

MOLECULES AS PROBES OF THE STANDARD MODEL AND ITS EXTENSIONS (THEORETICAL PERSPECTIVE)

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Research:

Using state of the art methods from computational chemistry for high accuracy calculations of atomic and molecular properties in the context of fundamental problems in physics:

- Search for new physics with low-energy precision measurements
 - Violation of fundamental symmetries in atoms and in molecules
 - Search for variation of fundamental constants
- Highly accurate calculations of spectra and properties of heavy and superheavy atoms and highly charged ions

Most work done in collaboration with experimental groups

TALK OVERVIEW

- Motivation: why do we use molecules to test the Standard Model?
- Why do we need electronic structure theory?
- How do we perform accurate calculations and assign uncertainties?
- Applications

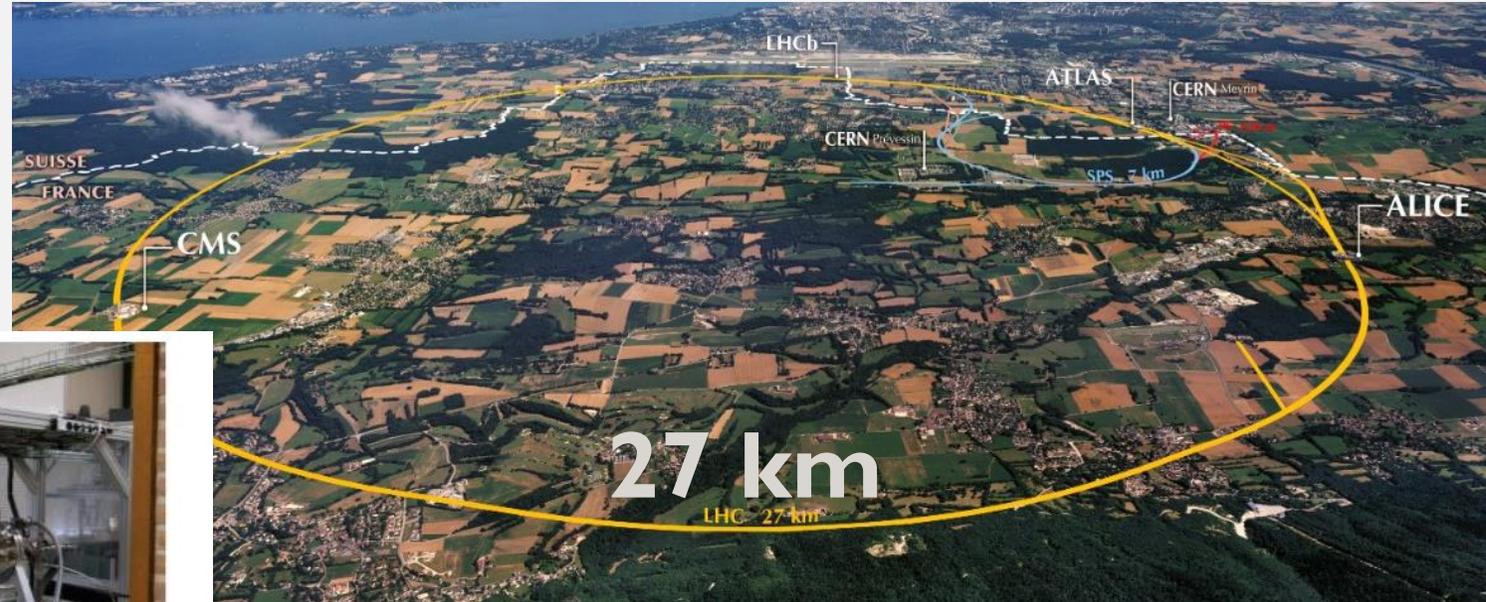
TALK OVERVIEW

- ~~Motivation: why do we use molecules to test the Standard Model?~~
- Why do we need electronic structure theory?
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MOTIVATION

Why look for physics beyond the SM with atoms and molecules?

- Accelerator research (LHCb, etc.)
- **Table-top experiments**



A panoramic picture of the four meter long traveling-wave decelerator that has been built in Groningen. It is in use for decelerating packets of the heavy diatomic molecule SrF, which is a prototypical system for the investigation of broken symmetries.

Why look for physics beyond the SM with molecules?

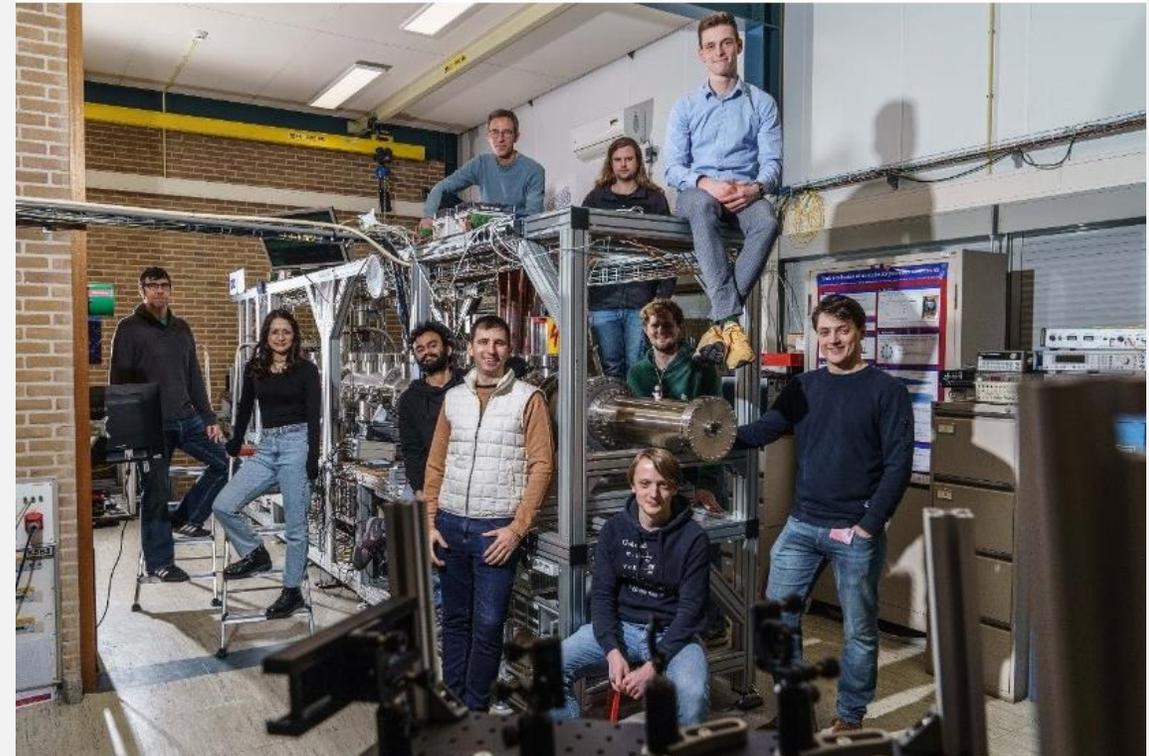
- We are looking for truly tiny effects (signs of asymmetry)
- Electronic structure acts as an amplifier, bringing these effects into measurable range
- Example: electric dipole moment of the electron (eEDM)
 - Violates P,T – symmetry
 - Standard Model prediction: $\sim 10^{-38}$ e·cm (too small for direct measurement)
 - Other theories predict larger values (still too small for direct measurement)
 - In a polar molecule, the internal electric field amplifies the electron EDM by 5 orders of magnitude → we can try to measure it

Why look for physics beyond the SM with molecules?

- We are looking for truly tiny effects (signs of asymmetry)
- Electronic structure acts as an amplifier, bringing these effects into measurable range
- For example: electric dipole moment of the electron (eEDM)
 - Similar for other P(T)-violating properties and BSM effects
 - Similar for parity violation

Why look for physics beyond the SM with molecules?

- **Table-top experiments with molecules: a promising alternative to high energy research**
 - Versatile, sensitive to different phenomena
 - Parity violation
 - EDMs (electron, hadronic)
 - Variation of fundamental constants
 - Dark matter
 - ..
 - Various enhancement effects → high sensitivity
 - Small scale
 - (Relatively) inexpensive



Why look for physics beyond the SM with **radioactive** molecules?

- Heavy systems, effects scale as Z^{2-5}
- Possible strong further enhancements due to the nuclear structure
- Sensitive to hadronic EDMs, nuclear magnetic quadrupole moments, etc.

Reports on Progress in **Physics**

REPORT ON PROGRESS

Opportunities for fundamental physics research with radioactive molecules

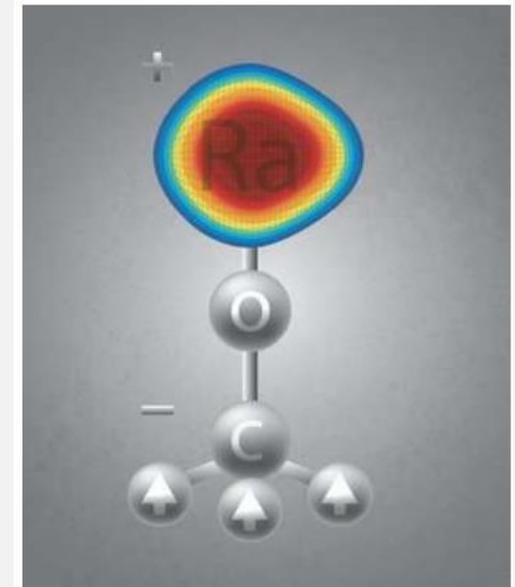
Gordon Arrowsmith-Kron, Michail Athanasakis-Kaklamanakis, Mia Au, Jochen Ballof, Robert Berger, Anastasia Borschevsky, Alexander A Breier, Fritz Buchinger, Dmitry Budker, Luke Caldwell, Christopher Charles, Nike Dattani, Ruben P de Groot, David DeMille, Timo Dickel, Jacek Dobaczewski, Christoph E Düllmann, Ephraim Eliav, Jonathan Engel, Mingyu Fan, Victor Flambaum, Kieran T Flanagan, Alyssa N Gaiser, Ronald F Garcia Ruiz, Konstantin Gaul, Thomas F Giesen, Jacinda S M Ginges, Alexander Gottberg, Gerald Gwinner, Reinhard Heinke, Steven Hoekstra, Jason D Holt, Nicholas R Hutzler, Andrew Jayich, Jonas Karthein, Kyle G Leach, Kirk W Madison, Stephan Malbrunot-Ettenauer, Takayuki Miyagi, Iain D Moore, Scott Moroch, Petr Navratil, Witold Nazarewicz, Gerda Neyens, Eric B Norrgard, Nicholas Nuscgart, Lukáš F Pašteka, Alexander N Petrov, Wolfgang R Plaß, Roy A Ready, Moritz Pascal Reiter, Mikael Reponen, Sebastian Rothe, Marianna S Safronova, Christoph Scheidenerger, Andrea Shindler, Jaideep T Singh, Leonid V Skripnikov, Anatoly V Titov, Silviu-Marian Udrescu, Shane G Wilkins and Xiaofei Yang

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[Reports on Progress in Physics, Volume 87, Number 8](#)

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<https://phys.org/news/2021-03-radioactive-molecules-mystery-antimatter.html>

Promising yet challenging...

- Challenge of achieving unprecedented sensitivity needed to detect the tiny effects of new physics
- For radioactive systems, extra challenge of dealing with small amounts of short-lived atoms
- **Alongside specially developed experimental techniques, theoretical support becomes crucial**

TALK OVERVIEW

- Motivation: why do we use molecules to test the Standard Model?
- **Why do we need electronic structure theory?**
- How do we perform accurate calculations and assign uncertainties?
- Applications

How can (molecular) theory be of use?

(Important to remember: for us the practical considerations do not play a role!)

- Insight into effect of new physics on molecular properties
- Identification of promising candidates for measurements
 - High sensitivity
 - Experimental considerations (stability, laser-coolability, etc.)
- Practical parameters for experiment (predictions of transition energies, laser-cooling schemes, etc.)
- Parameters for the interpretation of the results (coupling parameters for new physics phenomena, etc.)

How can (molecular) theory be of use?

(Important to remember: for us the practical considerations do not play a role!)

- Parameters for the interpretation of the results (coupling parameters for new physics phenomena, etc.)
- $P(T)$ -violating energy (frequency) shift is obtained in experiment by reversing parity (reversing the direction of external fields): ΔE
- This energy difference can be related to the underlying $P(T)$ -violating phenomena through electronic structure parameters (coupling constants)
- Example: P,T-violating effects in a paramagnetic molecule:

$\Delta E = (W_d d_e + W_s \kappa_s) \langle \hat{S} \cdot \hat{n} \rangle (E_{\text{ext}})$

What we measure

What we want to know

Comes from theory

Theoretical predictions

For use in experiments, theoretical predictions should be

- Reliable
- Based on high accuracy calculations
- Be accompanied by uncertainties:
 - Important for planning the measurements (e.g. what range should be scanned?)
 - Important for extraction of properties of interest (uncertainty on the extracted new physics parameters will include the theoretical uncertainty)

Thus, we need an accurate method that allows uncertainty estimation!

TALK OVERVIEW

- Motivation: why do we use molecules to test the Standard Model?
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- **How do we perform accurate calculations and assign uncertainties?**
- Applications

