \% \text{ }^{235}\text{U}$
 - Medium enrichment plant (MP) $> 1 \% \text{ }^{235}\text{U}$
 - High enrichment plant (HP) $> 3 \% \text{ }^{235}\text{U}$
 - Very high enrichment plant (VHP1, VHP2, VHP3) $> 20 \% \text{ }^{235}\text{U}$



Age of the facility and enrichment level play an important role on the type of particles emitted



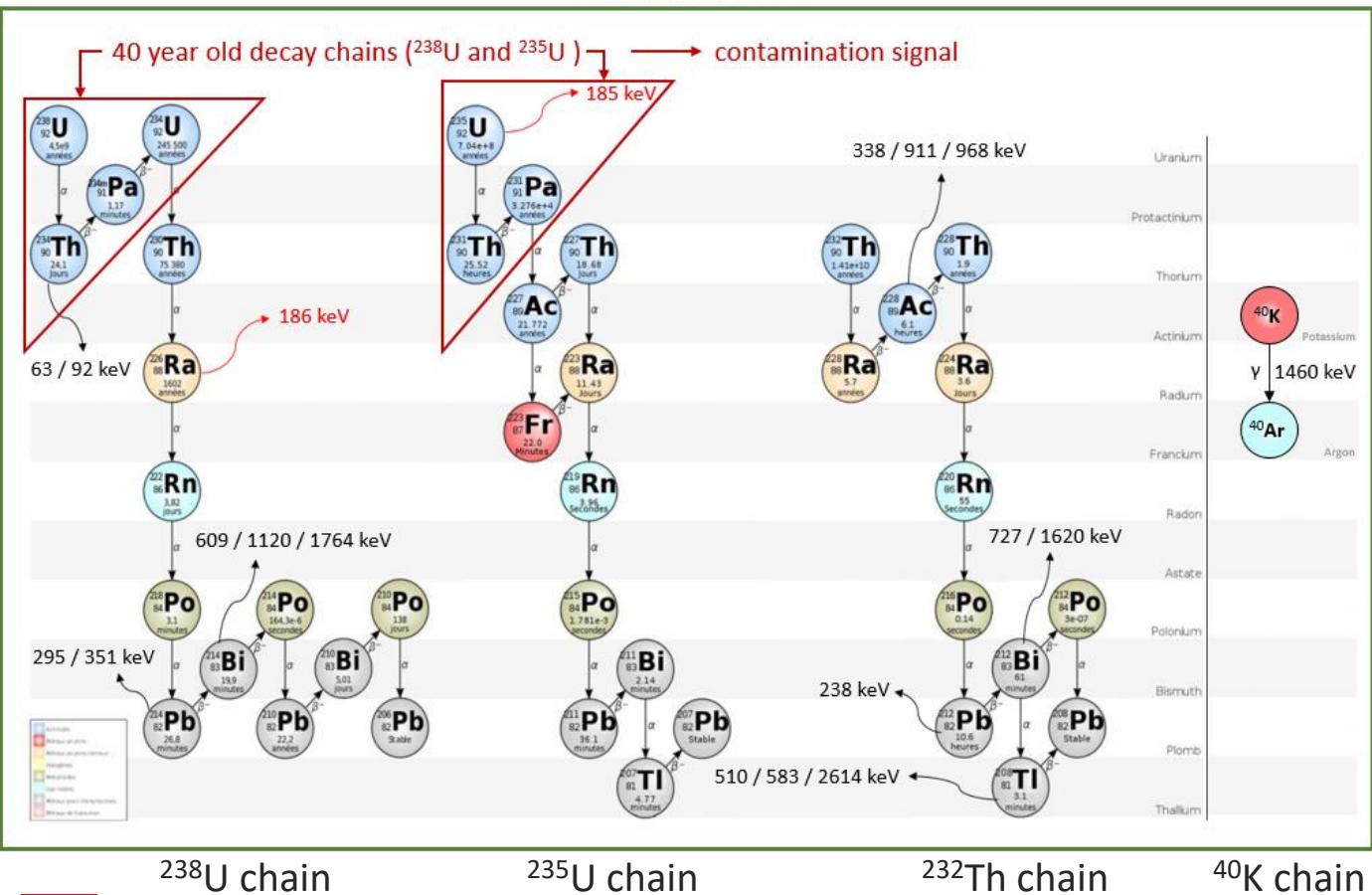
UDG and Tricastin nuclear site location in France

Technical challenges and chosen detection methods

Two main challenges

Discriminate the contamination signal from the background noise

Background noise



Detect very low activity contaminations with fast counting times

- Total area of 700 000 m² ⇒ measuring time as low as possible
- Decommissioning threshold ⇒ 0,4 Bq.cm⁻² (α activity)



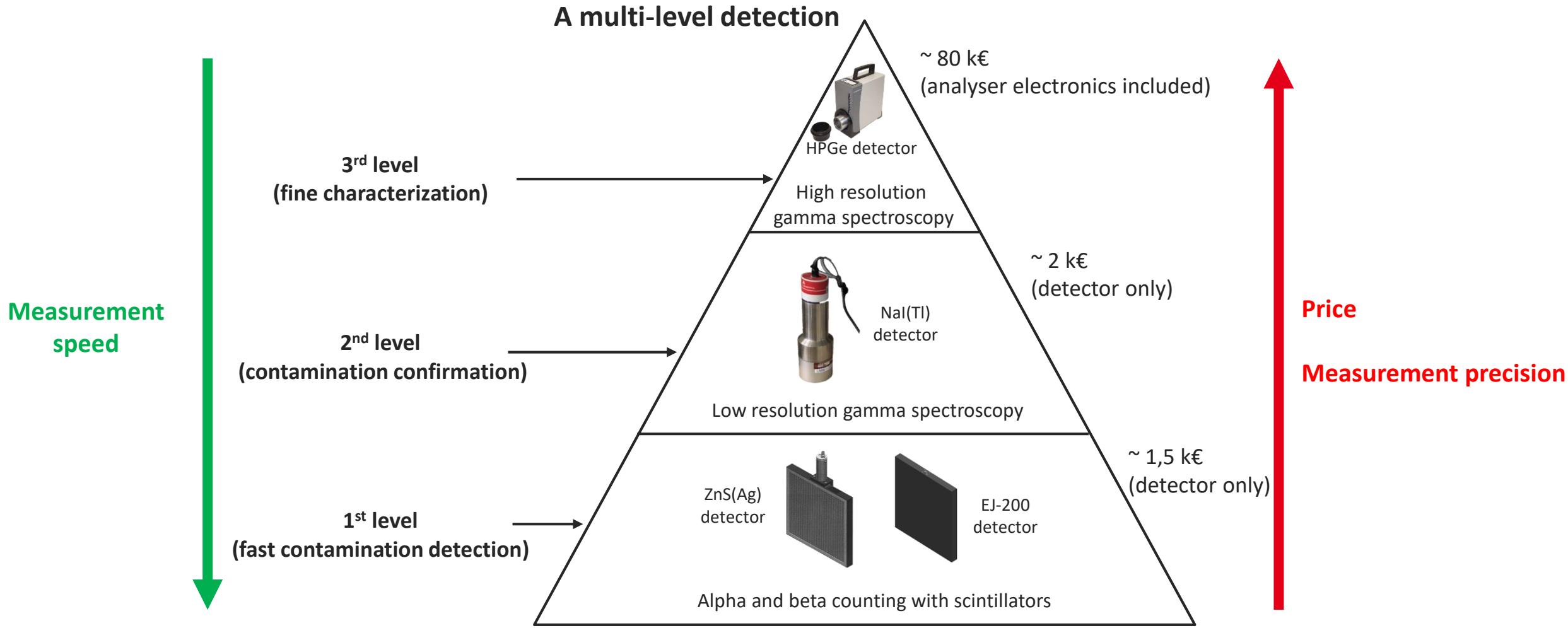
UDG corridor

Source : <https://www.francetnp.gouv.fr/Pierrelatte-l-usine-d>



Gamma and beta measurements sensitive to enrichment

Technical challenges and chosen detection methods

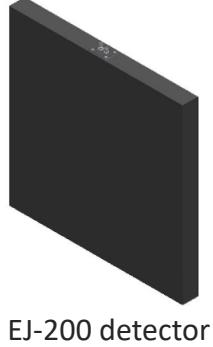




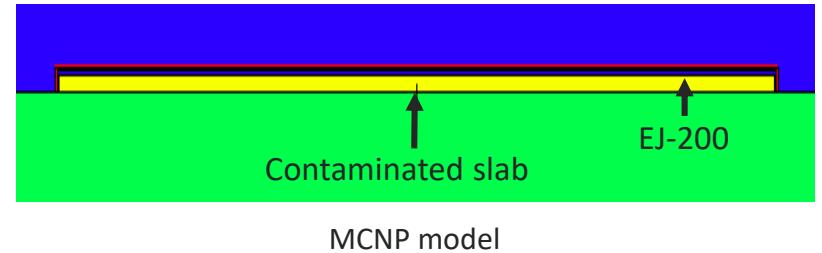
2 ■ Contamination detection modelling

Beta particle detection

Detect the beta particles produced by ^{234m}Pa (^{238}U) within a region of interest between 300 and 2500 keV