COMPUTATIONAL METHODS

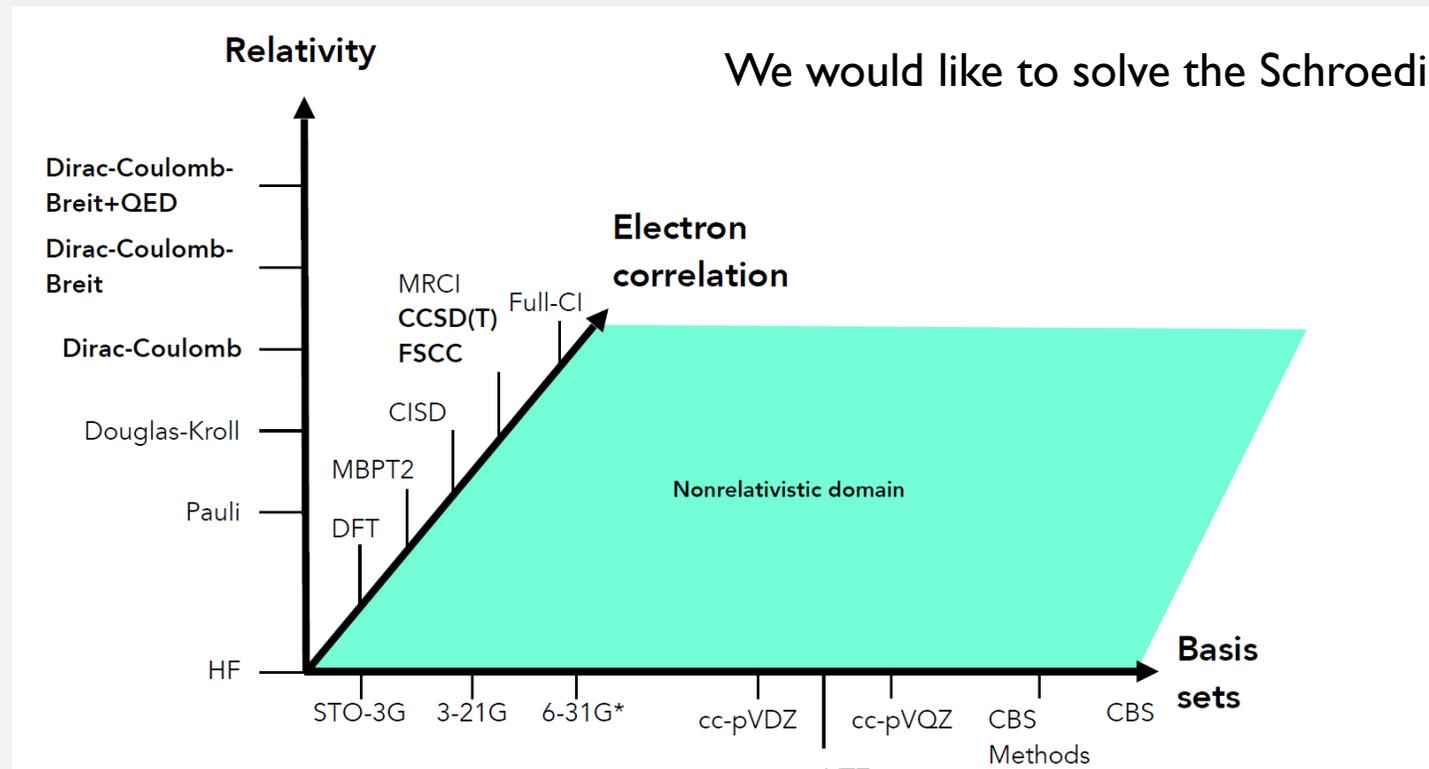
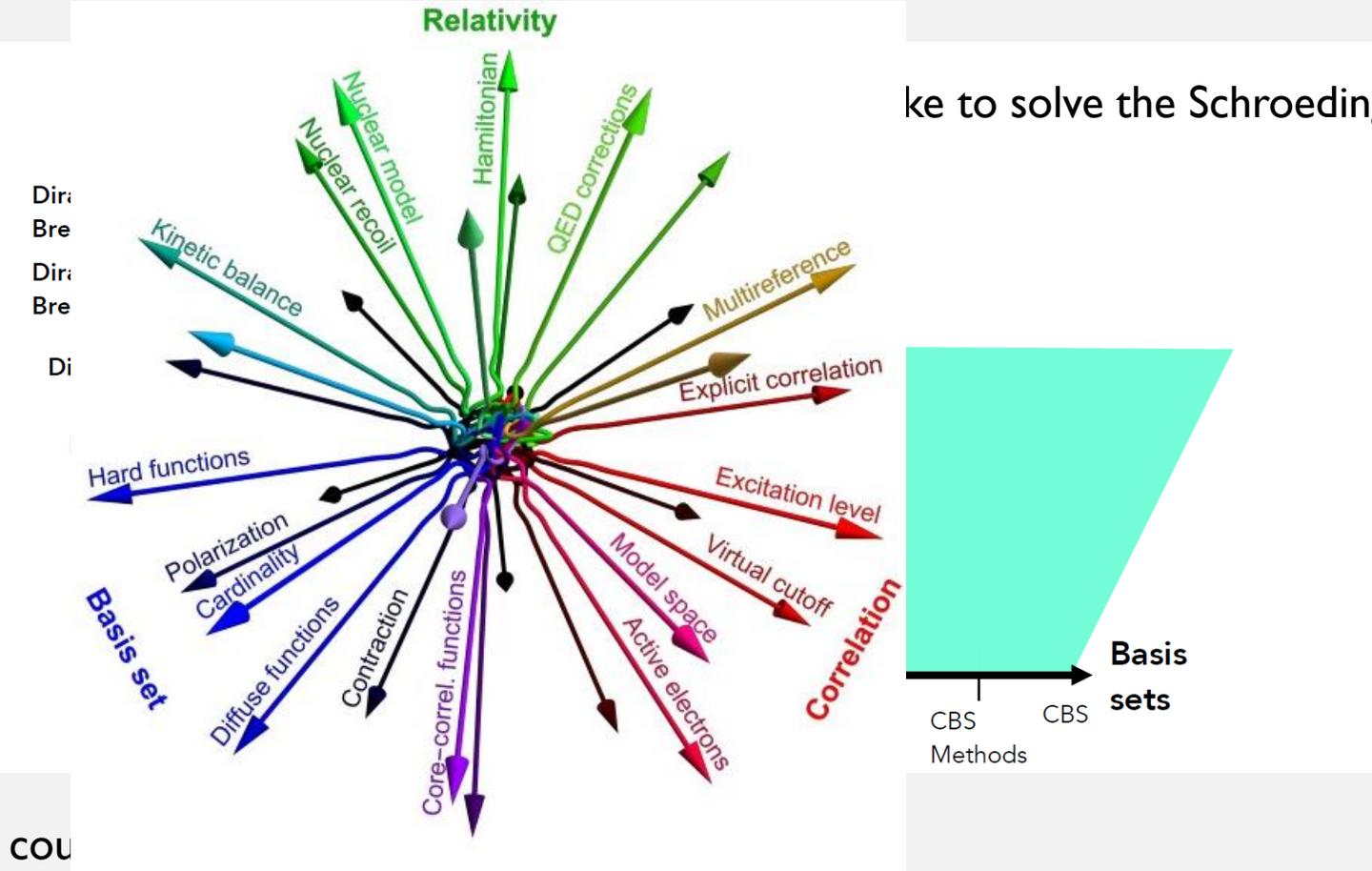


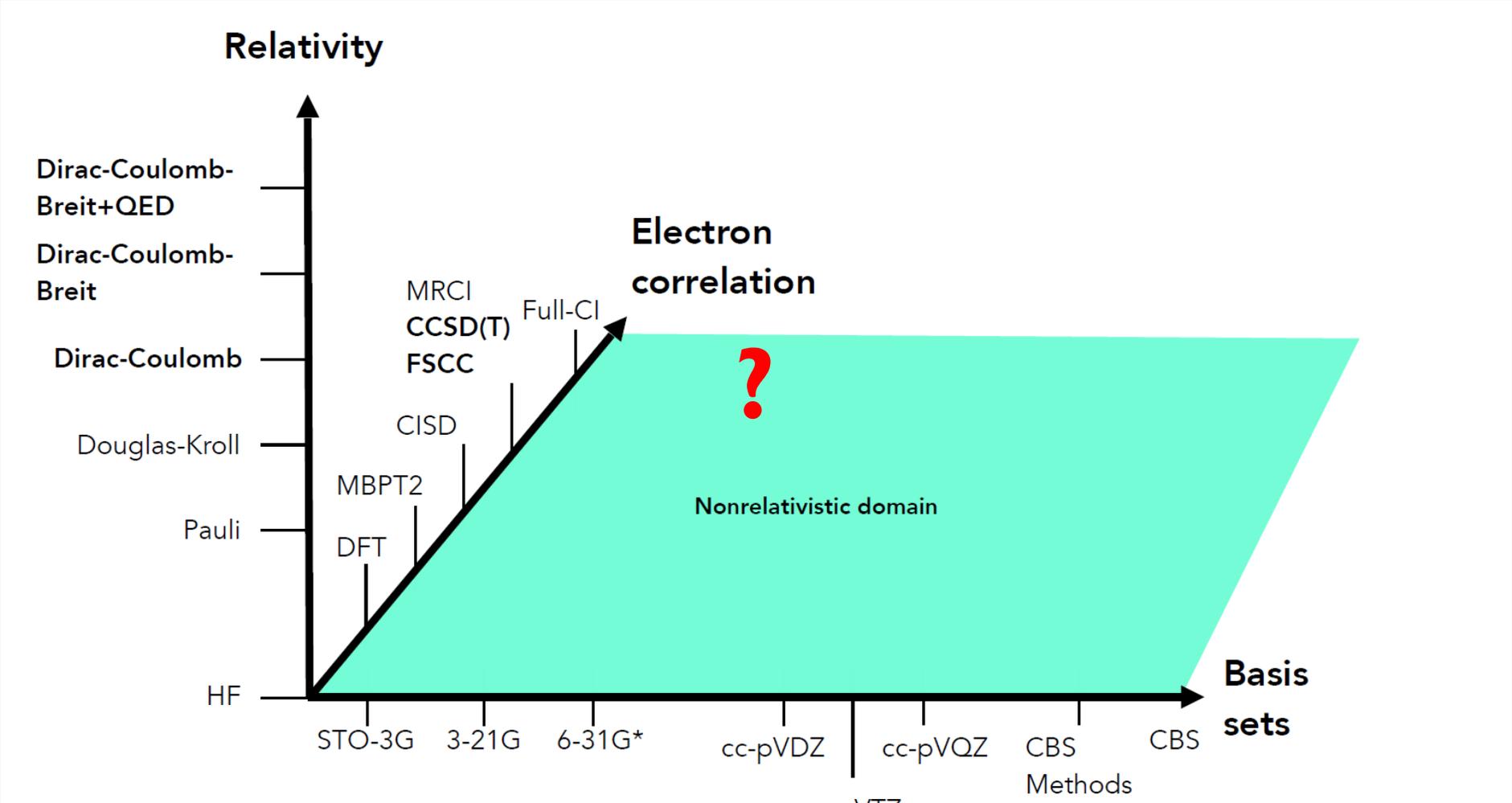
Figure courtesy of P. Schwerdtfeger

COMPUTATIONAL METHODS



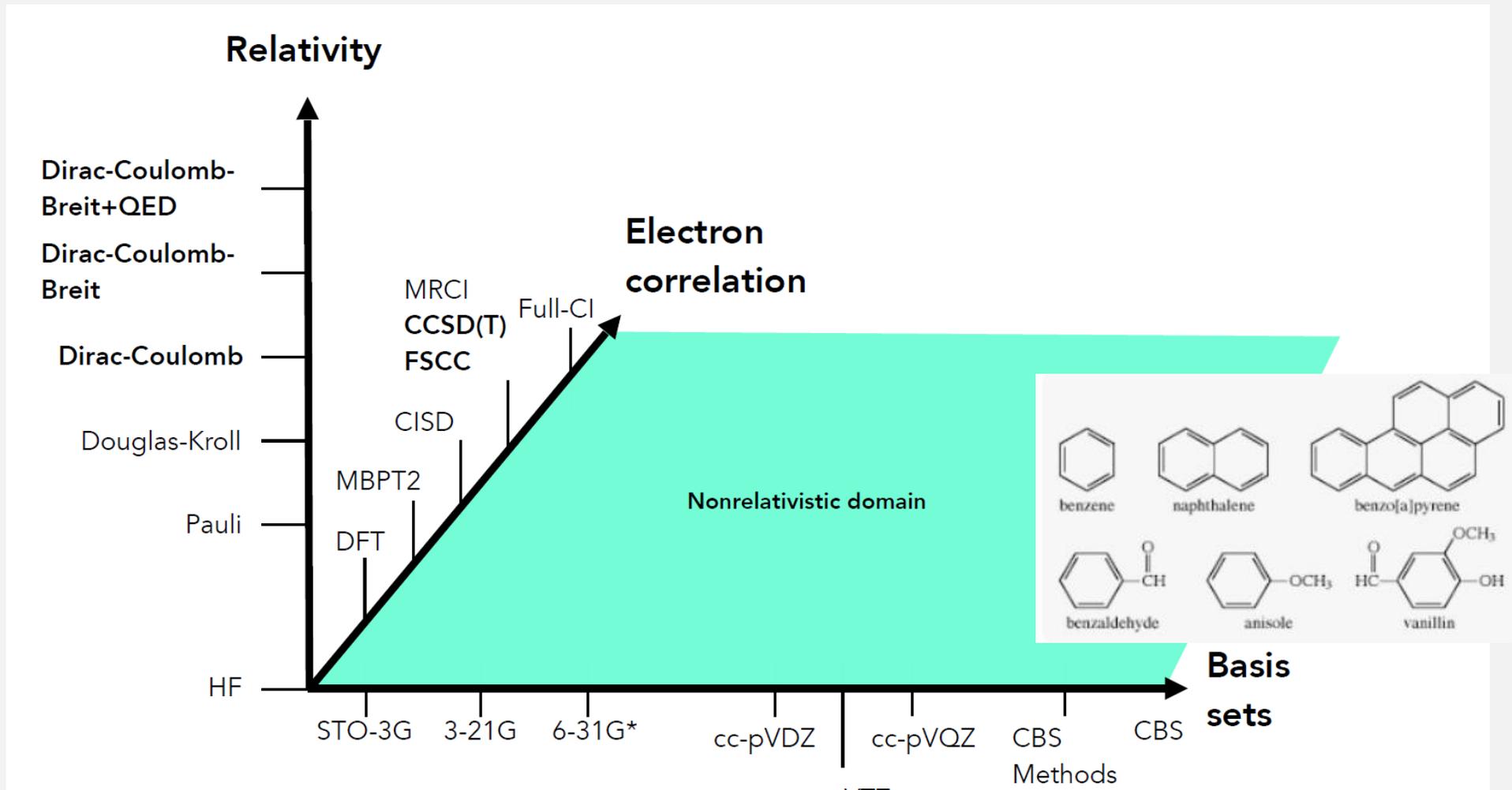
ke to solve the Schroedinger equation.

Figure cou

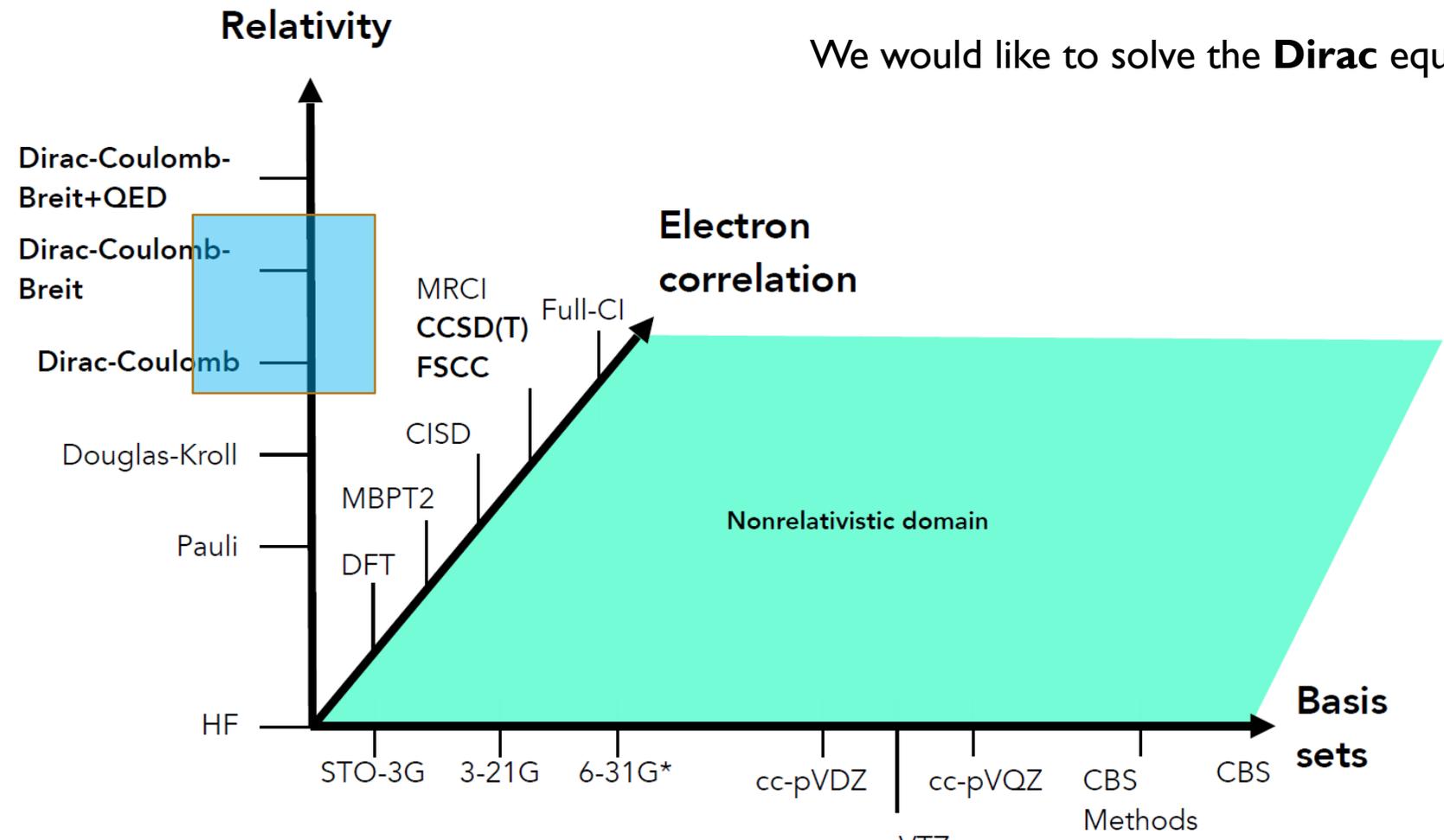


What do we want to calculate?

- Coupling parameters describing the effect of $P(T)$ -violating phenomena on electronic structure
 - Relativistic in nature, hence relativistic methods
- Molecular parameters needed in experiments
 - (usually) heavy systems, hence relativistic methods
-



We would like to solve the **Dirac** equation.



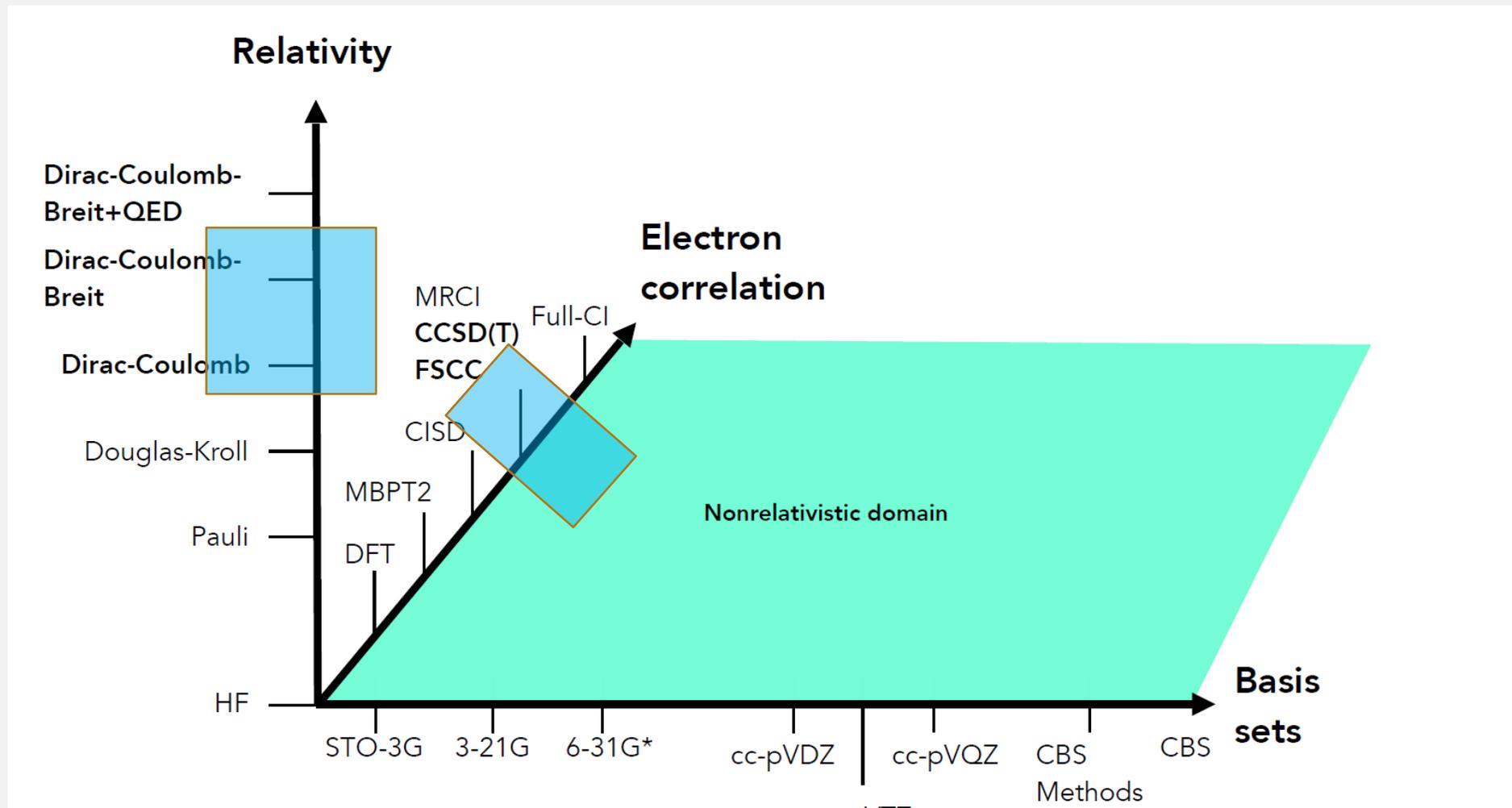
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- Coupling parameters describing the effect of $P(T)$ -violating phenomena on electronic structure
 - Relativistic in nature, hence relativistic methods
- Molecular parameters needed in experiments
 - (usually) heavy systems, hence relativistic methods
- High accuracy
 - State-of-the-art treatment of correlation
-

Electron correlation

- Explicit interaction between the electrons
- Important for light and heavy elements and for all properties
- Formidable problem, different approaches available
- Our method of choice: **Relativistic coupled cluster (CCSD(T), FSCC)**
- Transparent and robust approach, allows uncertainty estimation

Relativistic coupled cluster

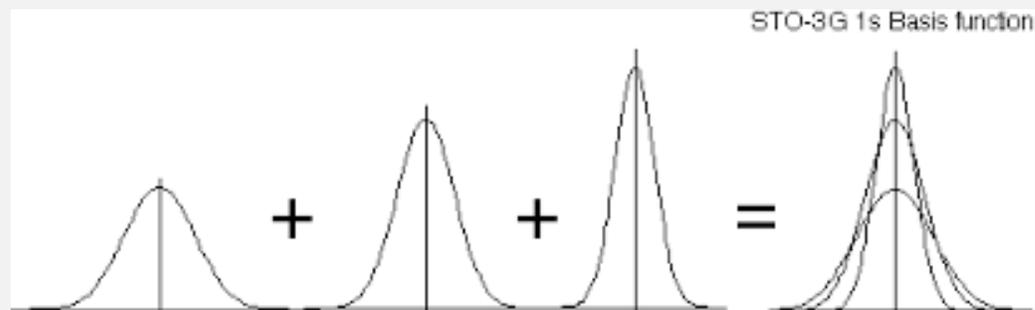


What do we want to calculate?

- Coupling parameters describing the effect of $P(T)$ -violating phenomena (or variation of constants) on electronic structure
 - Relativistic in nature, hence relativistic methods
- Atomic and molecular parameters needed in experiments
 - (usually) heavy systems, hence relativistic methods
- High accuracy
 - State-of-the-art treatment of correlation, large basis sets
-

Basis sets (a bit of alchemy)

- Sets of (Gaussian) functions that are used to represent the electronic WF.
- Atom specific, different basis sets for different properties



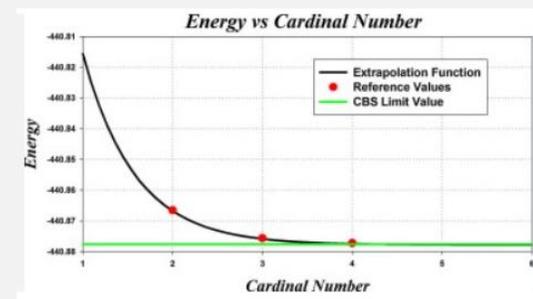
$$g = x^{\ell} y^m z^n e^{-\alpha r^2}$$

Gaussian type orbital

exponent' α

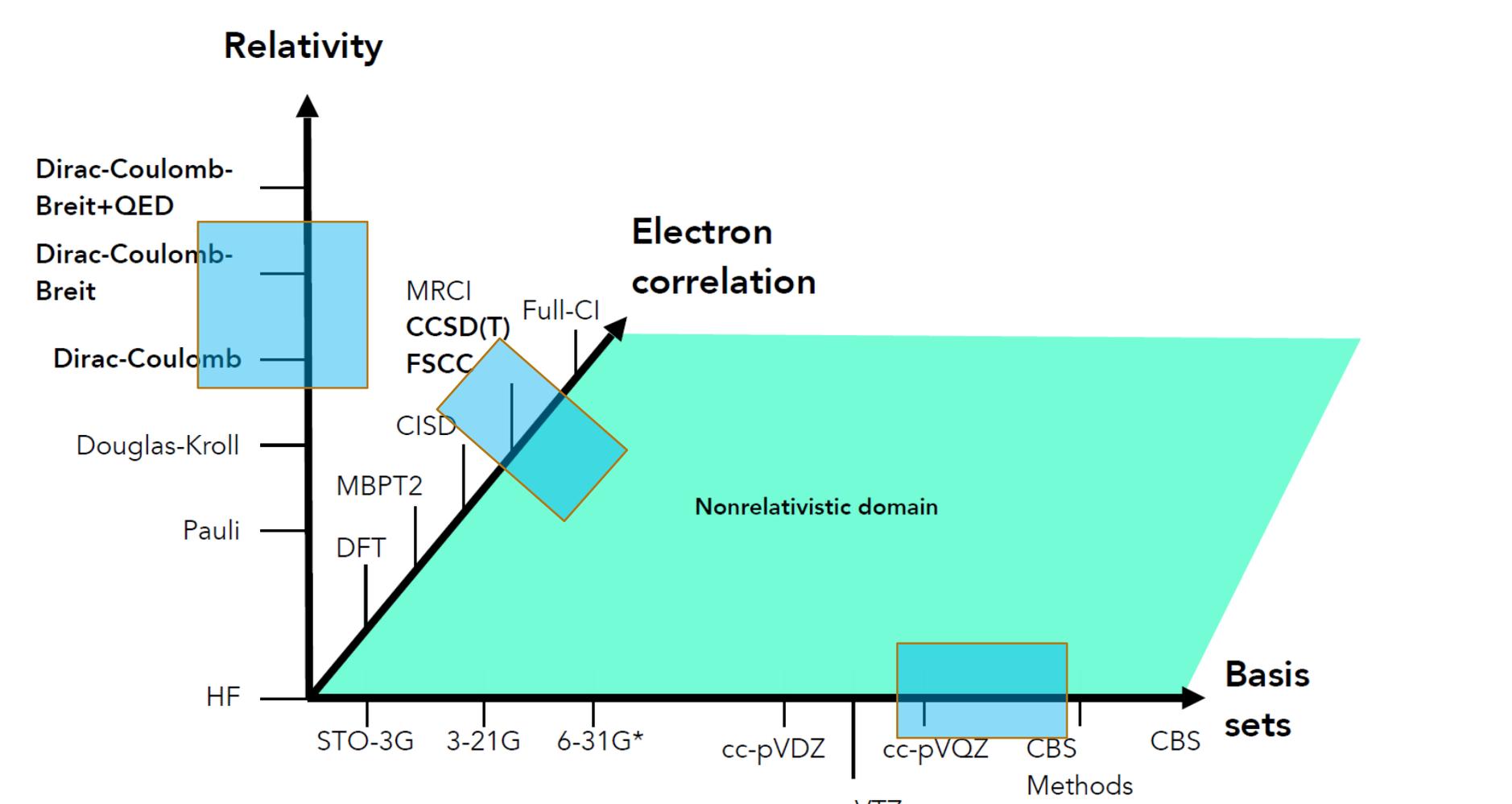
type of function:
 $\ell = m = n = 0$ for s type
 ℓ or m or $n = 1$ for p type
etc.

- Dyall's relativistic basis sets
- The larger the basis the better
- Complete Basis Set Limit (CBSL) extrapolation

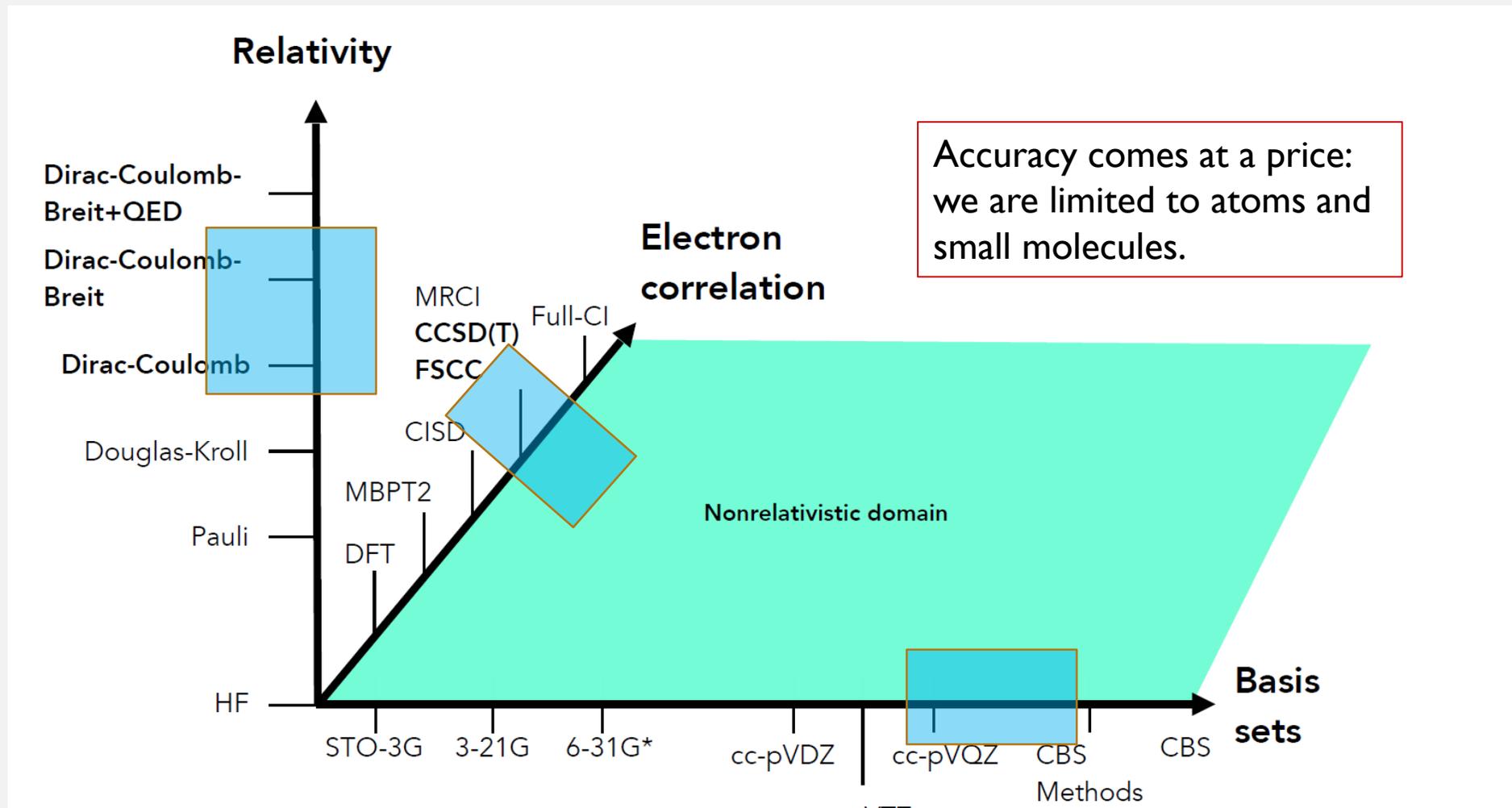


(K.G. Dyall, Theor. Chem Acc. 2002, 2004, 2006, 2007, 2009, 2011, 2012, etc.)

Relativistic coupled cluster



Relativistic coupled cluster

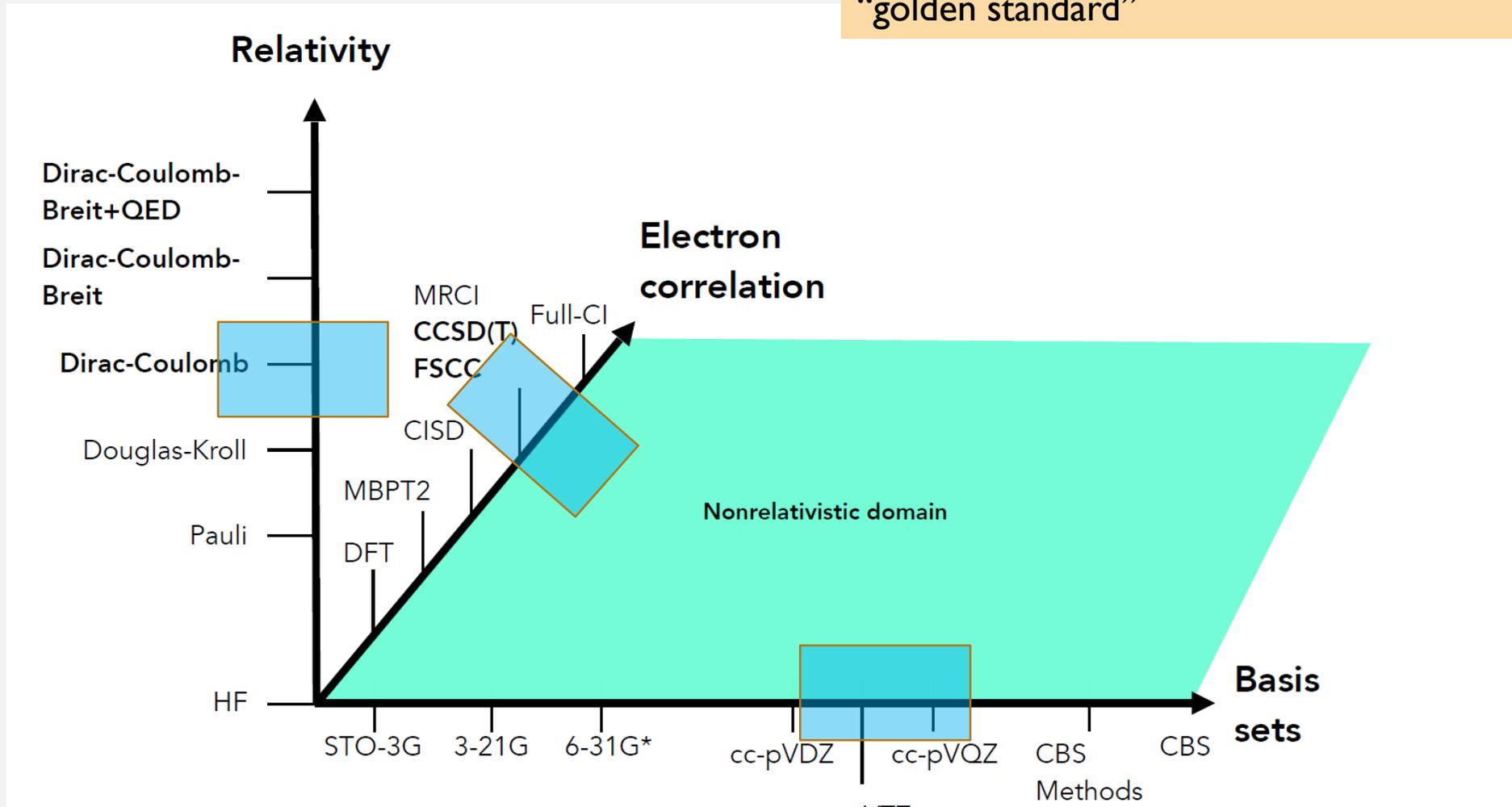


What can we calculate?