Simulation of a EJ-200 detector over a contaminated concrete slab



EJ-200 detector

MCNP model

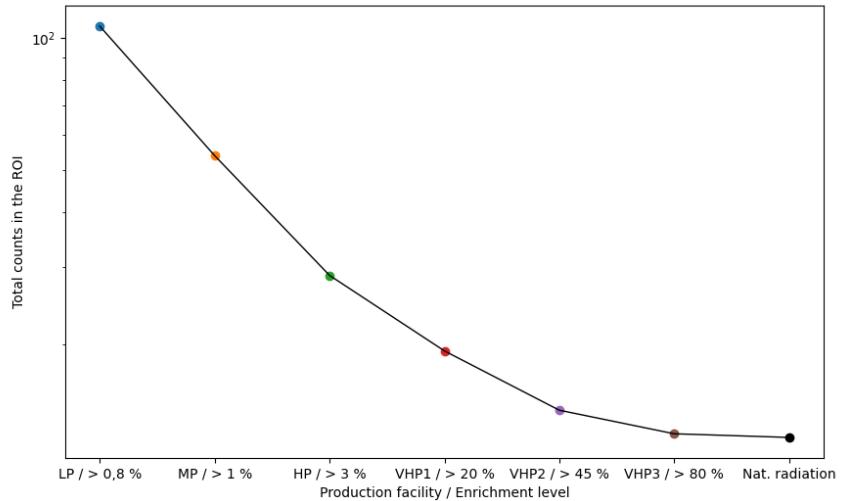
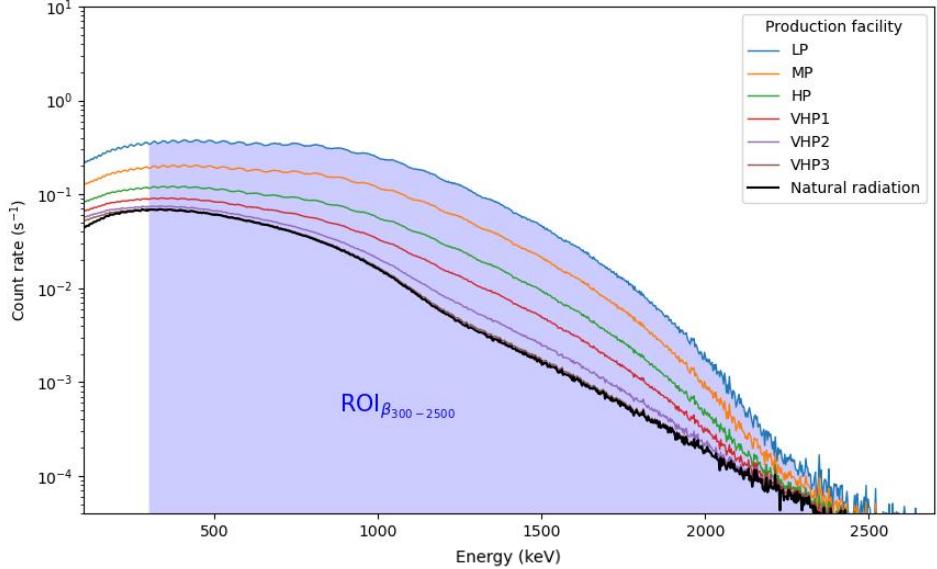
- EJ200 detector size : $32 \times 32 \times 0,4 \text{ cm}^3$
- 1 mm thick aluminum cover surrounds the scintillator
- Contamination characteristics
 - 1000 Bq distributed homogenously inside a $32 \times 32 \times 0,1 \text{ cm}^3$ concrete slab (1 Bq.cm^{-2})



Moderate attenuation in concrete. Detection possible up to mm deep

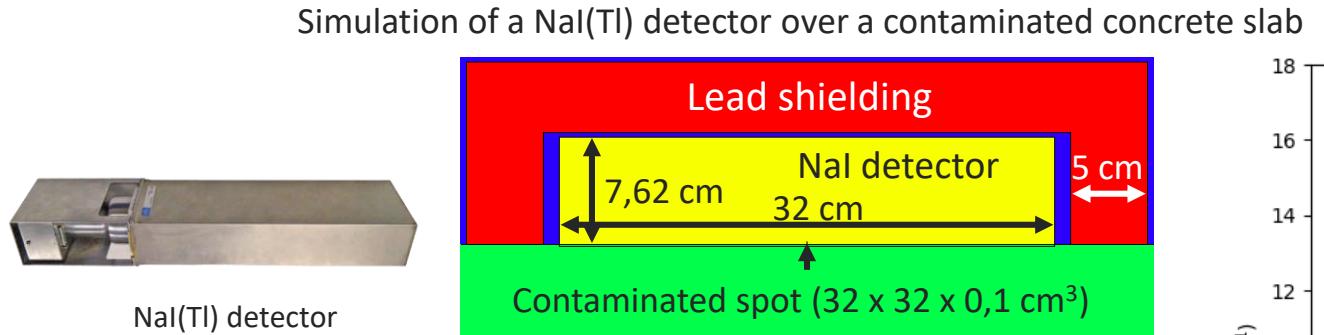


Highly dependent on ^{235}U enrichment percentage

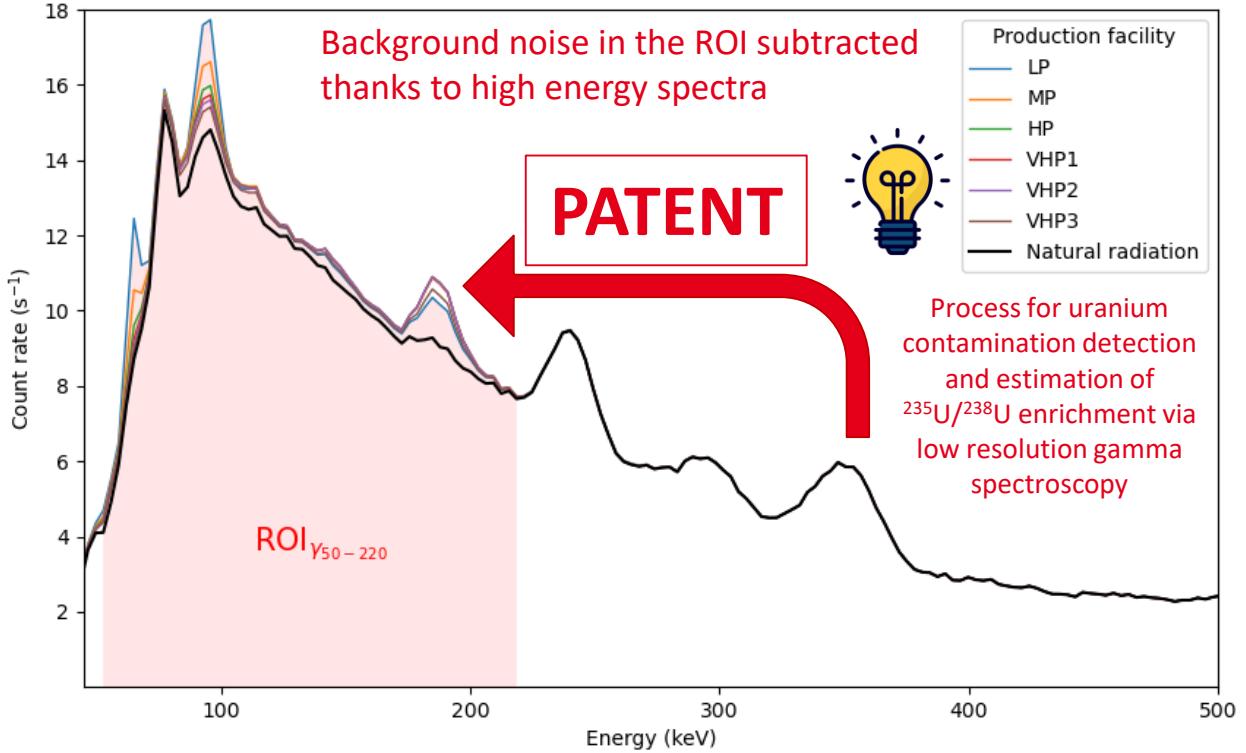


Gamma ray detection

Detect the gamma rays produced by both ^{238}U and ^{235}U isotopes within a region of interest between 50 and 220 keV



- NaI(Tl) detector size : $32 \times 32 \times 7,62 \text{ cm}^3$
- Natural radiation emitted by a $100 \times 100 \times 30 \text{ cm}^3$ concrete block (density : $2,35 \text{ g.cm}^{-3}$)
- Contamination characteristics
 - 1000 Bq distributed homogenously inside a $32 \times 32 \times 0,1 \text{ cm}^3$ concrete slab (1 Bq.cm^{-2})



 Able to detect deeper contaminations, patented method allows to determine the contamination enrichment

 Slower detection because of the high sensitivity to background noise, moderate dependence to enrichment

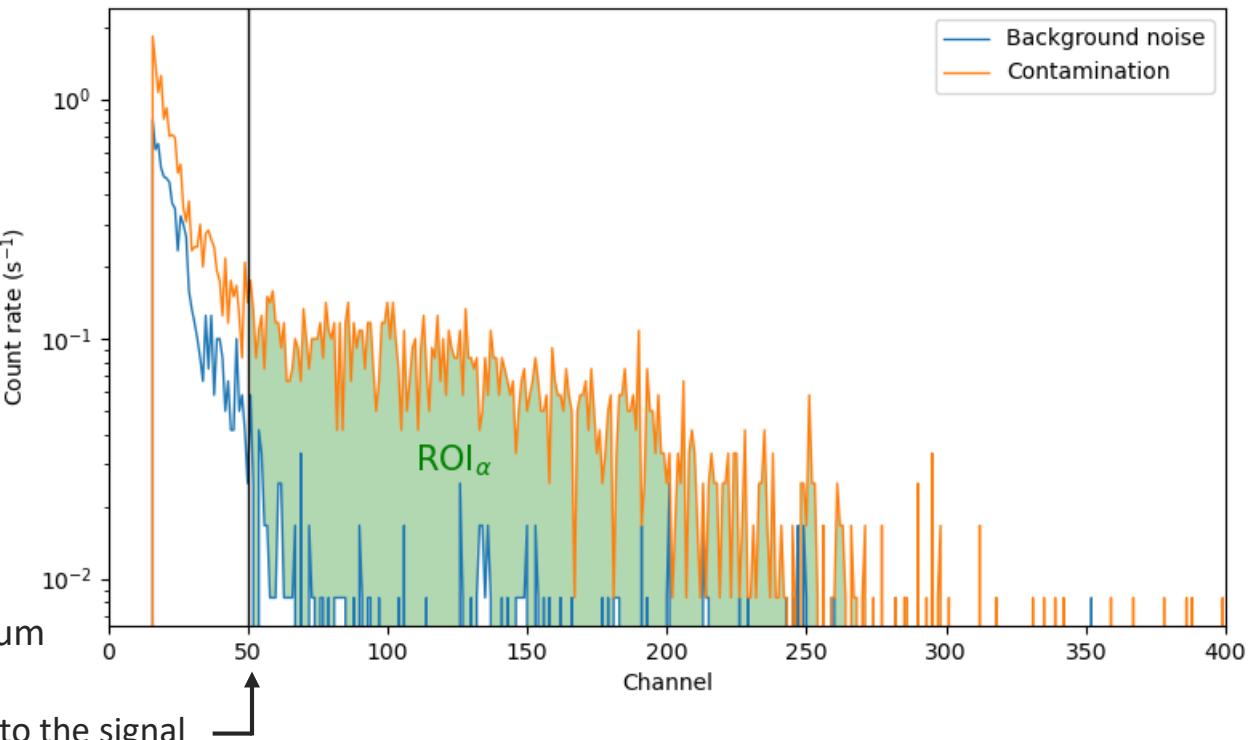
Alpha particle detection

Detect the alpha particles emitted by the three uranium isotopes ^{238}U , ^{235}U and ^{234}U