- Atomic properties: energies, IPs, EAs, spectra, hyperfine structure parameters, polarisabilities
- Molecular properties: geometries, molecular parameters, electronic structure, Franck-Condon Factors (FCFs), transition strengths
- Properties needed for SM tests:
 - W_d, W_s (eEDM experiments)
 - W_A (NSD-PV, nuclear anapole moments)
 - E_{PV} (parity-violating energy shifts)
 - Sensitivity to variation of α
 - ...
- Ground and excited states; closed and open shells
- Accuracy: ~ 10 s of meV for energies, single % for properties (can be pushed further to meV accuracy)
- Systematic investigation of effect of computational parameters and uncertainty evaluation

How do we reach meV accuracy?

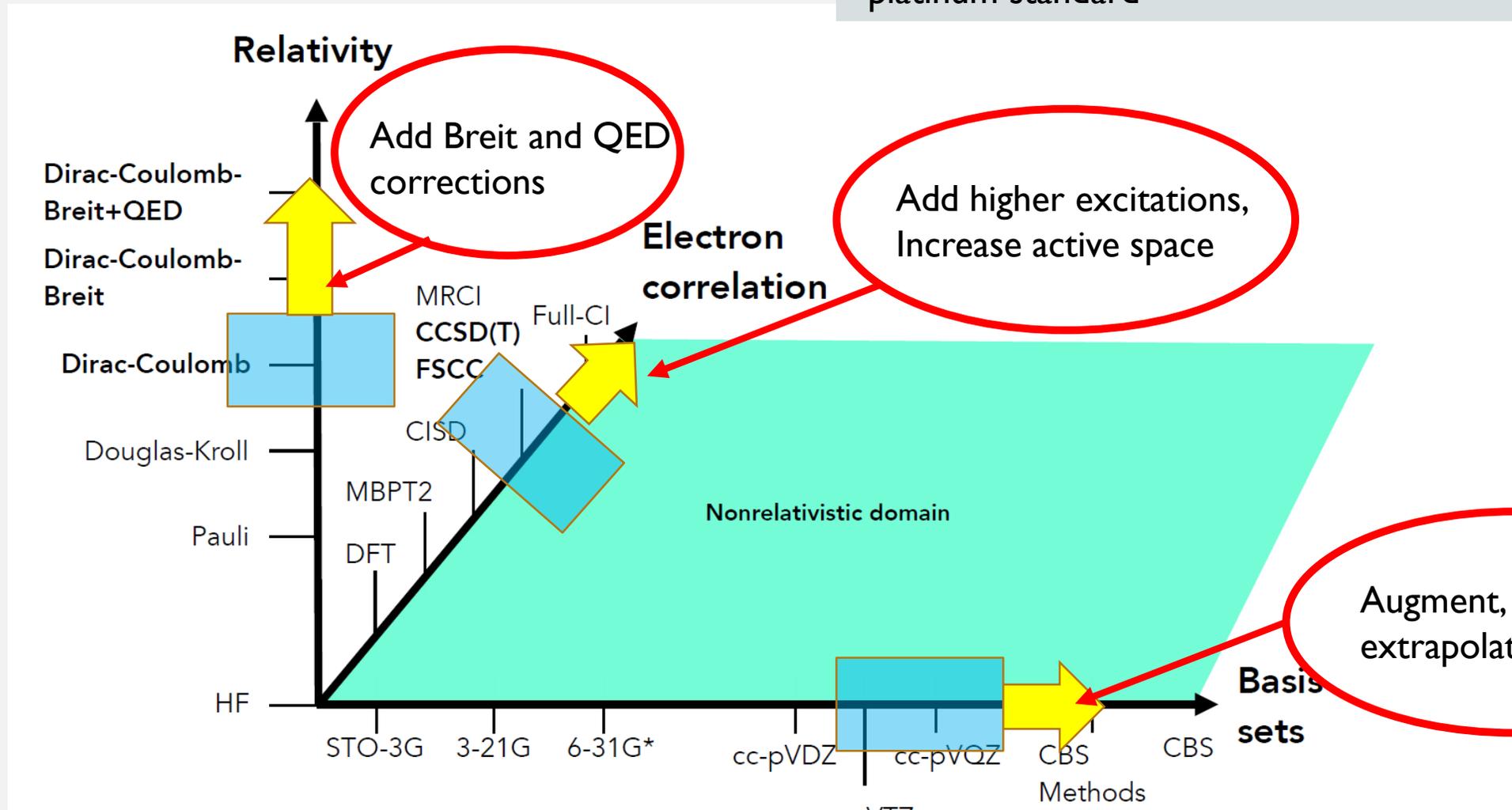
4c-CCSD(T), s-aug-cv4z, active space -20 to 50 a.u.
“golden standard”



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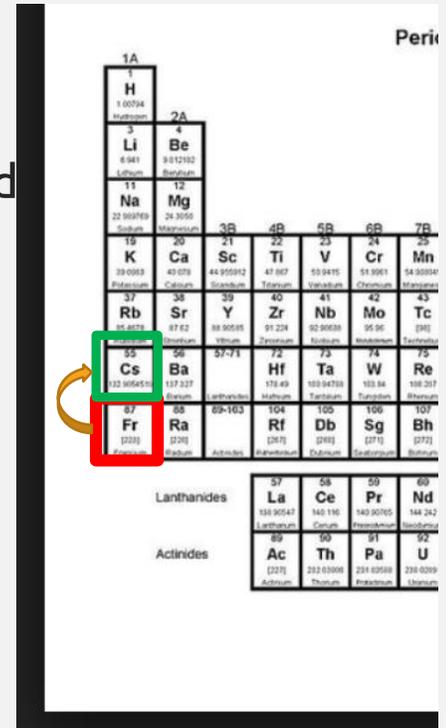
How do we reach meV accuracy?

4c+Breit+QED-CCSDT(Q), CBS limit,
all electrons correlated
“platinum standard”



How do we assign uncertainties (no clear recipe available)?

- Calculate the same property in a lighter homologue (and compare to experiment)
 - E.g., accuracy of the calculated electron affinity in a light element, compared to available experiment, can be used to set error bars on the same property in the heavier homologue
- Calculate a different property in the same atom/molecule (and compare to experiment)
- Perform a computational investigation to assign uncertainties based on method incompleteness



The image shows a portion of the periodic table. The elements Cesium (Cs) and Francium (Fr) are highlighted with red boxes. A yellow arrow points from Cs to Fr, indicating a relationship between these two elements. The table includes element symbols, atomic numbers, and names. The lanthanides and actinides series are also shown at the bottom.

1A	2A	3B	4B	5B	6B	7B
1 H 1.00794 Hydrogen						
3 Li 6.941 Lithium	4 Be 9.012182 Beryllium					
11 Na 22.989769 Sodium	12 Mg 24.3050 Magnesium					
19 K 39.0983 Potassium	20 Ca 40.078 Calcium	21 Sc 44.955912 Scandium	22 Ti 47.887 Titanium	23 V 50.9415 Vanadium	24 Cr 51.9961 Chromium	25 Mn 54.938044 Manganese
37 Rb 85.4678 Rubidium	38 Sr 87.62 Strontium	39 Y 88.90585 Yttrium	40 Zr 91.224 Zirconium	41 Nb 92.90638 Niobium	42 Mo 95.96 Molybdenum	43 Tc [98] Technetium
55 Cs 132.90545196 Cesium	56 Ba 137.327 Barium	57-71 Lanthanides	72 Hf 178.49 Hafnium	73 Ta 180.94788 Tantalum	74 W 183.84 Tungsten	75 Re 186.207 Rhenium
87 Fr [223] Francium	88 Ra [226] Radium	89-103 Actinides	104 Rf [261] Rutherfordium	105 Db [262] Dubnium	106 Sg [266] Seaborgium	107 Bh [264] Bohrium
Lanthanides		57 La 138.90547 Lanthanum	58 Ce 140.12 Cerium	59 Pr 140.90766 Praseodymium	60 Nd 144.242 Neodymium	
Actinides		89 Ac [227] Actinium	90 Th 232.0375 Thorium	91 Pa 231.03688 Protactinium	92 U 238.02891 Uranium	

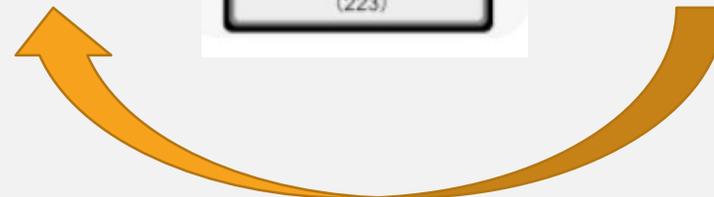
How do we assign uncertainties?

- Calculate the same property in a lighter homologue (and compare to experiment)
- Calculate a different property in the same atom/molecule (and compare to experiment)
 - E.g., we can use the accuracy of calculated ionization potential in a given element (compared to experiment) to set error bars on the calculated electron affinity in the same atom. Use with caution!
- Perform a computational investigation to assign uncertainties based on method incompleteness

Ionization potential,
Compare to exp.



Electron affinity?

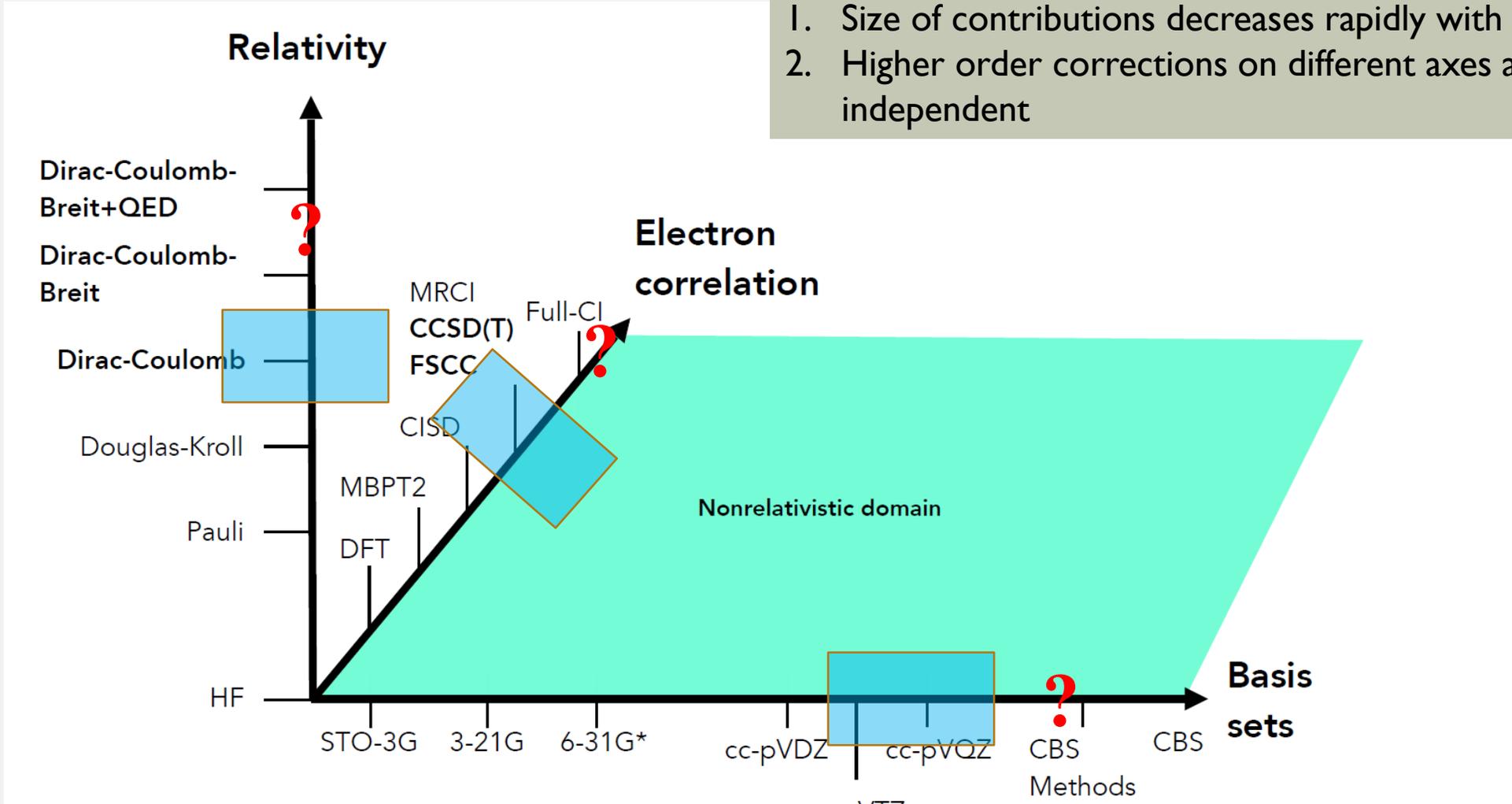


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How do we assign uncertainties?

Two assumptions:
 1. Size of contributions decreases rapidly with order
 2. Higher order corrections on different axes are independent

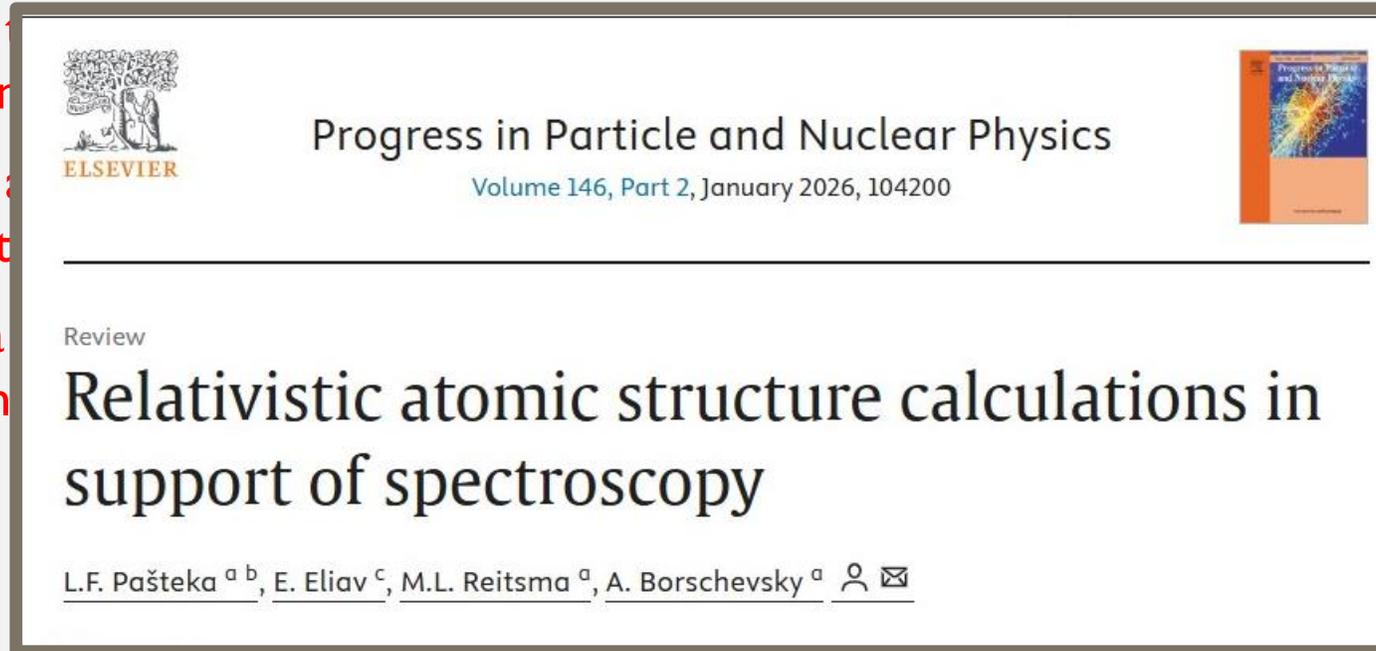


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- Perform a computational investigation to assign uncertainties based on method incompleteness

How do we assign uncertainties?