ZnS(Ag) detector



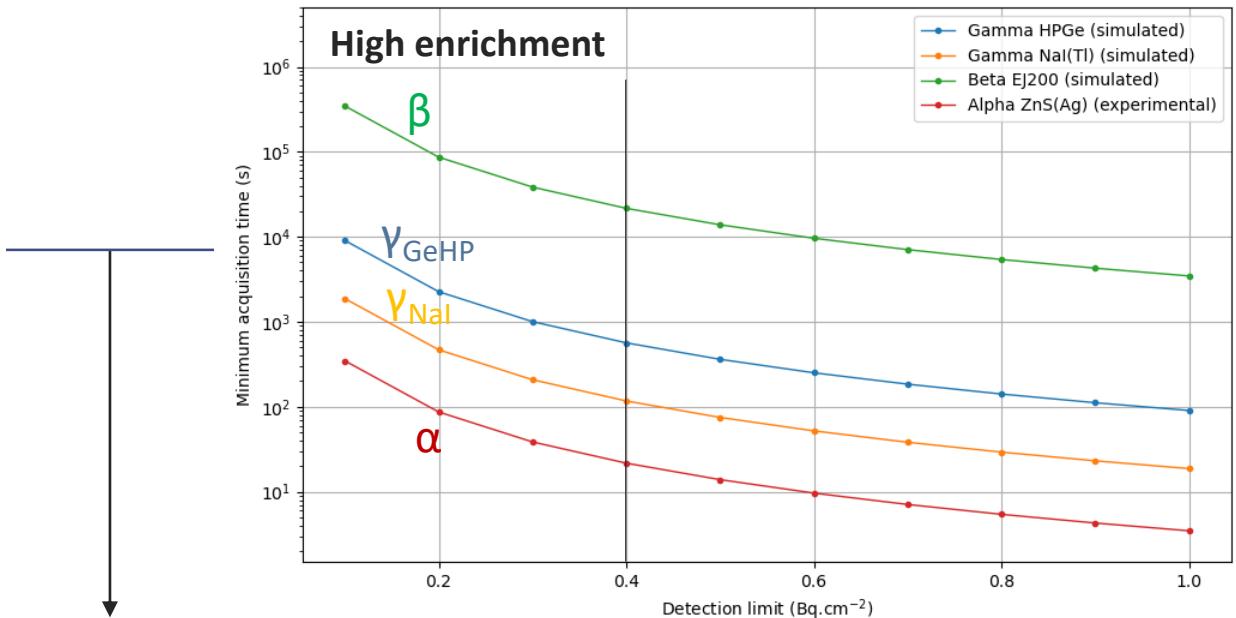
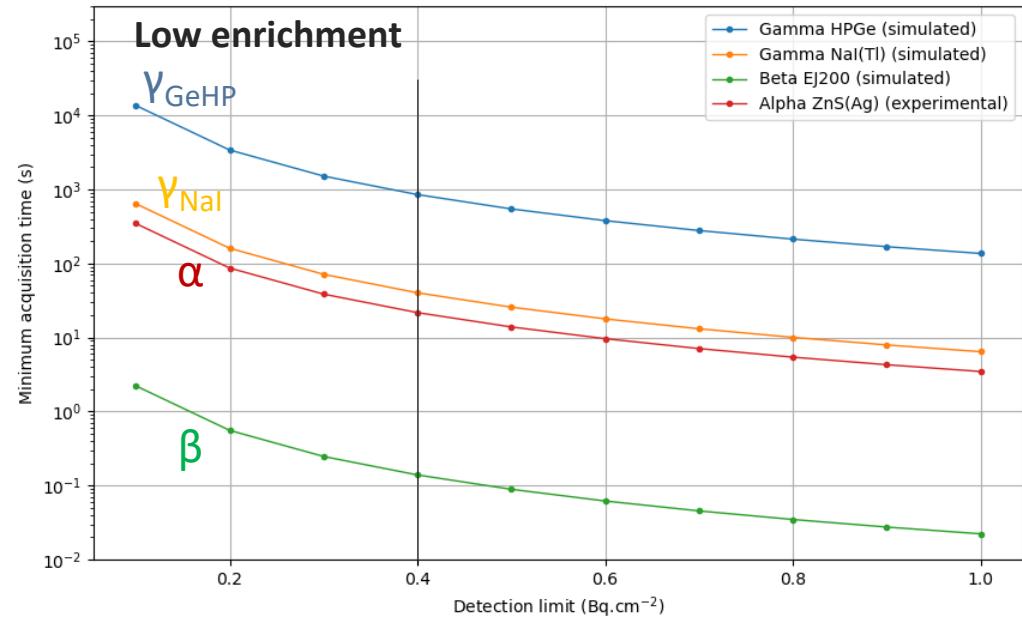
- ZnS(Ag) detector size : $25 \times 25 \times 2,2 \text{ cm}^3$ ($23 \times 23 \text{ cm}^2$ window)
- Measurements over two concrete blocks :
One block is **clean**, the other is **contaminated** with uranium



 Independent of ^{235}U enrichment, counting rate directly proportional to the contamination activity

 Can only detect surface contamination. High dependence on the state of the inspected area (dust, rough surface)

Estimation of acquisition times and conclusions



²³⁵ U enrichment	Minimum acquisition time to reach the 0,4 Bq.cm ⁻² DL			
	Gamma HPGe	Gamma NaI(Tl)	Beta EJ200	Alpha ZnS(Ag)
Low enrichment	854 s	40 s	14 x 10 ⁻² s	22 s
High enrichment	565 s	117 s	22 x 10 ³ s	22 s



Estimations for a 1 mm deep contamination

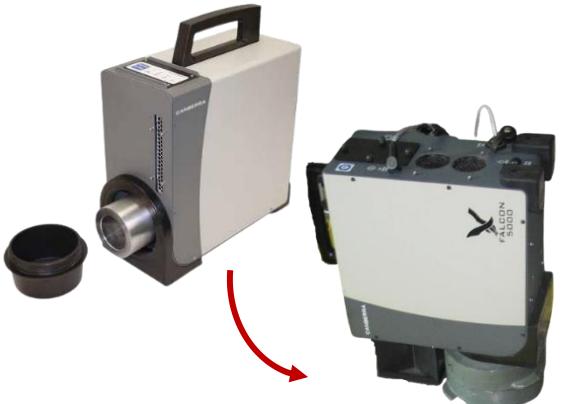
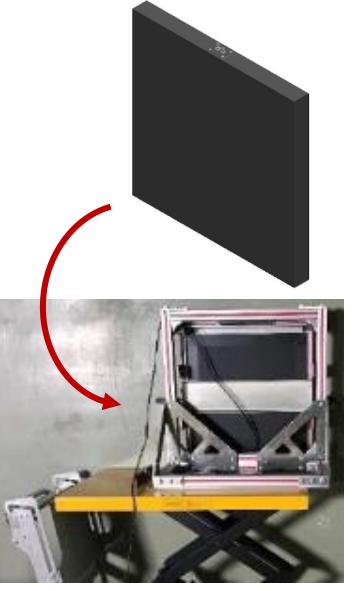
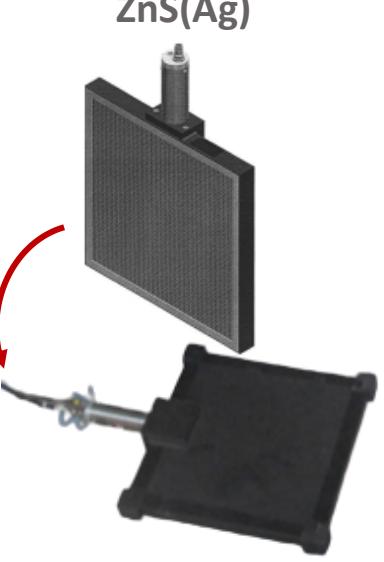
Detection time estimation
 Low enrichment \Rightarrow beta < alpha < gamma
 High enrichment \Rightarrow alpha < gamma < beta



In-situ measurements at the UDG enrichment facility

3 ■

Detectors and inspected zones

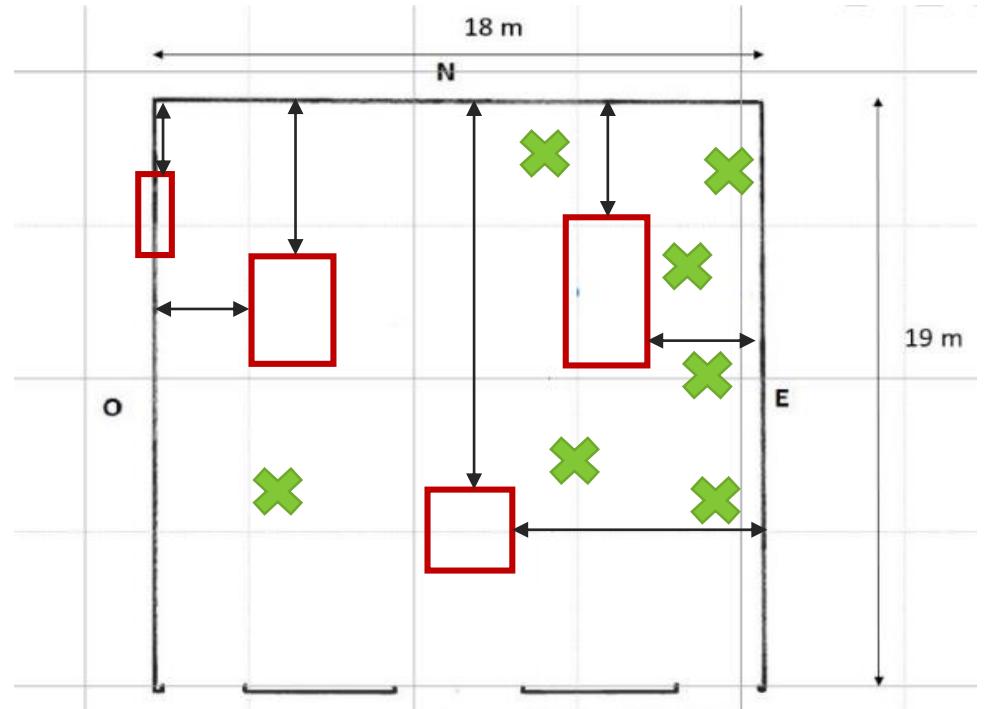
γ spectrometry		β counting	α counting
High resolution	Low resolution	Scintillation detector	Scintillation detector
Falcon 5000 	Nal 3" x 3" 	 $S_{\text{det}} = 2430 \text{ cm}^2$ or 625 cm^2 <ul style="list-style-type: none"> Plastic scintillator EJ-200 Aluminized PET entrance window $D_{\text{detector - floor}} = 7,5 \text{ mm}$ 	 $S_{\text{det}} = 529 \text{ cm}^2$ <ul style="list-style-type: none"> ZnS(Ag) layer spread over a plastic sheet (EJ-440) $D_{\text{detector - floor}} = 2 \text{ mm}$ Aluminized PET entrance window protected by a steel mesh
$S_{\text{det}} = 28 \text{ cm}^2$ <ul style="list-style-type: none"> Semi-conductor detector (HPGe) Cylindrical crystal : 3 cm radius and 3 cm thick Electrical cooling Lead disks surround the detector entrance window 	$S_{\text{det}} \approx 530 \text{ cm}^2$ <ul style="list-style-type: none"> Scintillation detector Nal(Tl) Cylindrical crystal : 7,62 cm diameter and 7,62 cm thick $D_{\text{detector - floor}} = 10,2 \text{ cm}$ Lead disks around the detector + forklift for transportation 	Detection surface compatible between the detection methods	