- Calculate theoretical values from experiments
- Calculate theoretical values and compare to experiments
- Perform a systematic uncertainty analysis in the method in question



TALK OVERVIEW

- Motivation: why do we use atoms and molecules to test the Standard Model?
- Why do we need electronic structure theory?
- How do we perform accurate calculations and assign uncertainties?
- **Applications**

APPLICATIONS

- Ionization potentials of CaF, SrF, BaF, and RaF: meV accuracy, uncertainty estimation, and predictive power
- P,T-violating coupling constants and molecular parameters
 - Search for the electric dipole moment on the electron: the NL-eEDM collaboration
 - Triatomic molecules: BaOH and YbOH

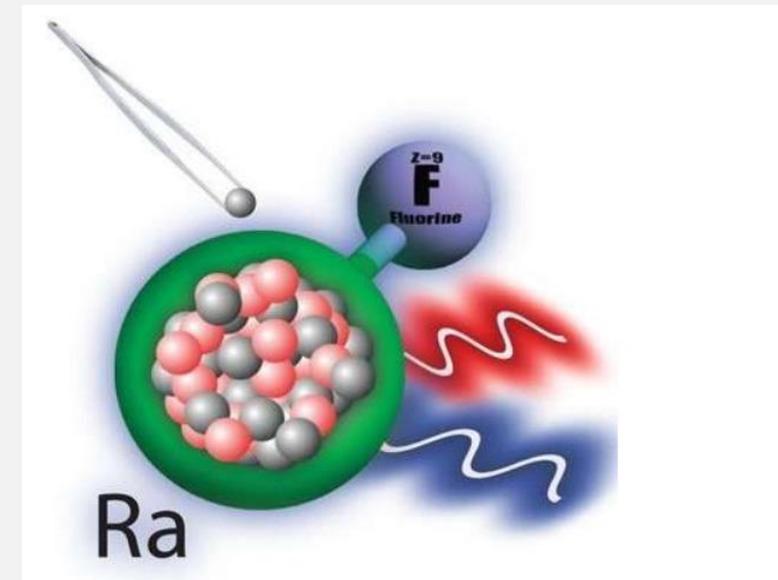
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MOLECULAR IONIZATION POTENTIALS

Ionization potential of RaF

- RaF is a promising candidate for new physics searches
- Many (high-impact) experimental studies
- Spectrum, isotope shifts
- New measurements: IP
- Theory before experiment



87. "Precision spectroscopy and laser cooling scheme of a radium-containing molecule".

S.M. Udrescu, S. Wilkins, A. Breier, M. Athanasakis-Kaklamanakis, R.F. Garcia Ruiz et al.

[Nature Physics \(2023\)](#).

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Spectroscopy of short-lived radioactive molecules

[R. F. Garcia Ruiz](#) , [R. Berger](#) , [J. Billowes](#), [C. L. Binnersley](#), [M. L. Bissell](#), [A. A. Breier](#), [A. J. Brinson](#), [K. Chrysalidis](#), [T. E. Cocolios](#), [B. S. Cooper](#), [K. T. Flanagan](#), [T. F. Giesen](#), [R. P. de Groote](#), [S. Franchoo](#), [F. P. Gustafsson](#), [T. A. Isaev](#), [Á. Koszorús](#), [G. Neyens](#), [H. A. Perrett](#), [C. M. Ricketts](#), [S. Rothe](#), [L. Schweikhard](#), [A. R. Vernon](#), [K. D. A. Wendt](#), ... [X. F. Yang](#)  [Show authors](#)

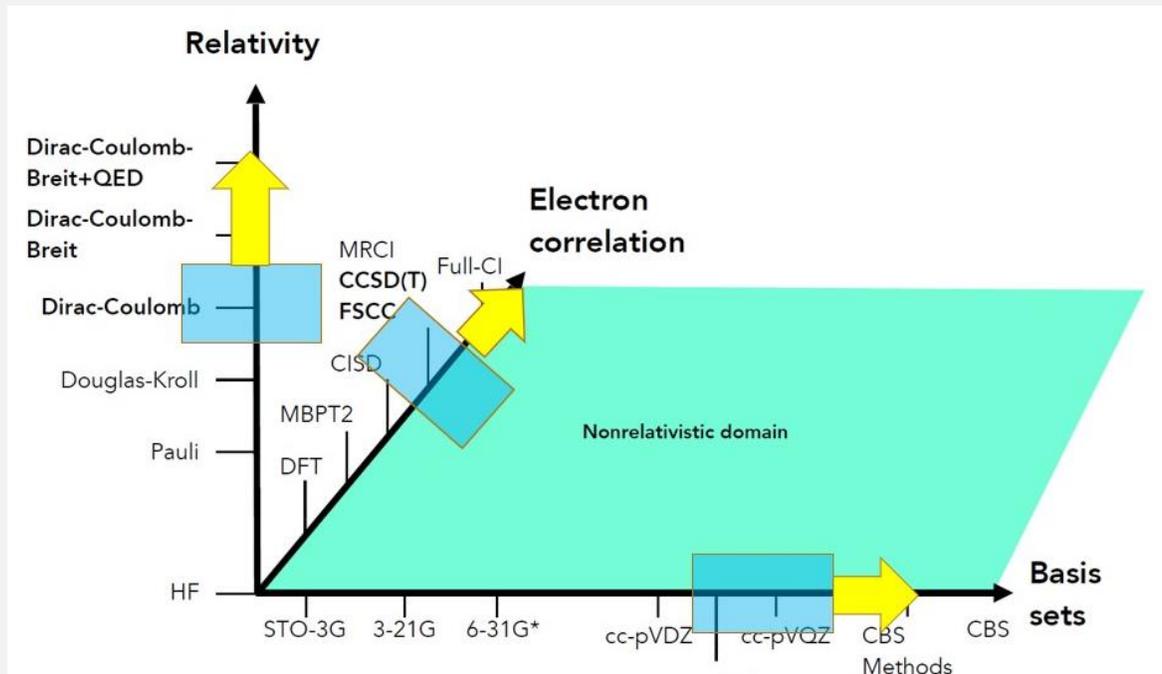
Precision spectroscopy and laser-cooling scheme of a radium-containing molecule

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Ionization potential of RaF

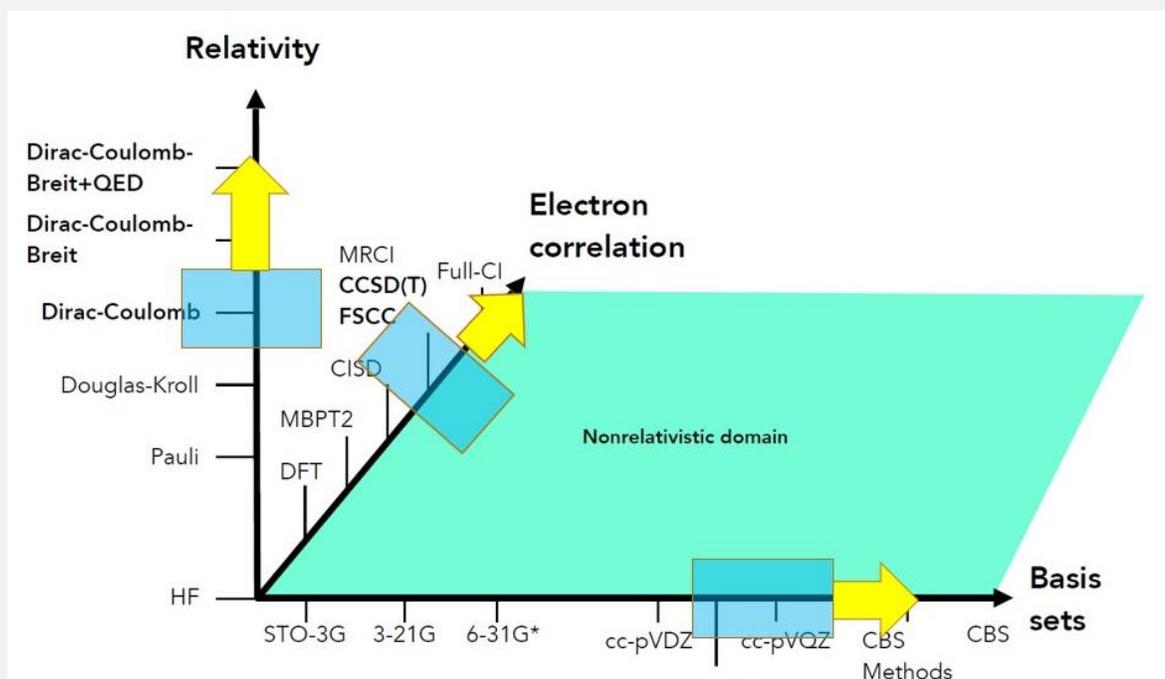
- Theory: relativistic CC with higher order corrections



Method	IP (eV)
CBS-DC-CCSD	4.932 ^a
CBS-DC-CCSD(T)	4.983 ^a
+augmentation	4.983
+core corr.+active space	4.989
+ ΔT	4.990
+Breit	4.988
+QED	4.981
Theoretical	4.981(7)

Ionization potential of RaF

- Theory: relativistic CC with higher order corrections



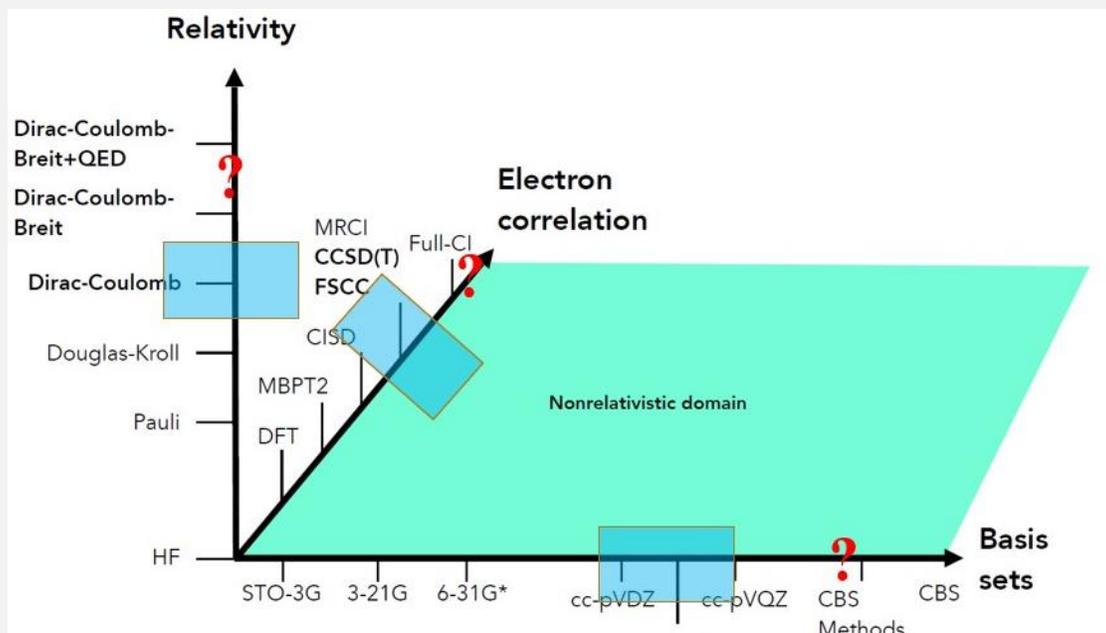
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“golden standard”

“platinum standard”

Ionization potential of RaF

- Theory: relativistic CC with higher order corrections
- Uncertainty evaluation: similar contributions from all sources



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Theoretical	4.981(7)

TABLE II. Summary of the main sources of uncertainty in the calculated IP and D_e of RaF.

Category	Error source	IP (meV)
Basis set	Cardinality	2.6
	Augmentation	0.3
Correlation	Core electrons	4.4
	Higher-order excitations	0.9
Relativity	QED	4.2
Uncertainty:		7

How do we assign uncertainties?

- Calculate the same property in a lighter homologue (and compare to experiment)
- Calculate a different property in the same atom/molecule (and compare to experiment)
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Ionization potentials of CaF, SrF, and BaF

- Test the methods on lighter homologues

-

TABLE IV: Calculated IPs of MF (M = Ca,Sr,Ba) [eV] including higher-order contributions.

Method	CaF	SrF	BaF
DC-CCSD	5.8180	5.3940	4.7700
DC-CCSD(T)	5.8236	5.4174	4.8036
+augmentation	-0.0015	0.0003	0.0004
+core corr.+active space	0.0007	-0.0007	-0.0012
+ ΔT	0.0005	0.0008	0.0010
+Breit	-0.0010	-0.0011	-0.0009
+QED	-0.0008	-0.0020	-0.0031
Final result	5.8216	5.4148	4.7998

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+core corr.+active space	0.0007	-0.0007	-0.0012
+ ΔT	0.0005	0.0008	0.0010
+Breit	-0.0010	-0.0011	-0.0009
+QED	-0.0008	-0.0020	-0.0031
Final result	5.8216	5.4148	4.7998

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Ionization potentials of CaF, SrF, and BaF

- Evaluate the weight of different sources of uncertainty
- Here, basis set dominates

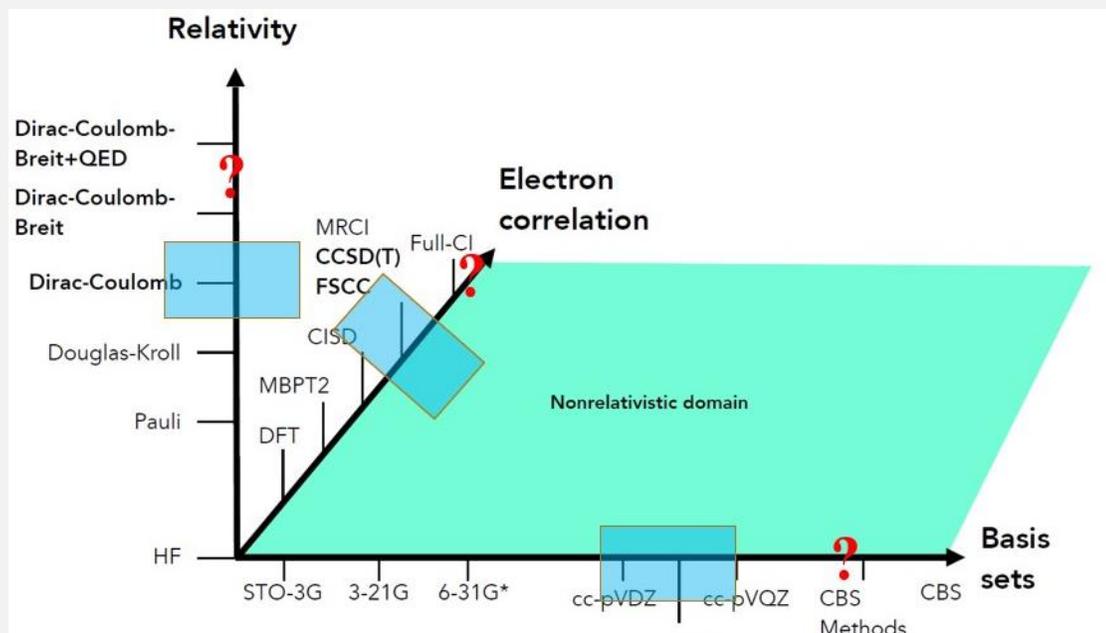


TABLE V: Main sources of uncertainty in the calculated IP of MF (M = Ca, Sr, Ba)[meV].

Category	Error source	CaF	SrF	BaF
Basis set	Cardinality	2.21	1.10	0.50
	augmentation	1.47	0.27	0.37
	inner-core functions	0.01	0.00	2.72
Correlation	virtual space	1.23	0.15	0.44
	higher excitations	0.53	0.84	0.97
Relativity	QED	0.02	0.15	0.52
Total uncertainty, meV		2.97	1.43	3.03

Ionization potentials of CaF, SrF, and BaF

- How accurate are we?
- Excellent agreement for CaF and BaF, where accurate experiment is available
- For SrF, at the time, no accurate experiment was available...

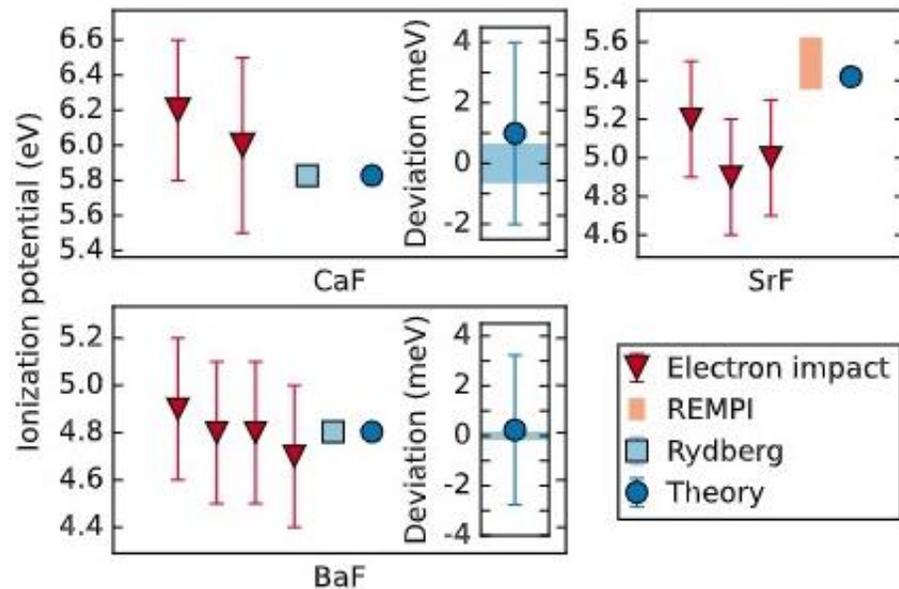


TABLE VI: Recommended theoretical IP of MF (M = Ca, Sr, Ba) [eV] with uncertainties.

IP	CaF	SrF	BaF
Adiabatic	5.821(3)	5.415(1)	4.800(3)
Adiabatic+ZPE	5.828(3)	5.420(1)	4.804(3)
Experiment ^a	5.8270(6)	5.36–5.62	4.80377(1)

^a-CaF[69], SrF[70] and BaF[39]

Ionization potentials of CaF, SrF, and BaF



- How accurate are we?
- Excellent agreement for CaF and BaF, where accurate experiment is available
- For SrF, no accurate experiment is available
- Theoretical determination of the ionization potentials of CaF, SrF, and BaF

A. A. Kyuberis, L. F. Pašteka, E. Eliav, H. A. Perrett, A. Sunaga, S. M. Udrescu, S. G. Wilkins, R. F. Garcia Ruiz, and A. Borschevsky
 Phys. Rev. A **109**, 022813 – Published 22 February 2024

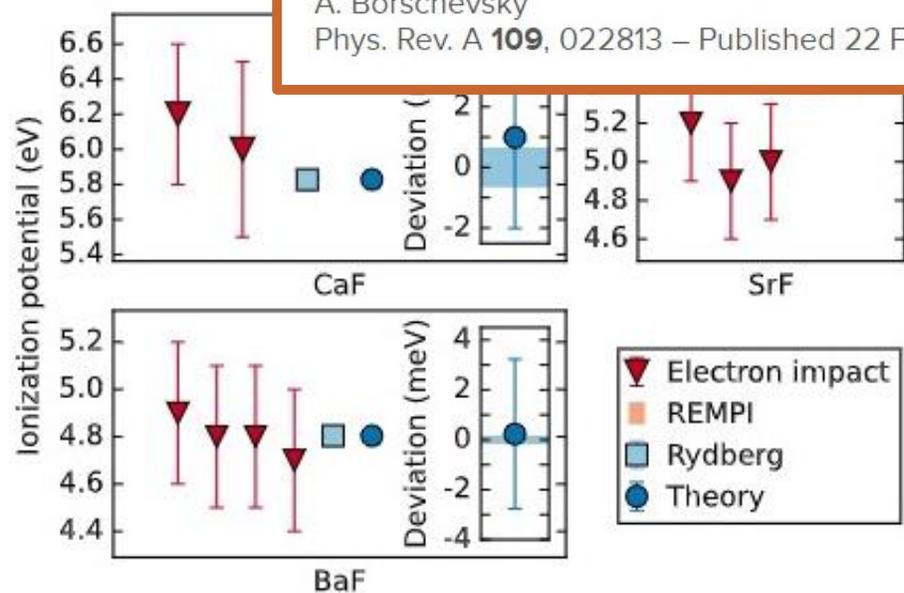


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Back to the ionization potential of RaF

- Measurement: ionization threshold under multi-step laser excitation
- Both two-step and three-step ionization schemes.

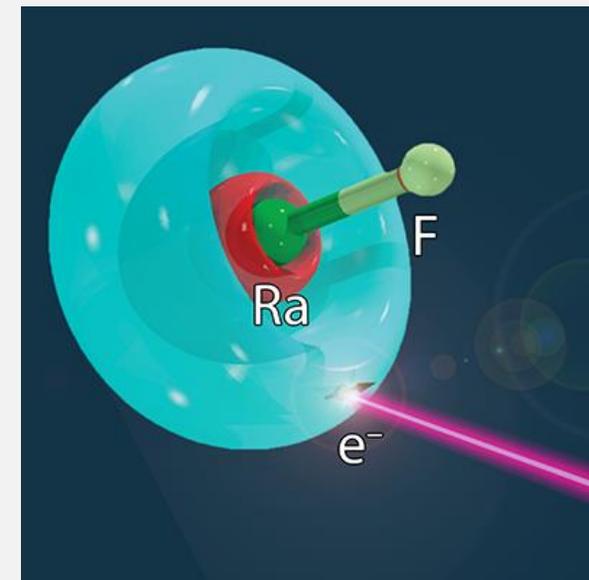
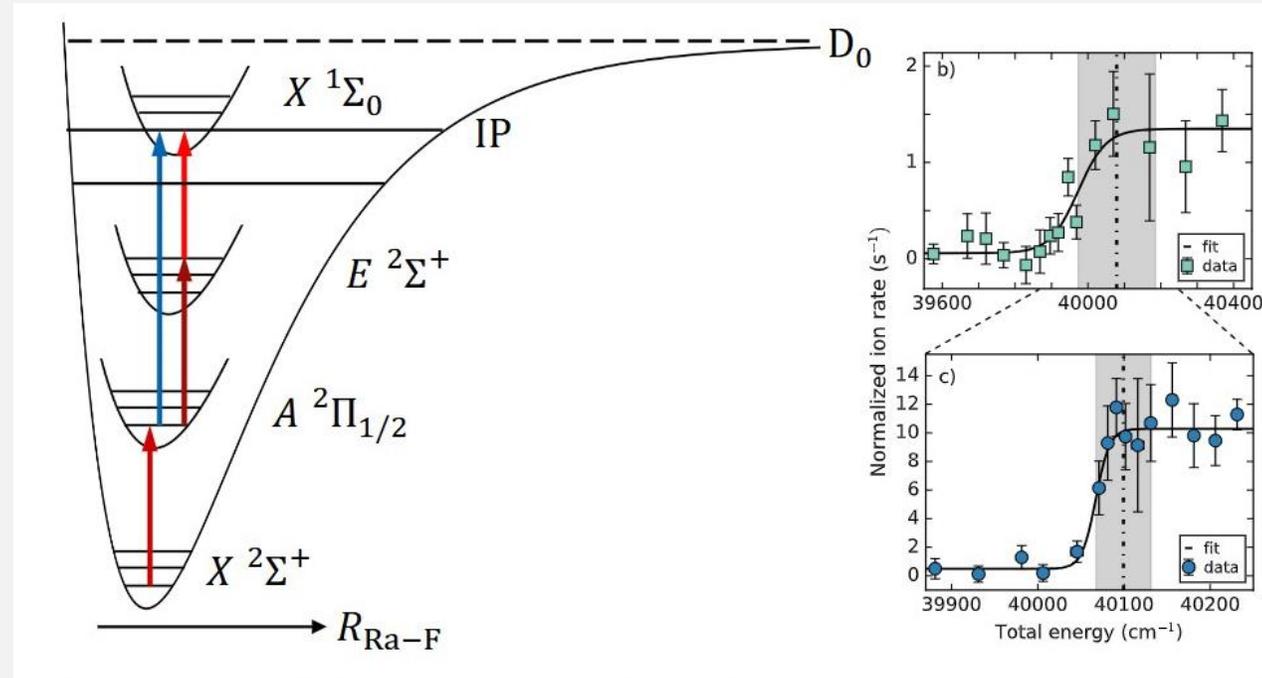


TABLE I. Experimental and calculated IP of RaF

Method	IP (eV)
CBS-DC-CCSD	4.932 ^a
CBS-DC-CCSD(T)	4.983 ^a
+augmentation	4.983
+core corr.+active space	4.989
+ ΔT	4.990
+Breit	4.988
+QED	4.981
Theoretical	4.981(7)
Experimental	



Ionization potential of RaF

- Excellent agreement between theory and experiment
-

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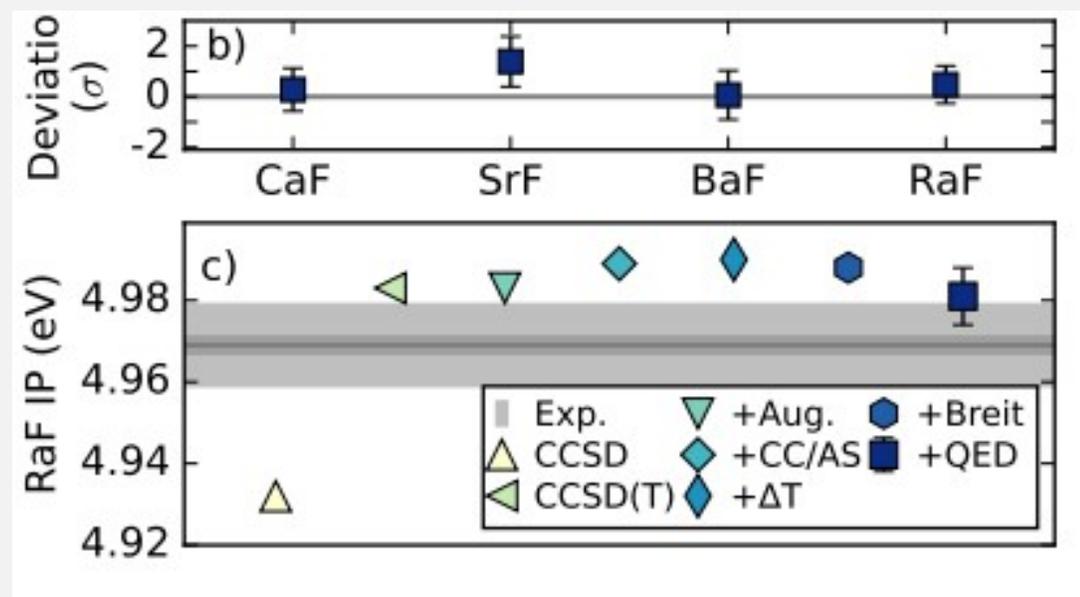
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Ionization potential of RaF

- Excellent agreement between theory and experiment
- Getting the right answer for the right reason

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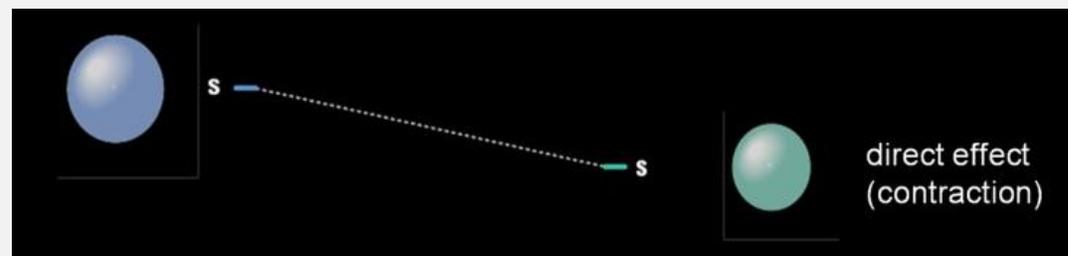
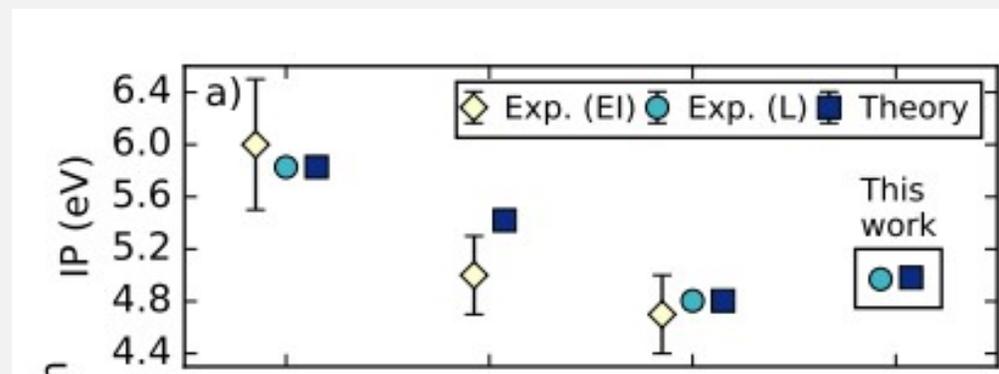


Ionization potential of RaF

- Excellent agreement between theory and experiment
- Getting the right answer for the right reason
- Relativity in action!

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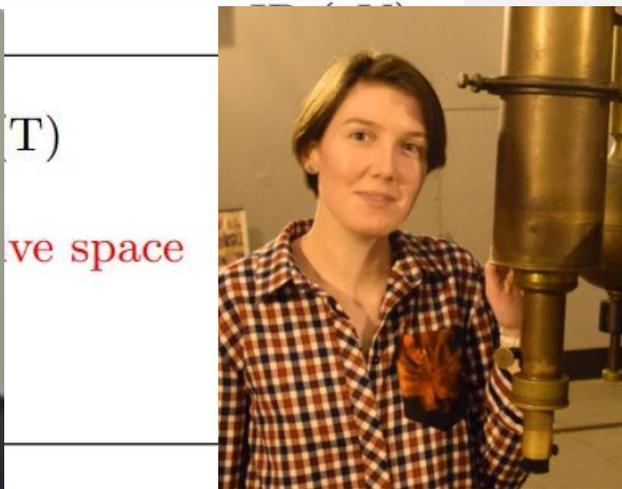
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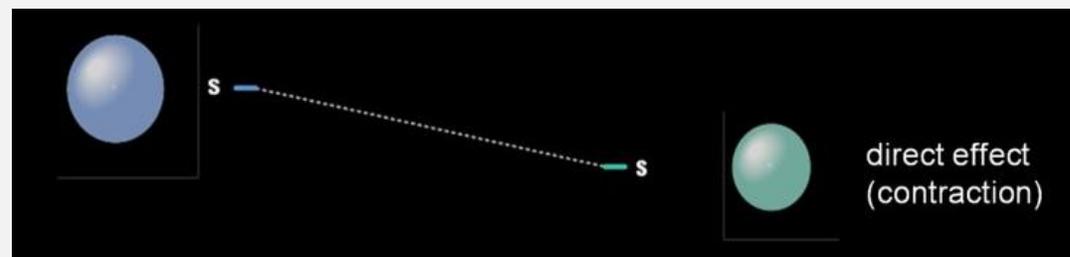
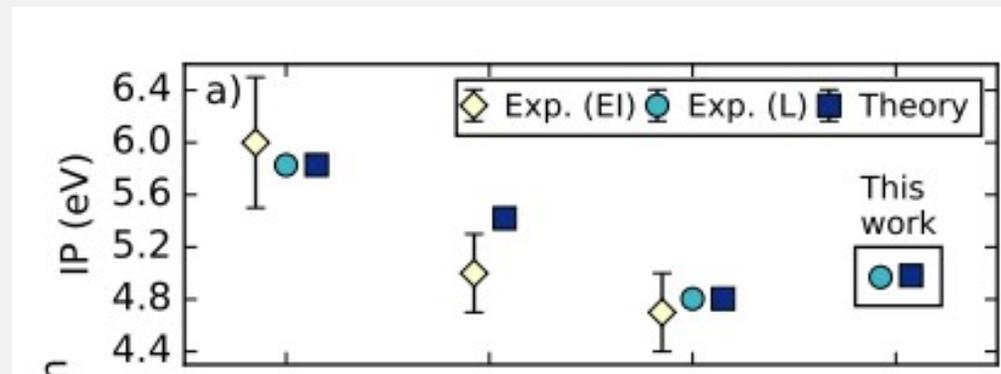
TABLE I. Experimental and calculated IP



Experimental
Shane Wilkins, MIT



4.969(2)[10]
Aleksandra Kuyberis



Ionization potential of RaF

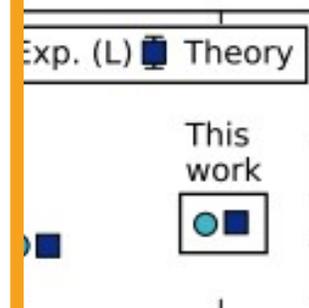
- Exc
- Get
- Rela

PHYSICAL REVIEW RESEARCH 8, L012012 (2026)