Detectors and inspected areas

Low enrichment plant

Counting time per point : $\alpha = 2$ to 3 min | $\beta = 2$ min | $\gamma = 15$ min

→ Measurement areas composed of multiple 25×25 cm 2 measurement points



Total area analyzed $\approx 13,3$ m 2

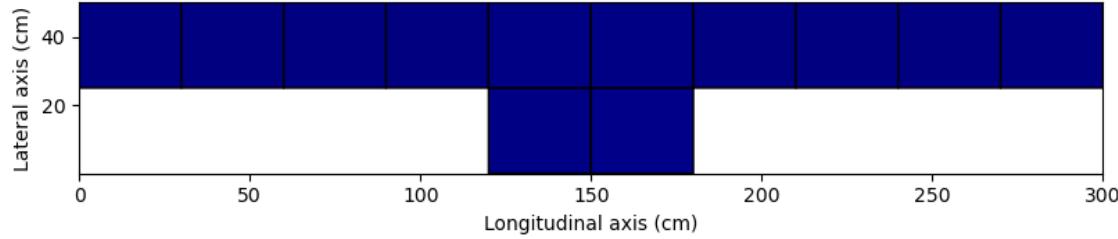
- low enrichment plant $\Rightarrow 9$ m 2
- high enrichment plant $\Rightarrow 4,3$ m 2



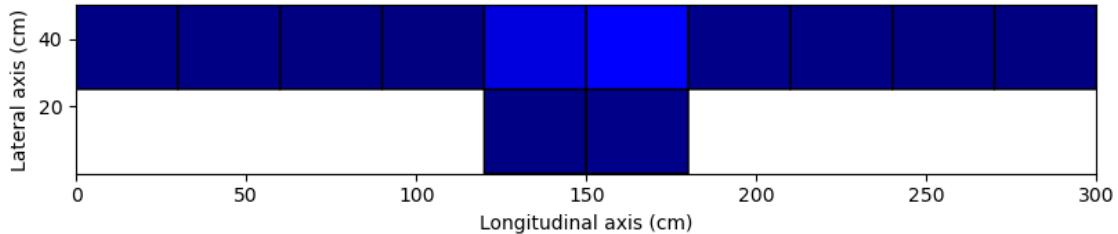
Results for the low enrichment plant

Hypothesis of contamination migration depth : **10 µm**

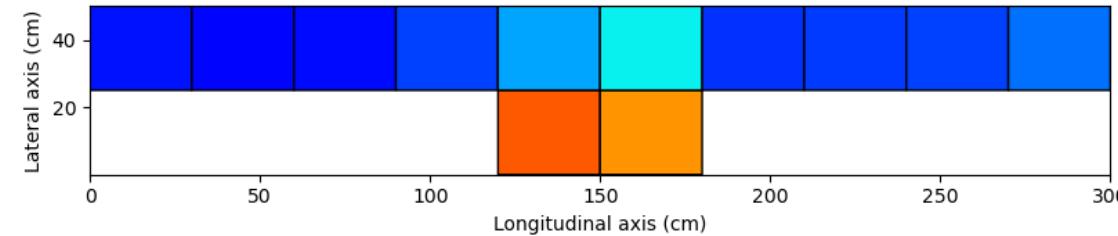
α



β



γ

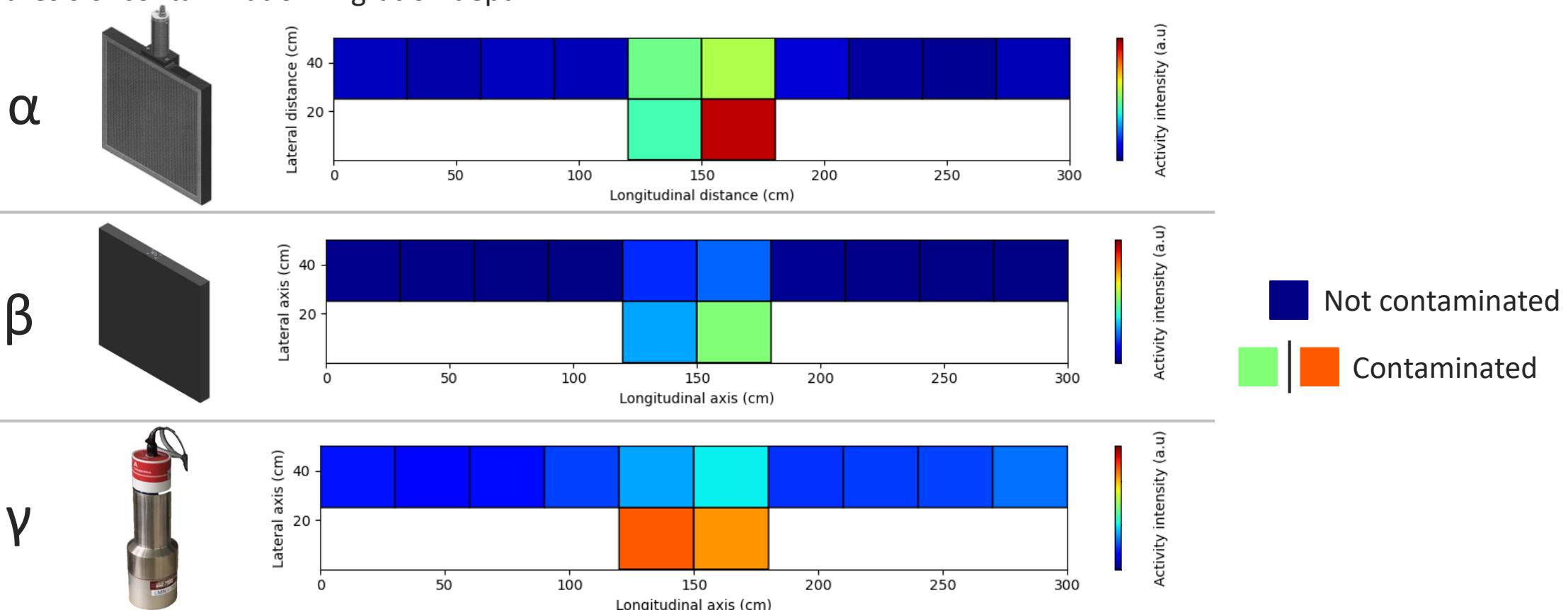


Not contaminated
Contaminated

Activity measured by α and β techniques $< 0.4 \text{ Bq/cm}^2$ with the hypothesis of a surface contamination ($< 10 \mu\text{m}$)

Results for the low enrichment plant

Hypothesis of contamination migration depth : 1 mm



Contamination depth is an important factor for alpha and beta detection from 10 μ m to 1 mm

- Factor 80 for alpha detection
- Factor 2 for beta detection
- No impact for gamma detection

Background noise corrections are also an important aspect

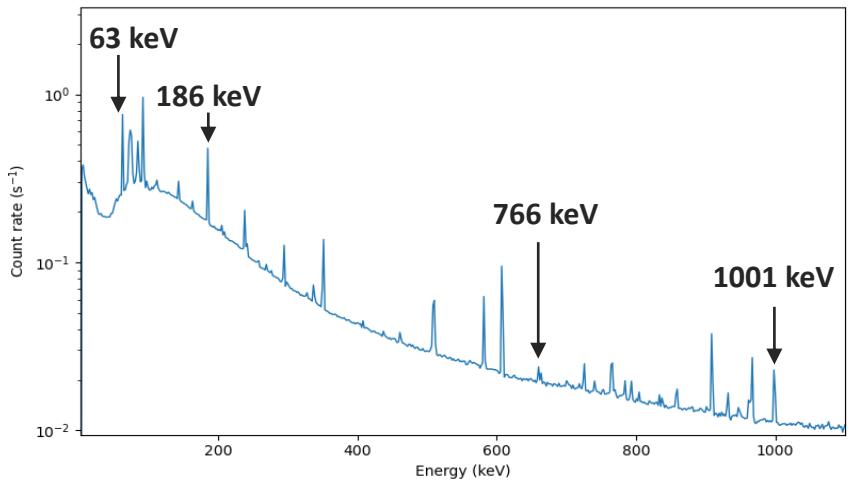
- Activity estimations rely on correct background subtractions
- Background fluctuates from one point to another

Uranium enrichment estimation

Measurements done over the same contaminated area

HPGe

64,3 hours measurement



Emitter isotope	Peak energies (keV)
^{238}U	63
	112
	766
	1001
	144
	163
	186
	205

Estimated surface activities :

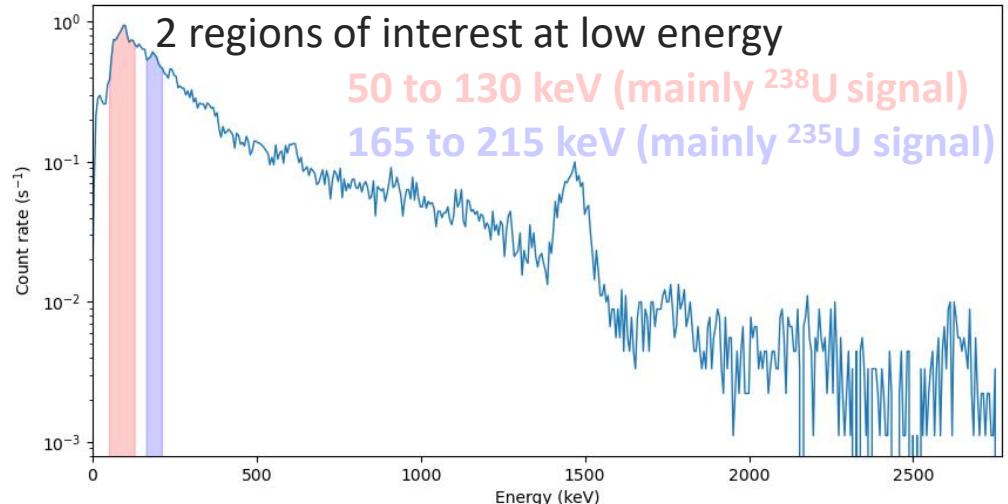
- $^{238}\text{U} : 8,7 \pm 1,1 \text{ Bq.cm}^{-2}$
- $^{235}\text{U} : 0,45 \pm 0,05 \text{ Bq.cm}^{-2}$

After conversion
to masses

$$\%_{\text{Enrichment}} = \frac{M_{235}}{M_{238} + M_{235}} = 0,79 \pm 0,07 \%$$

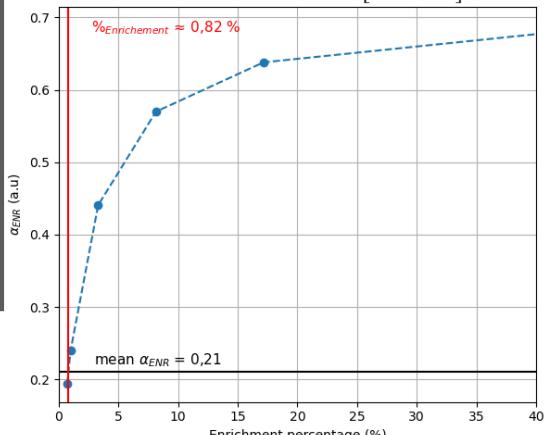
NaI(Tl)

15 minutes measurement



Calculation of a enrichment indicator using experimental data :

$$\alpha_{\text{ENR}} = \frac{C_{[165-215]}^{\text{NET}}}{C_{[50-130]}^{\text{NET}}} = \frac{C_{[165-215]}^{\text{RAW}} - C_{[165-215]}^{\text{NOISE}}}{C_{[50-130]}^{\text{RAW}} - C_{[50-130]}^{\text{NOISE}}} = 0,21$$



Calibration curve between α_{ENR} and enrichment level determined via simulation

Enrichment estimation : 0,82 % ranging from 0,62 % to 1,3 % when considering statistical uncertainties



4 Conclusions and perspectives

Conclusions and perspectives

Conclusions drawn from the measurements at the UDG enrichment plant

- The migration depth must be a well-known parameter for correct activity estimations
- The background noise must be finely characterized for a good correction of raw counts
- Measurement times differ depending on the method

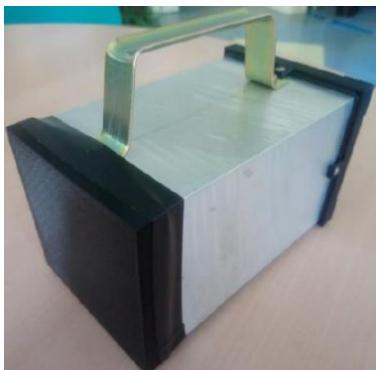
Particle detected	Impact of migration depth	Impact of ^{235}U enrichment	Impact of background noise fluctuation	In-situ measurement time (for 1000 cm 2)
Alpha	High	No impact	High	~ 1 min.
Beta	Moderate	High	High	~ 1 min.
Gamma	No impact	Moderate Estimate via patented method	High	~ 15 min.

Perspectives

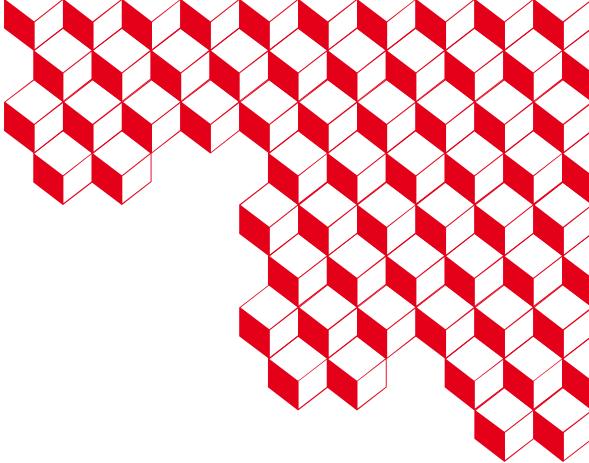
- Large NaI (~ 30 x 30 cm 2) for increased detection efficiency
- New measurements for background noise characterization (2023)
- New contamination detection tests (2024)
- Ongoing feasibility studies for uranium detection by active XRF (K and L lines)



NaI(Tl) 4"x4"



Gamma camera for x-ray imaging



Thank you for your attention

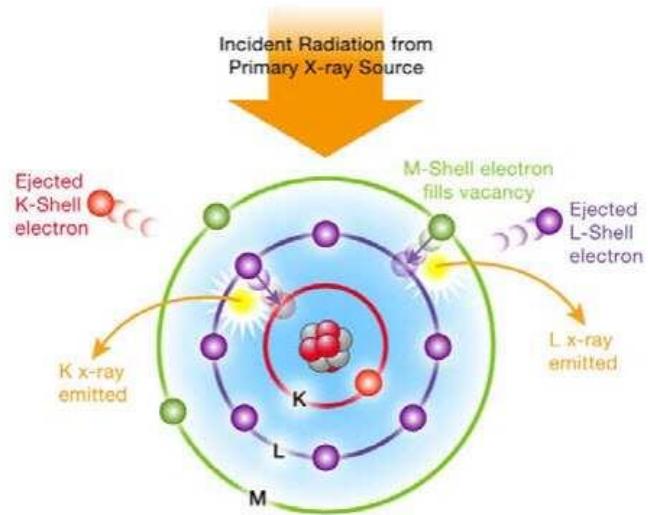


5 ■ Annexes



X-Ray fluorescence

Working principle :



Source : <https://www.thermofisher.com/blog/ask-a-scientist/what-is-xrf-x-ray-fluorescence-and-how-does-it-work/>

Uranium L-ray energies :

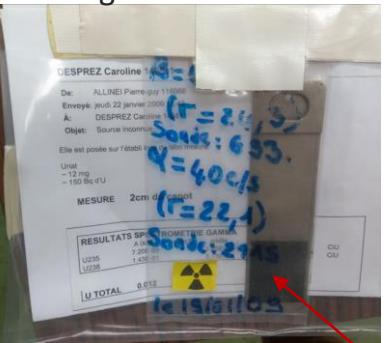
L-ray	Energy (keV)
L_I	11,6
$L_{\alpha 1}$	13,6
$L_{\beta 2}$	16,4
$L_{\beta 1}$	17,2
$L_{\gamma 1}$	20,2

Application to uranium detection :

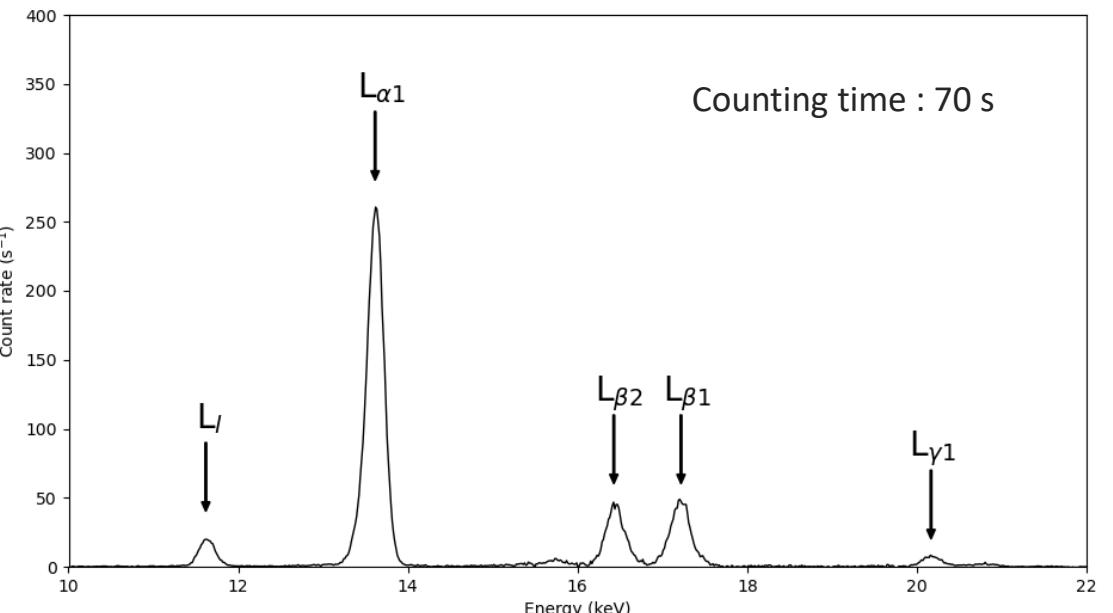
Source + detector :
Niton GOLDD XL2



Sample :
12 mg of natural uranium



Uranium distributed over a 11,1 cm² surface





Emission rates of alpha, beta and gamma radiation

Gamma, alpha and beta emission rates for a uranium contamination with a total activity of 1000 Bq

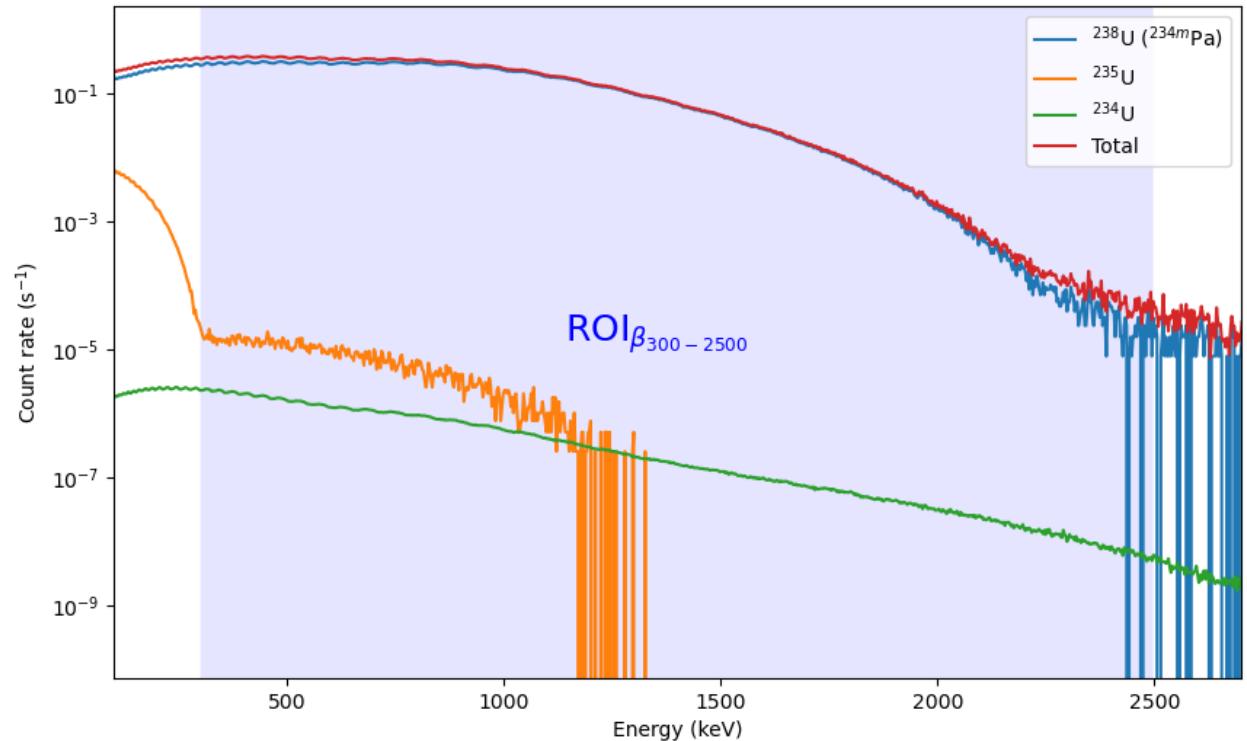
Facility	Emission rate γ (/s)				Emission rate α (/s)				Emission rate β (/s)		
	^{238}U	^{235}U	^{234}U	Total	^{238}U	^{235}U	^{234}U	Total	^{238}U	^{235}U	Total
LP	109,2	58,9	59,6	227,7	400,0	26,0	574,0	1000,0	800,0	26,0	826,0
HP	19,1	90,7	92,4	202,1	70,0	40,0	890,0	1000,0	140,0	40,0	180,0