Letter

Ionization potential of radium monofluoride

S. G. Wilkins^{1,2,*} H. A. Perrett^{3,†} S. M. Udrescu^{1,‡} A. A. Kyuberis^{4,8} L. F. Pašteka^{4,5} M. Au^{2,6}
 I. Belošević^{1,7} R. Berger⁸ C. L. Binnersley³ M. L. Bissell³ A. Borschevsky⁴ A. A. Breier⁹ A. J. Brinson¹
 K. Chrysalidis² T. E. Cocolios¹⁰ B. S. Cooper³ R. P. de Groot¹⁰ A. Dorne¹⁰ E. Eliav¹¹ R. W. Field¹
 K. T. Flanagan^{12,3} S. Franchoo¹³ R. F. Garcia Ruiz¹ K. Gaul⁸ S. Geldhof¹⁰ T. F. Giesen⁹ F. P. Gustafsson¹⁰
 D. Hanstorp¹⁴ R. Heinke² Á. Koszorús² S. Kujanpää¹⁵ L. Lalanne¹⁰ G. Neyens¹⁰ M. Nichols¹⁴ J. R. Reilly³
 C. M. Ricketts³ S. Rothe² A. Sunaga^{16,17} B. van den Borne¹⁰ A. R. Vernon³ Q. Wang¹⁸ J. Wessolek³
 F. Wienholtz² X. F. Yang¹⁹ Y. Zhou²⁰ and C. Zülch⁸



TABLE

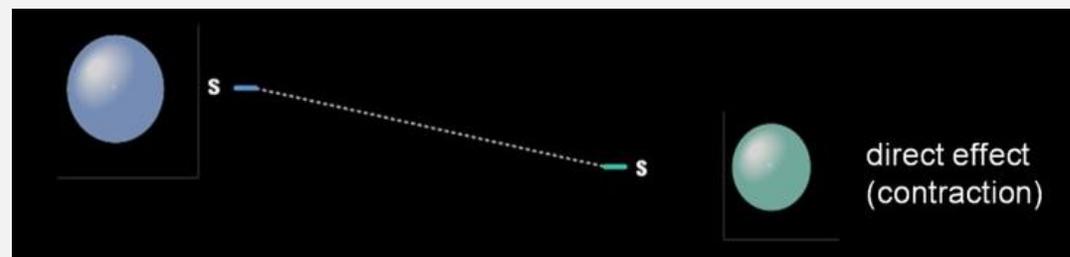


Experimental
Shane Wilkins, MIT

ve space



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Aleksandra Kuyberis



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- How accurate are we?
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- **For SrF, at the time, no accurate experiment was available...**
- Motivated by our work, measurements were performed in the group of M. Heaven
- The measured IP is just 17 cm^{-1} below the predicted value (5.418(2) eV)

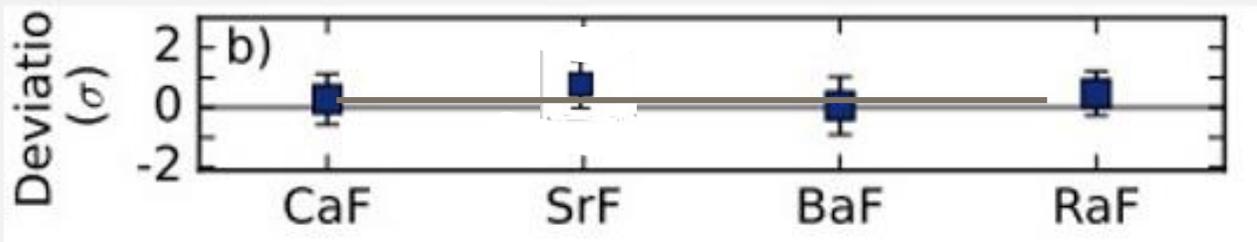


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• *Measurement of the ionization energy of SrF and the vibrational constants of SrF⁺*

Quynh Giao Lindsey, Jiande Han, Jiayue Lin, Aleksandra A. Kyuberis, and Michael C. Heaven

In review

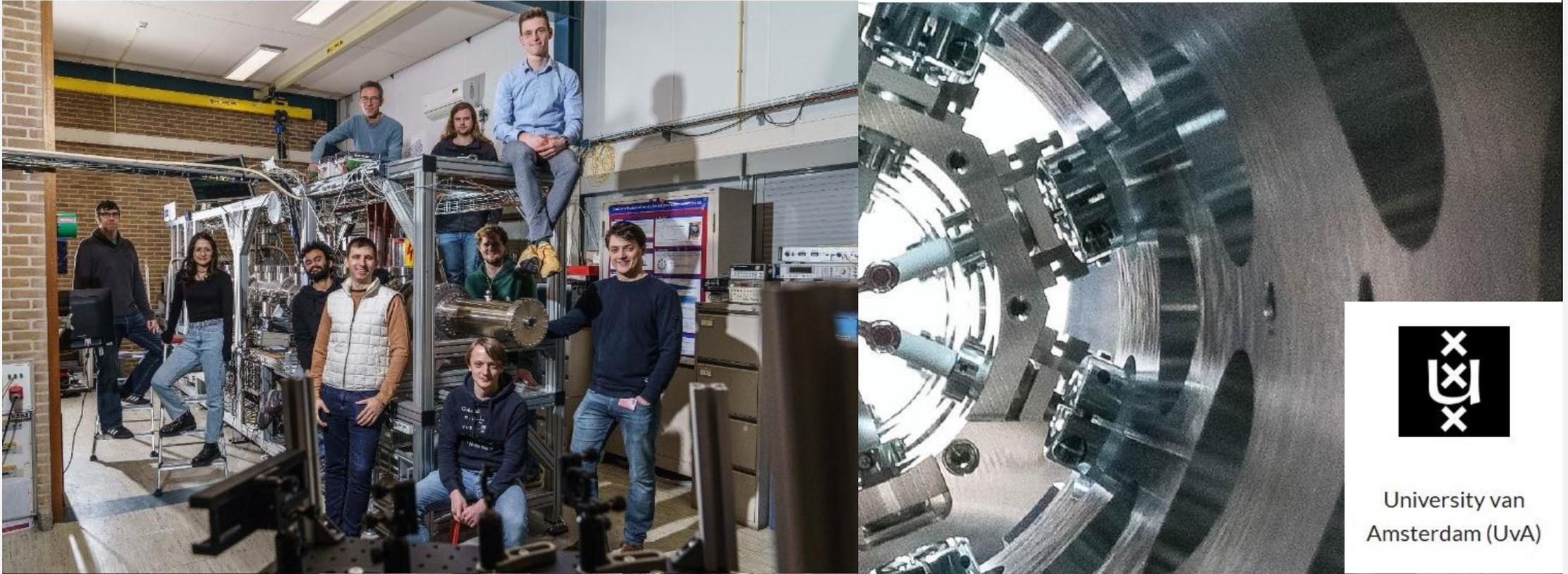
APPLICATIONS

- Ionization potentials of CaF, SrF, BaF, and RaF: meV accuracy, uncertainty estimation, and predictive power
- P,T-violating coupling constants and molecular parameters
 - Search for the electric dipole moment on the electron: the NL-eEDM collaboration
 - Triatomic molecules: BaOH and YbOH

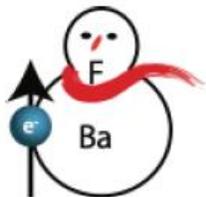
NL-EEDM EXPERIMENT

NL-eEDM

Measuring the electron-EDM with BaF molecules



University van
Amsterdam (UvA)



university of
 groningen
 van swinderen institute for
 particle physics and gravity

Nikhef

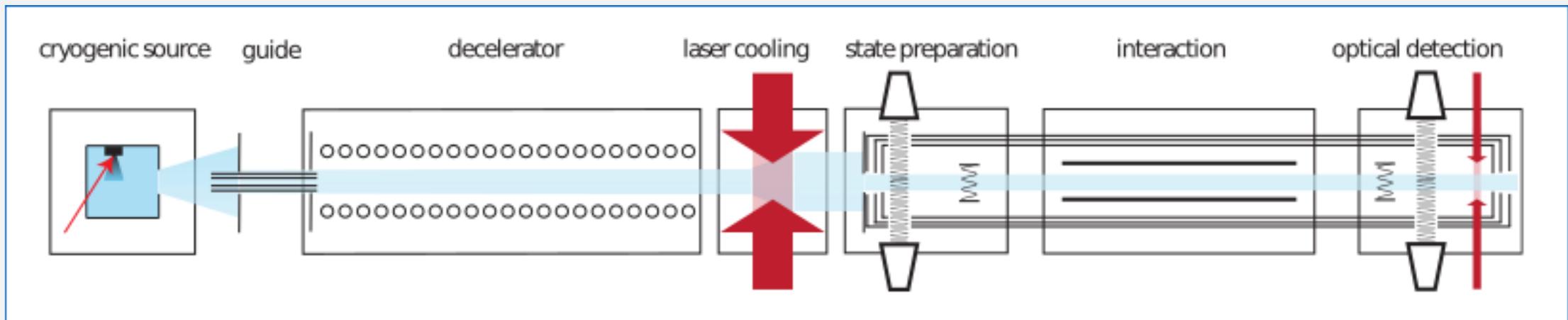
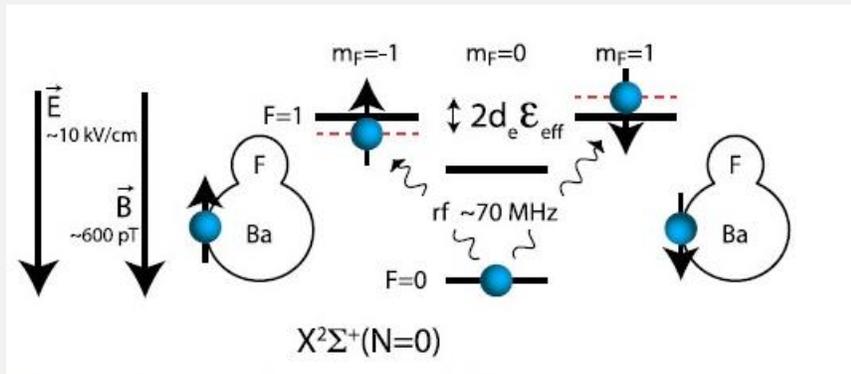
Dutch National Institute for (astro)Particle Physics



VRIJE
UNIVERSITEIT
AMSTERDAM

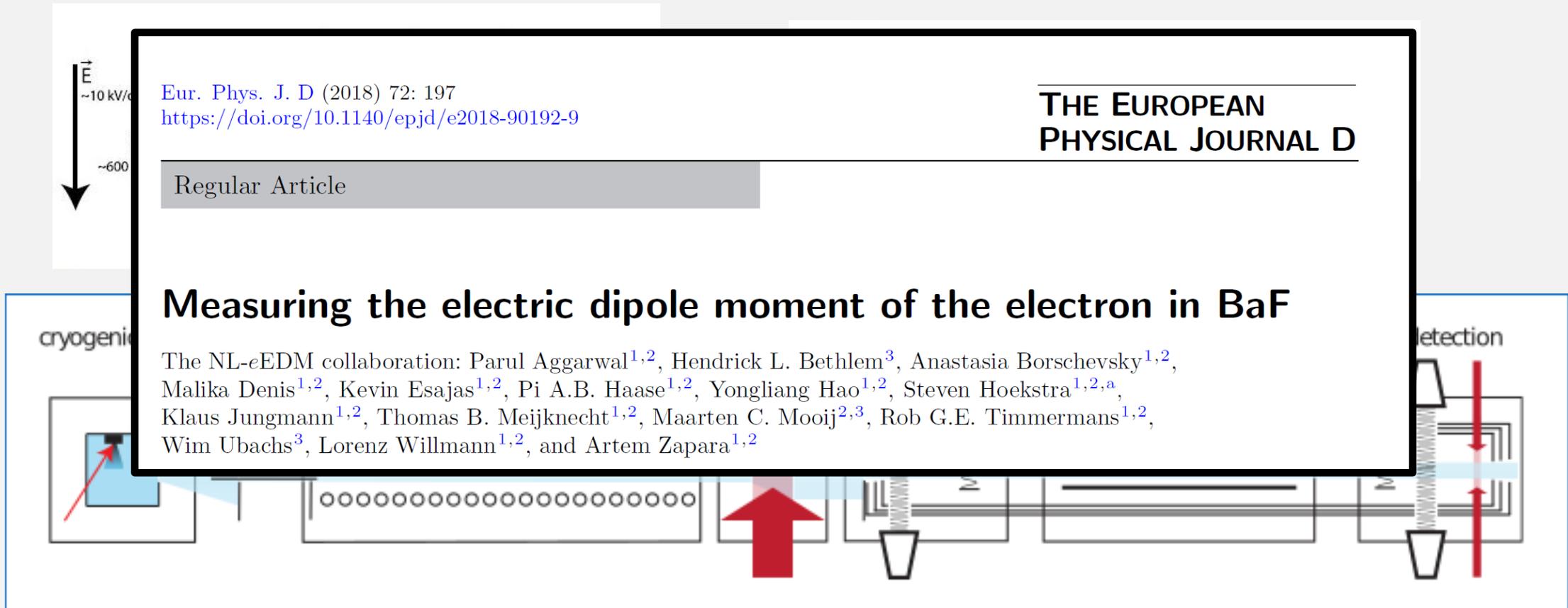
The experiment:

- The EDM signal is detectable through a difference in the total accumulated phase for the parallel and the antiparallel orientation of the external magnetic and the electric fields.



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\vec{E}
~10 kV/cm
~600

Eur. Phys. J. D (2018) 72: 197
<https://doi.org/10.1140/epjd/e2018-90192-9>

THE EUROPEAN
PHYSICAL JOURNAL D

Regular Article

Measuring the electric dipole moment of the electron in BaF

The NL-eEDM collaboration: Parul Aggarwal^{1,2}, Hendrick L. Bethlem³, Anastasia Borschevsky^{1,2}, Malika Denis^{1,2}, Kevin Esajas^{1,2}, Pi A.B. Haase^{1,2}, Yongliang Hao^{1,2}, Steven Hoekstra^{1,2,a}, Klaus Jungmann^{1,2}, Thomas B. Meijknecht^{1,2}, Maarten C. Mooij^{2,3}, Rob G.E. Timmermans^{1,2}, Wim Ubachs³, Lorenz Willmann^{1,2}, and Artem Zapara^{1,2}

cryogenic

detection

The experiment:

- The EDM signal is antiparallel or

Statistics and systematics of electron EDM searches with BaF

The NL-eEDM collaboration: A. Boeschoten^{1,2}, V.R. Marshall^{1,2}, T.B. Meijknecht^{1,2}, A.P. Touwen^{1,2}, P. Aggarwal^{1,2}, N. Balasubramanian^{1,2}, R. Bause^{1,2}, H. L. Bethlem^{1,3}, A. Borschevsky^{1,2}, T.H. Fikkers^{1,2}, P.A.B. Haase^{1,2}, Y. Hao¹, S. Hoekstra^{1,2}, J.W.F. van Hofslot^{1,2}, S.A. Jones^{1,2}, K. Jungmann¹, J.E.J. Levenga^{1,2}, M.C. Mooij^{2,3}, H. Mulder^{1,2}, B.A. Nijman^{1,2}, E.H. Prinsen^{1,2}, B.J. Schellenberg^{1,2}, I.E. Thompson^{1,2}, R.G.E. Timmermans^{1,2}, L. van Sloten^{1,2}, W. Ubachs³, J. de Vries^{2,4}, L. Willmann^{1,2}, Y. Yin^{1,2}

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THE EUROPEAN
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Regular Article

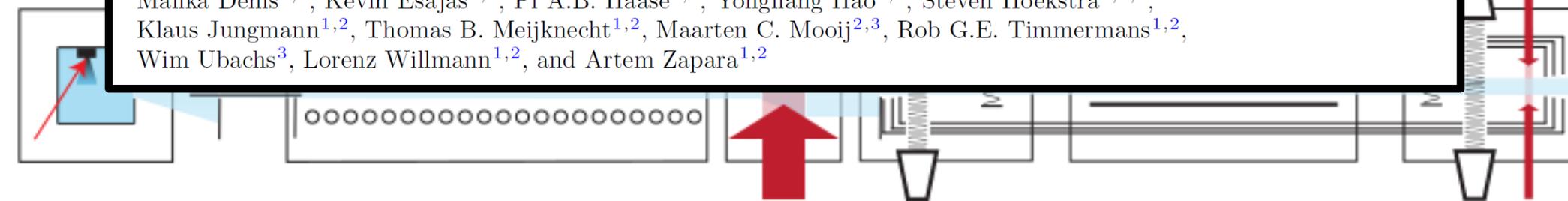
arXiv:2601.21781

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cryogenic

detection



Interpreting the experiment: 2 sources of P, T -effects

$$\hat{H}^{P,T} = (W_d d_e + W_s \kappa_s) \hat{S} \cdot \hat{n}$$

$$\Delta E = (W_d d_e + W_s \kappa_s) \langle \hat{S} \cdot \hat{n} \rangle (E_{\text{ext}})$$

What we want

d_e - electron EDM, κ_s - S - PS electron-nucleon interaction

measured

Have to come from calculations

Electric dipole moments: A global analysis

Timothy Chupp and Michael Ramsey-Musolf
 Phys. Rev. C **91**, 035502 – Published 6 March 2015

Investigations of optimal combinations of molecules:

- K. Gaul, S. Marquardt, T. Isaev, and R. Berger, Phys. Rev. A **99**, 032509 (2019)
- A. Sunaga, M. Abe, V. S. Prasanna, T. vAoki, and M. Hada, J. Phys. B **53**, 015102 (2019)

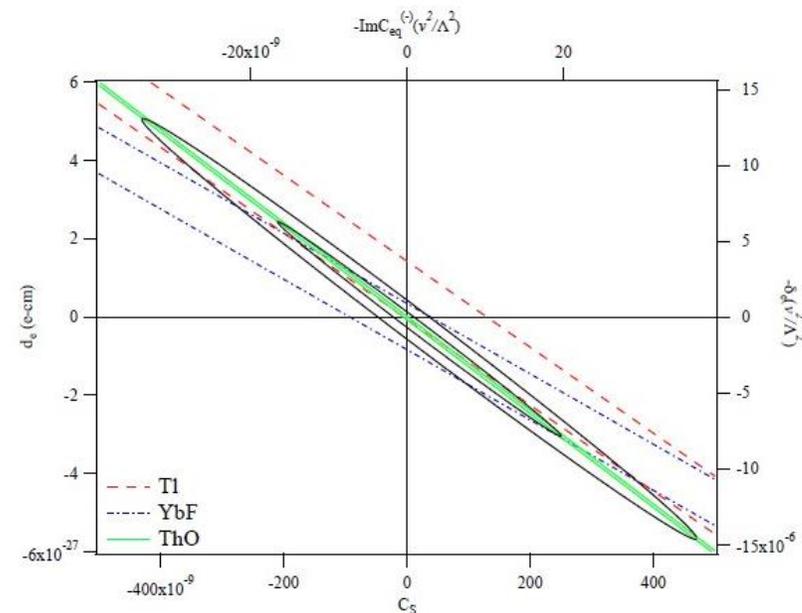


FIG. 1: Electron edm d_e as a function of C_5 from the experimental results in Tl, YbF and ThO. Also shown are 68% and 95% error ellipses representing the best-fit for the paramagnetic systems and including $d_A(^{199}\text{Hg})$ as discussed in the text. Also shown are the constraints on the dimensionless Wilson coefficients δ_e and $\text{Im} C_{eq}^{(-)}$ times the squared scale ratio $(v/\Lambda)^2$.