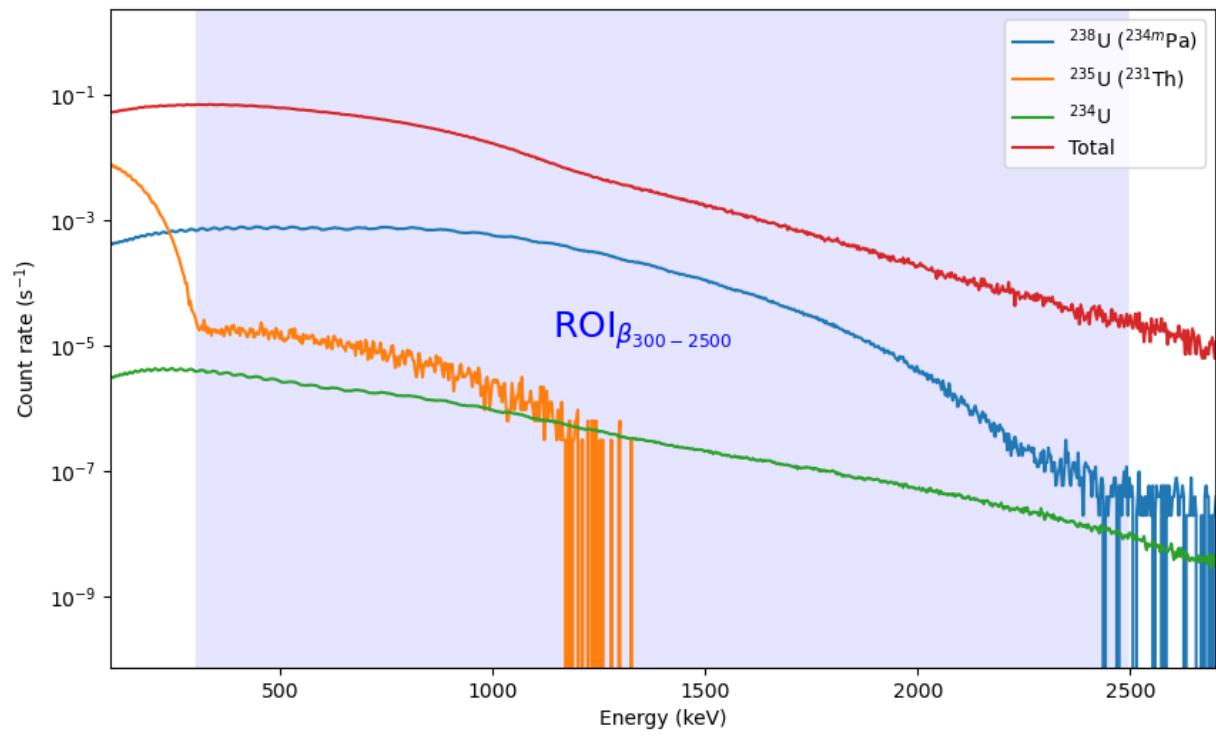
Beta particles detection

Decomposition of ^{238}U , ^{235}U and ^{234}U contributions to the signal (without natural radiation) in **1 %** and **80 % enriched contamination**

1 % enrichment



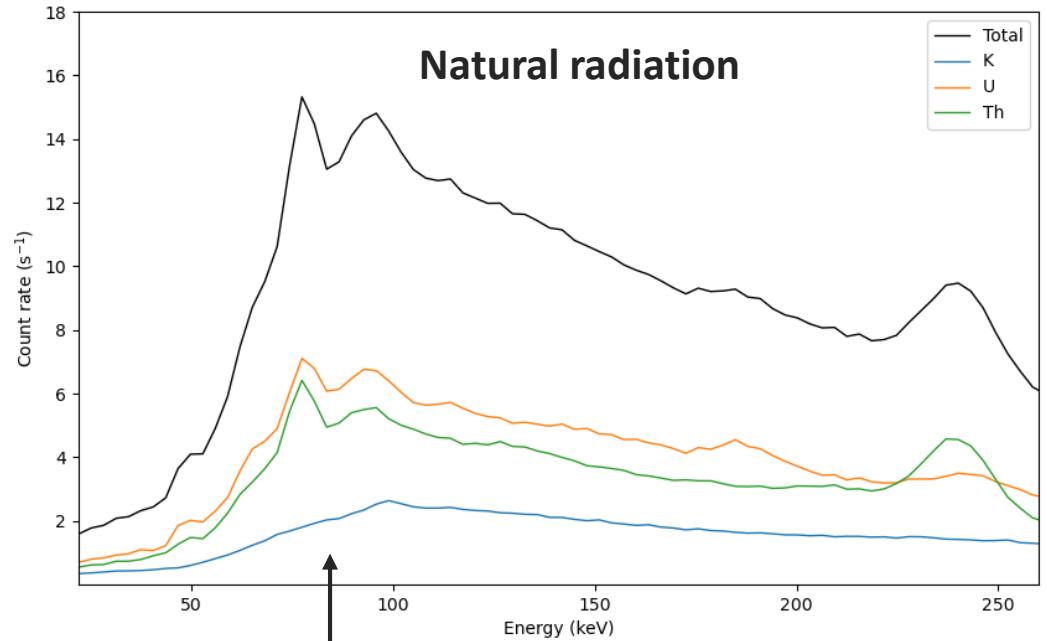
80 % enrichment



Decreased ^{238}U beta signal at 80 % enrichment



Gamma rays detection



Most natural gamma signal in the ROI comes from uranium and thorium

Natural abundances :

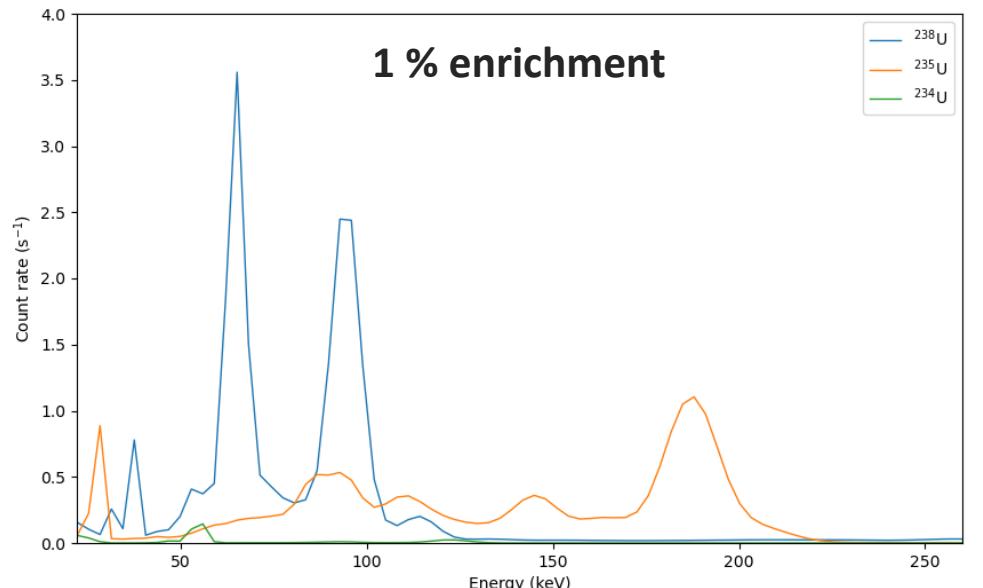
Uranium \rightarrow 99,27 % ^{238}U | 0,72 % ^{235}U | 0,005 % ^{234}U

Thorium \rightarrow virtually 100 % ^{232}Th

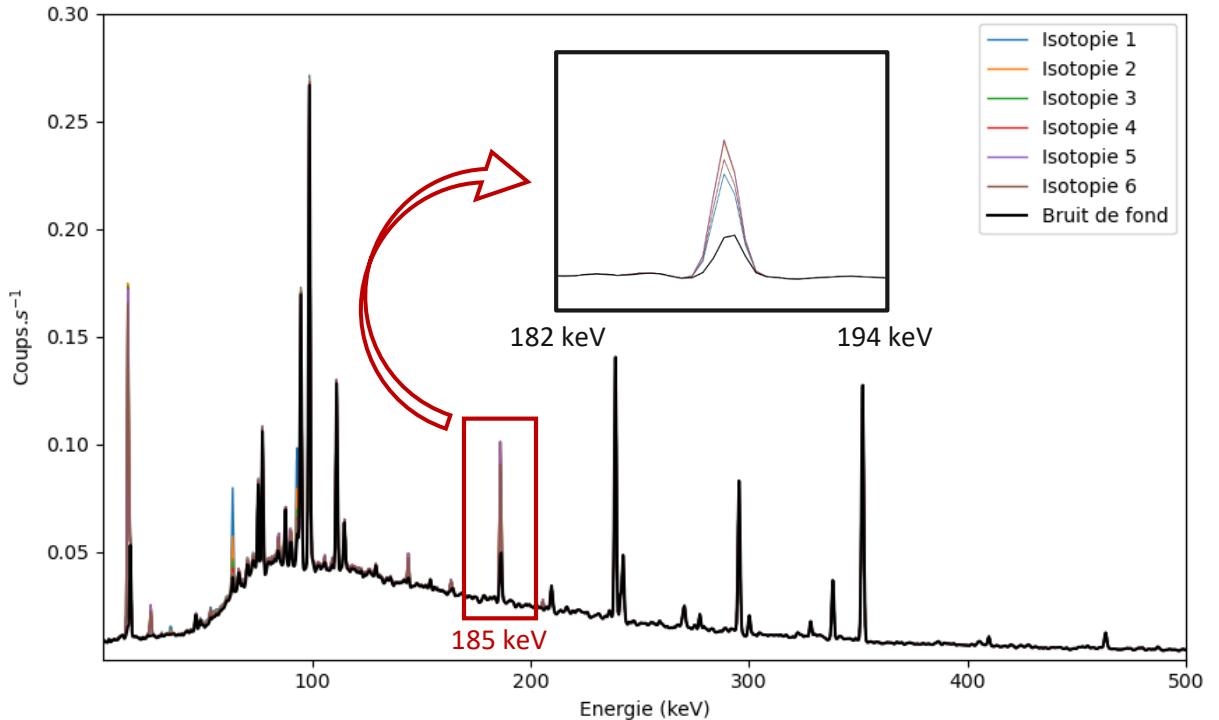
Potassium \rightarrow 93,26 % ^{39}K (stable) | 0,012 % ^{40}K | 6,75 % ^{41}K (stable)

At 80% enrichment :

- The ^{238}U contribution becomes negligible
- Slight increase in the ^{235}U and ^{234}U contributions



Gamma rays detection (HPGe)

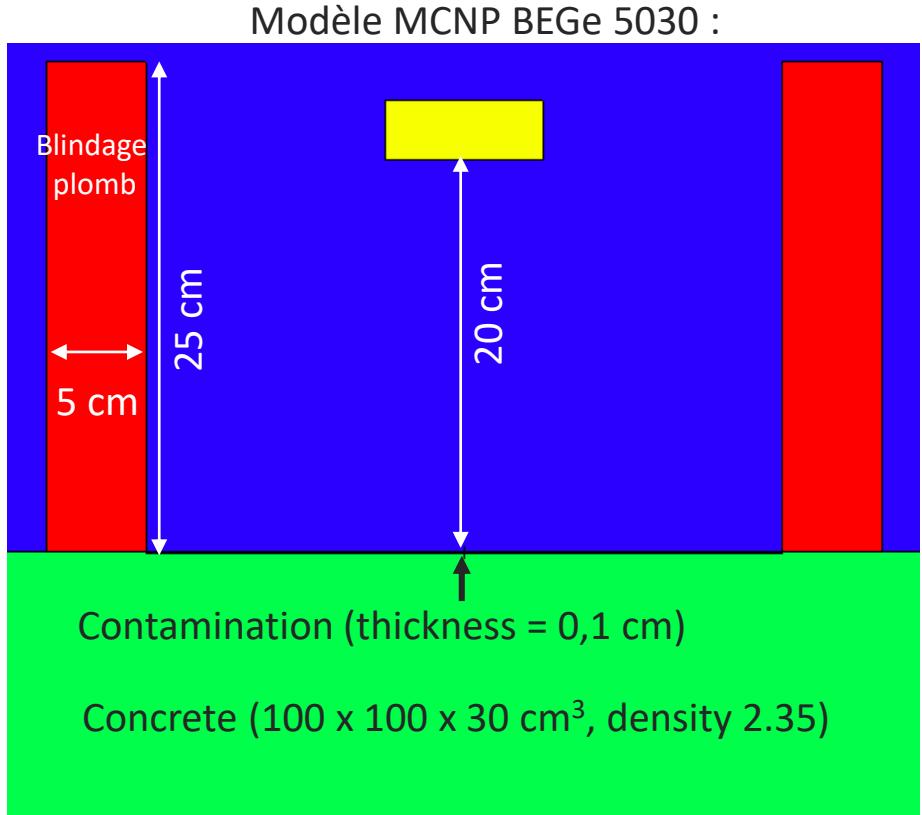


Background noise :

- 2.1 ppm_U, 4.2 ppm_{Th}, 0.78 %_K

Contamination :

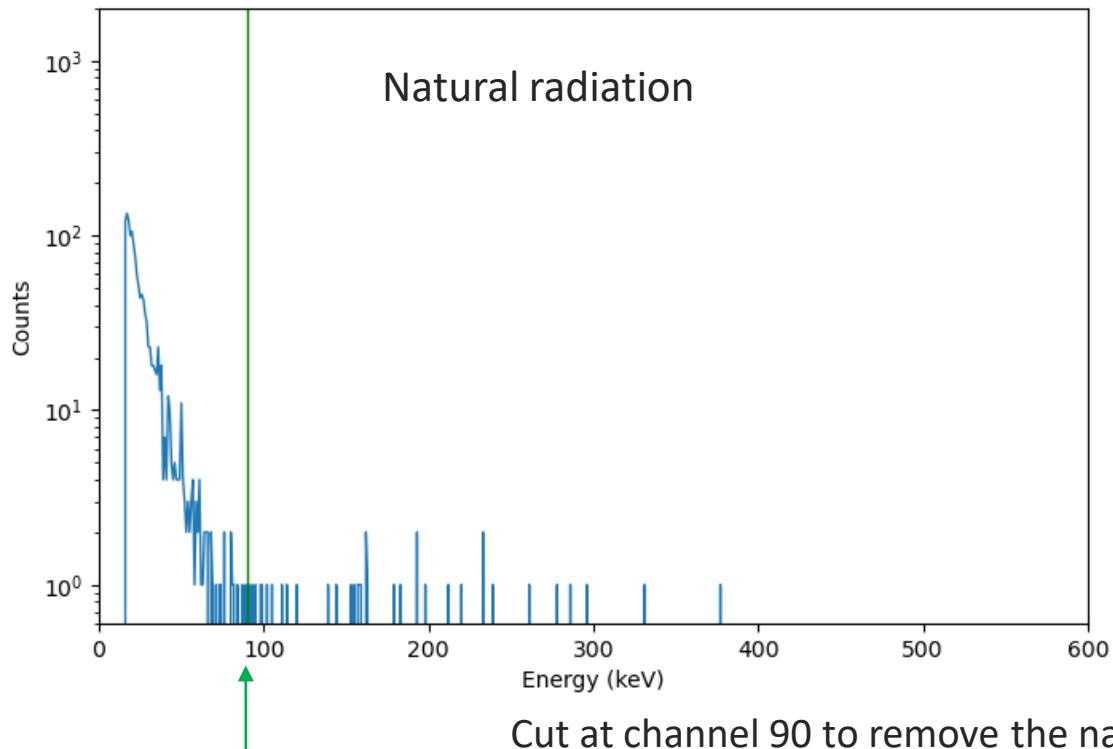
- 1000 Bq d'activité totale
- Distributed in a 32 x 32 x 0.1 cm³ slab
- Surface = 1024 cm² → 0.98 Bq/cm²



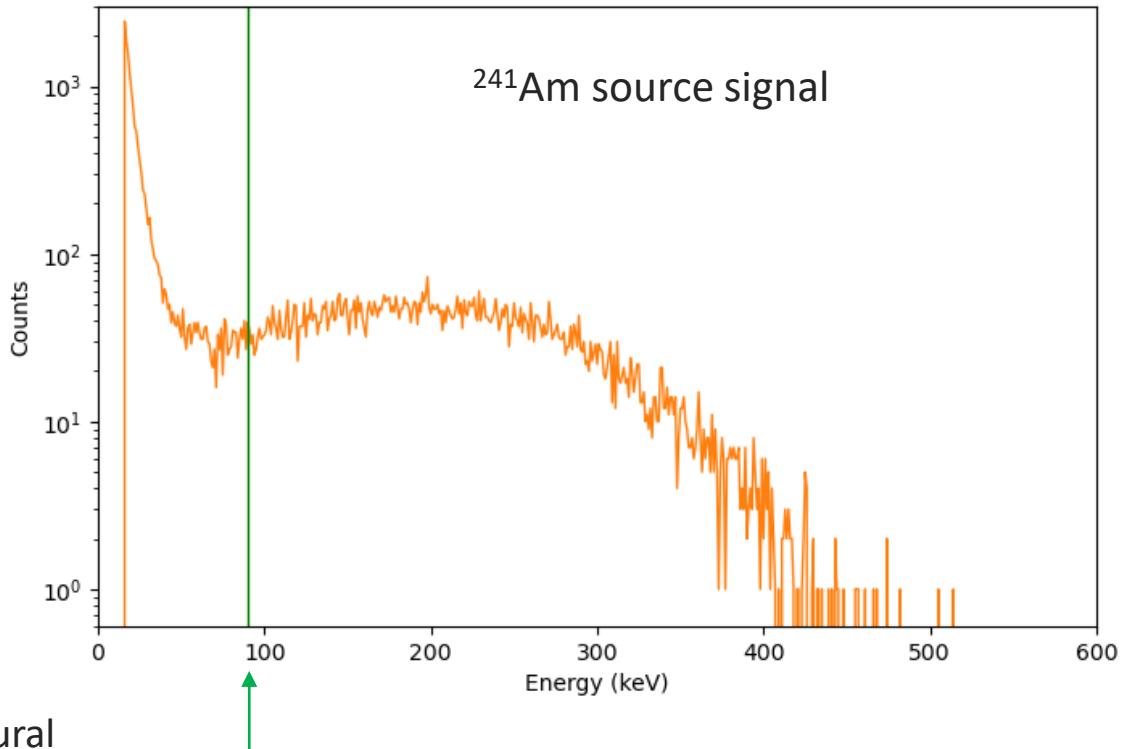
Alpha particles detection (experimental) LMN

Distance to source : 11 cm

Acquisition time : 3 minutes



Cut at channel 90 to remove the natural radiation signal + intrinsic noise



Note : alpha detection does not depend on enrichment
All 3 uranium isotopes decay via alpha emission with a 100 % probability

Alpha particles detection (experimental)

Alpha simulation → high uncertainty and resources consuming

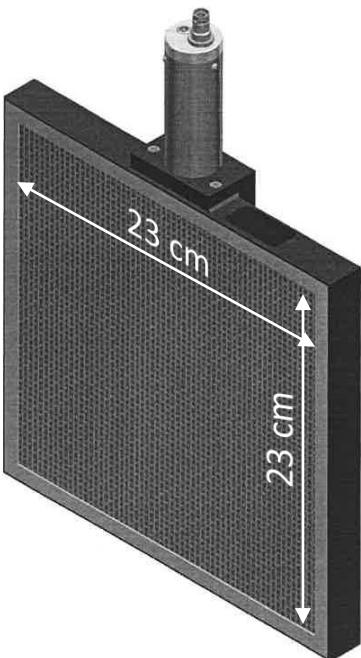
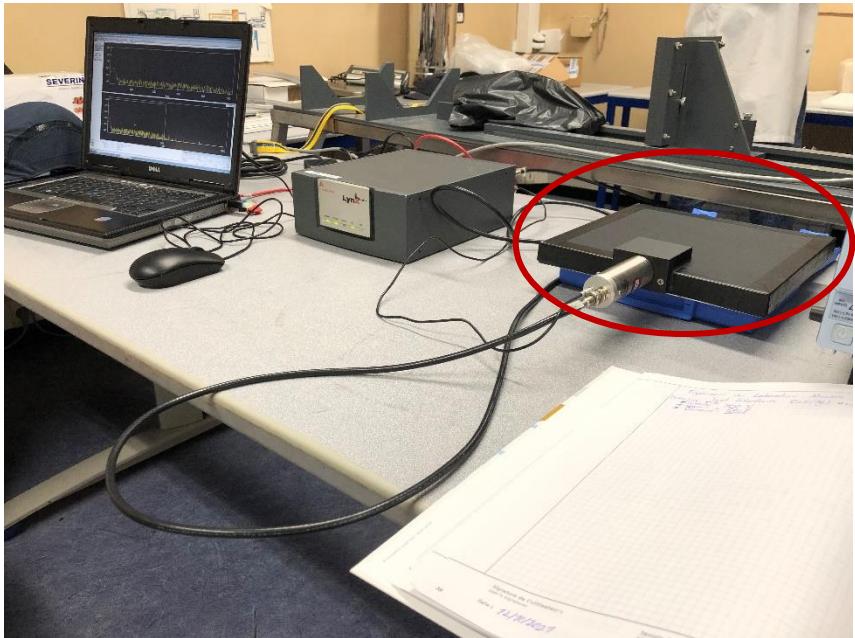
Experiments carried out using a ^{241}Am alpha source and a ZnS(Ag) scintillator detector

ZnS(Ag) characteristics :

Detection surface : $23 \times 23 \text{ cm}^2$

$6 \mu\text{m}$ aluminized PET entry window

Metal grid for screen protection

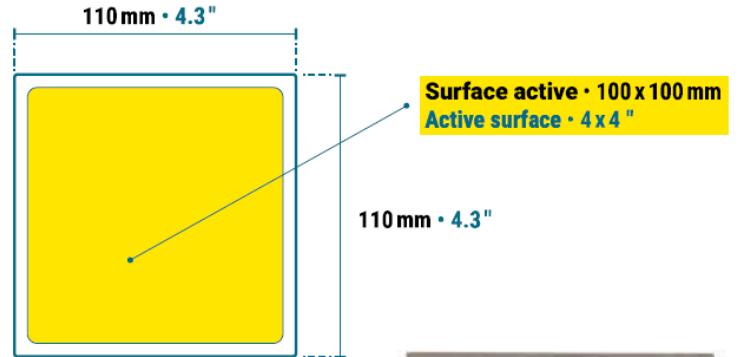


^{241}Am source characteristics :

Alpha emission rate $\approx 376 \text{ s}^{-1} (4\pi)$

Active surface = $10 \times 10 \text{ cm}^2$

Emission energy = 5,5 MeV



Source ^{241}Am

Estimation of acquisition times and conclusions

Estimation of required counting time to reach a certain detection limit :

$$DL_{c/s} = 4 \times \sqrt{BN} \times \frac{1}{\sqrt{T_{acq}}}$$

$$DL_{Bq/cm^2} = \frac{DL_{c/s}}{CF}$$

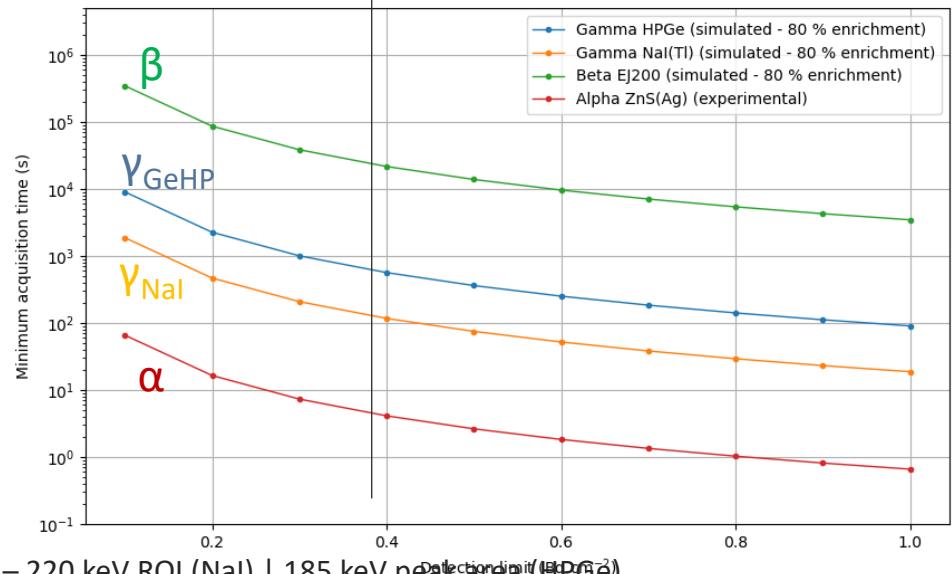
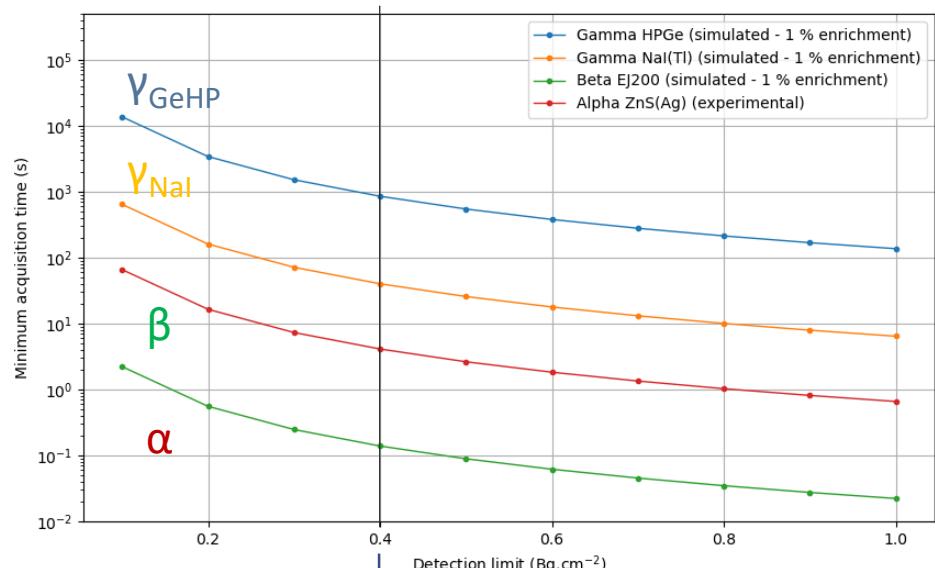
$$T_{acq \min} = \left(\frac{4 \times \sqrt{BN}}{DL_{Bq.cm^{-2}} \times CF} \right)^2$$

- BN : background noise produced by natural radiation in the ROI (in counts)
- CF : Calibration factor to convert $c.s^{-1}$ to $Bq.cm^{-2}$ (in $c.s^{-1}.Bq^{-1}.cm^2$)
- $DL_{Bq.cm^{-2}}$: detection limit to achieve (in $Bq.cm^{-2}$)
- $T_{acq \ min}$: minimum acquisition time required

Minimum acquisition time (s) to reach the 0,4 $Bq.cm^{-2}$ DL

^{235}U enrichment	Gamma HPGe	Gamma NaI(Tl)	Beta EJ200	Alpha ZnS(Ag)
1 %	854	40,2	$13,9 \times 10^{-2}$	$22,9 \times 10^{-3}$
80 %	565	117	$21,6 \times 10^3$	$22,9 \times 10^{-3}$

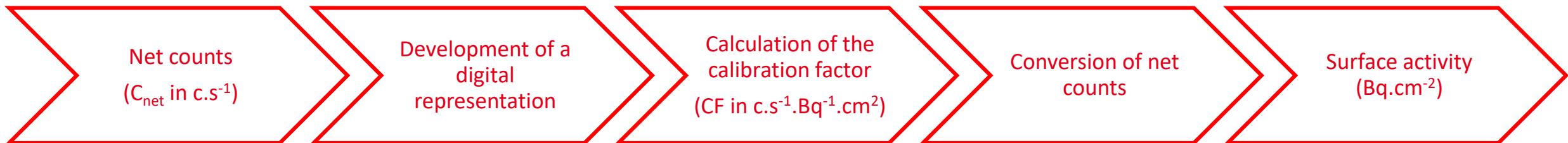
Detection time estimation allows us to determine the “detection levels” for the detection of uranium contamination (from fastest to slowest)
 → alpha > beta > gamma if the enrichment is low
 → alpha > gamma > beta if the enrichment is high



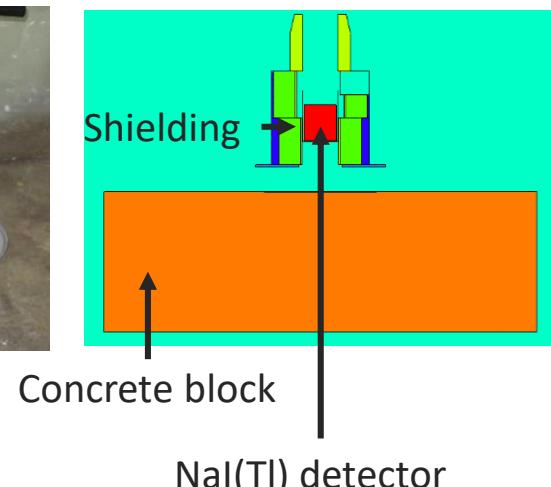


Method for surface activity estimation

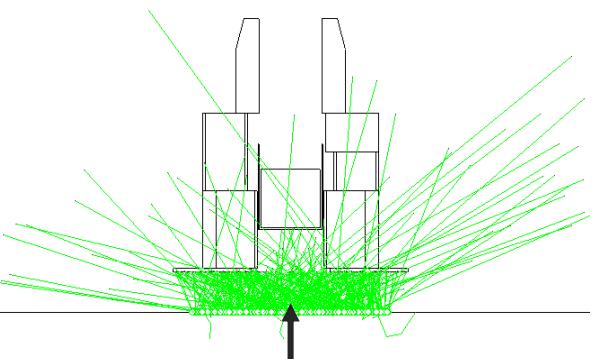
Count rate to surface activity estimation



Net counts =
gross counts - background
noise counts



Radiation detection simulation.
The source activity is normalized
to 1 Bq and its geometry is known



$$A_{surface} \text{ (Bq.cm}^{-2}\text{)} = \frac{C_{net} \text{ (c.s}^{-1}\text{)}}{CF \text{ (c.s}^{-1}\text{.Bq}^{-1}\text{.cm}^2\text{)}}$$