Interpreting the experiment: 2 sources of P, T -effects

$$\hat{H}^{P,T} = (W_d d_e + W_s k_s) \hat{\mathbf{S}} \cdot \hat{\mathbf{n}}$$

$$H^{\text{EDM}} = 2cd_e \sum_{i=1}^n v \gamma^0(i) \gamma^5(i) \mathbf{p}^2(i)$$

$$W_d = \frac{1}{d_e} \langle \psi_\Omega | H^{\text{EDM}} | \psi_\Omega \rangle$$

$$H_S = i \frac{G_F}{\sqrt{2}} Z \mathcal{K}_S \sum_{i=1}^n \gamma^0 \gamma^5 \rho_A(r_i)$$

$$W_S = \frac{1}{\mathcal{K}_S \Omega} \langle \psi_\Omega | H_S | \psi_\Omega \rangle$$

- Use relativistic CC (Finite Field) to calculate W_d and W_s .
- Systematically improve the calculation up to convergence
- Perform an extensive computational study to estimate uncertainties

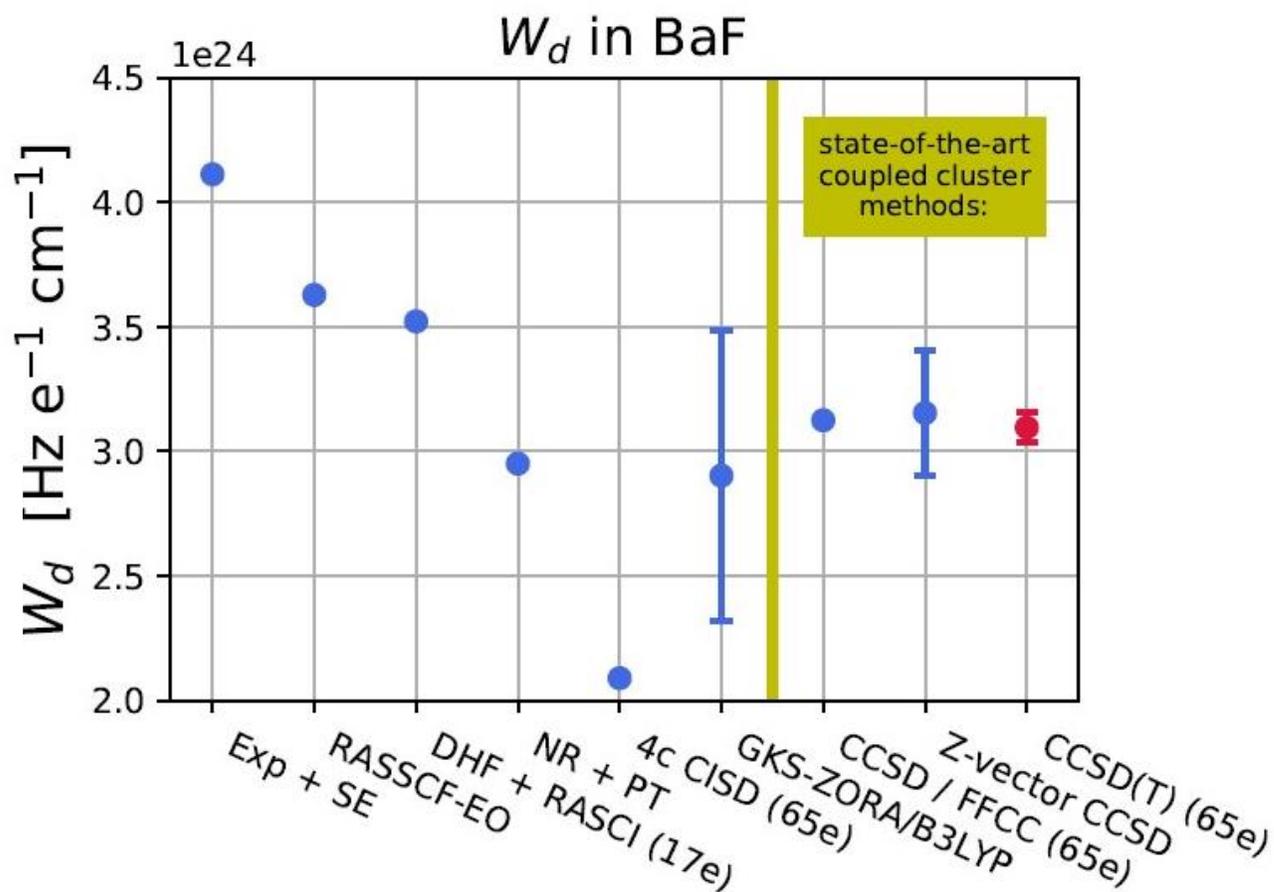
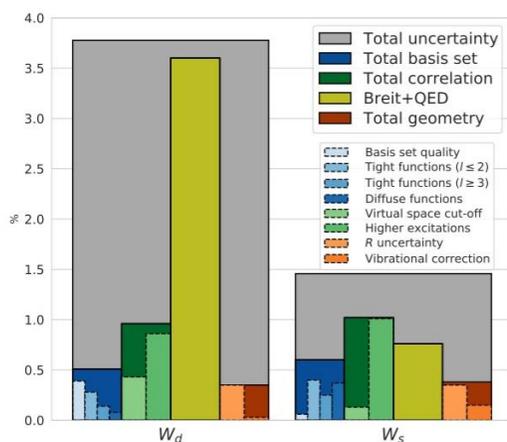
Interpreting the experiment

- Final recommended values:

BaF	W_d^* [$\frac{10^{24}\text{Hz}}{\text{e cm}}$]	W_s [Hz]
DC CCSD(T)	3.13 ± 0.24	8.29 ± 0.12



Pi Haase



Interpreting the experiment

- Final recom

BaF	W_d^* [
DC CCSD(T)	3.13

The Journal of
Chemical Physics

Systematic study and uncertainty evaluation of P , T -odd molecular enhancement factors in BaF

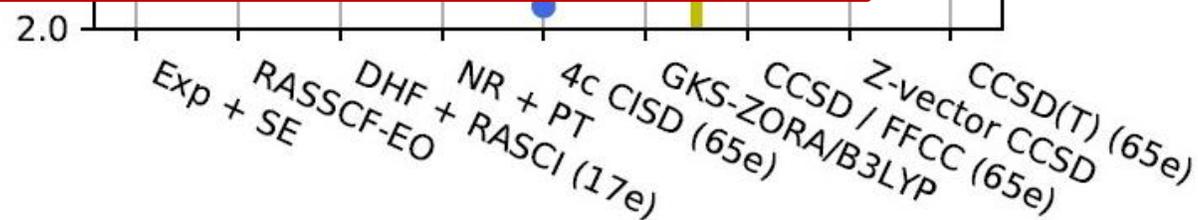
Cite as: J. Chem. Phys. 155, 034309 (2021); <https://doi.org/10.1063/5.0047344>

Submitted: 12 February 2021 • Accepted: 17 June 2021 • Published Online: 21 July 2021

 Pi A. B. Haase,  Diewertje J. Doeglas,  Alexander Boeschoten, et al.

COLLECTIONS

Paper published as part of the special topic on [Special Collection in Honor of Women in Chemical Physics and Physical Chemistry](#)



Pi Haase

Interpreting the experiment

- Is the uncertainty estimate realistic?

How do we assign uncertainties?

- Calculate the same property in a lighter homologue (and compare to experiment)
- Calculate a different property in the same atom/molecule (and compare to experiment)
- Perform a computational investigation to assign uncertainties based on method incompleteness

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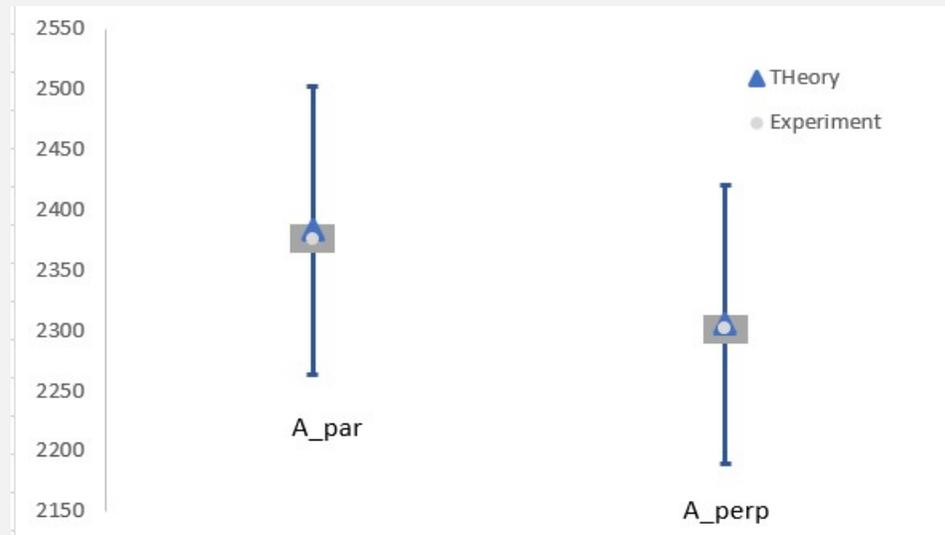
Interpreting the experiment

- Is the uncertainty estimate realistic?
- Use a similar property, where comparison to experiment is possible, as a sanity check
- Magnetic hyperfine structure constants

BaF	$W_d^* [\frac{10^{24}\text{Hz}}{e\text{ cm}}]$	$W_s [\text{Hz}]$
DC CCSD(T)	3.13 ± 0.24	8.29 ± 0.12

Table 8. A_{\parallel} and A_{\perp} of ^{137}Ba in BaF (MHz)

method	^{137}BaF			
	A_{\parallel}	%(exp)	A_{\perp}	%(exp)
GRECP SCF-EO ⁹⁰	2264	-4.71	2186	-5.00
GRECP RASSCF-EO ⁹⁰	2272	-4.38	2200	-4.39
DF RASCI ⁹¹	2240	-5.72	2144	-6.82
DF MBPT ⁹¹	2314	-2.61	2254	-2.04
DC CCSD (this work)	2383(129)	0.29	2305(132)	0.17
exp ⁷⁷	2376(12)		2301(9)	



Interpreting the experiment

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Theoretical uncertainty estimation

CCSD		exp
^{137}BaF		
A_{\parallel}	2383	2376
A_{\perp}	2305	2301
^{133}Cs		
A	2283	2298

ACS AUTHORCHOICE

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Hyperfine Structure Constants on the Relativistic Coupled Cluster Level with Associated Uncertainties

Pi A. B. Haase,* Ephraim Eliav, Miroslav Iliaš, and Anastasia Borschevsky

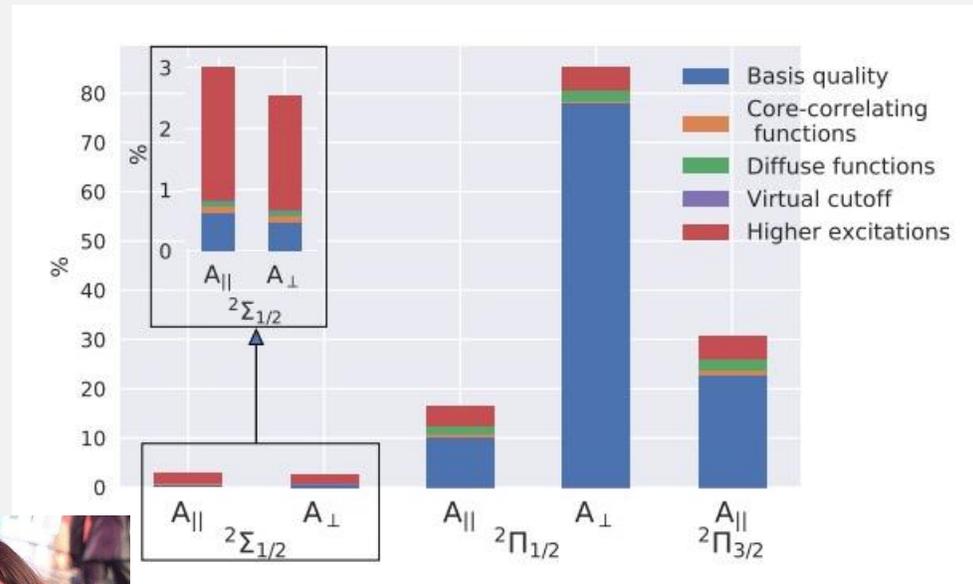
Cite This: *J. Phys. Chem. A* 2020, 124, 3157–3169

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Hyperfine structure constants: predictive power

- Calculations of excited state HFS parameters in molecules are rare
- We looked at the F HFS coupling constant in the ground and excited states of BaF

State		Theory	Experiment
$^2\Sigma_{1/2}$	A_{\parallel}	CCSD(T) 71.22 ± 2.14	71.73 ± 0.09^a
	A_{\perp}	" 64.02 ± 1.62	63.51 ± 0.03^a
$^2\Pi_{1/2}$	A_{\parallel}	FSCC 51.97 ± 8.25	58.64 ± 0.27^b
	A_{\perp}	" 2.02 ± 1.51	3.58 ± 0.14^b
$^2\Pi_{3/2}$	A_{\parallel}	FSCC 13.40 ± 3.86	15.85 ± 0.09^b



Malika Denis



Pi Haase



Yuly Chamorro

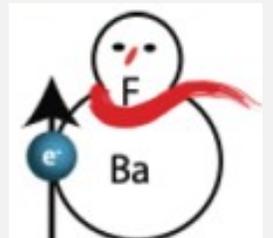
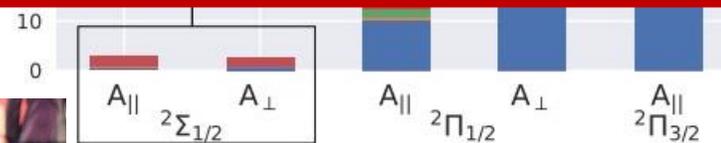
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Benchmarking of the Fock-space coupled-cluster method and uncertainty estimation: Magnetic hyperfine interaction in the excited state of BaF

Malika Denis, Pi A. B. Haase, Maarten C. Mooij, Yuly Chamorro, Parul Aggarwal, Hendrick L. Bethlem, Alexander Boeschoten, Anastasia Borschevsky, Kevin Esajas, Yongliang Hao, Steven Hoekstra, Joost W. F. van Hofslot, Virginia R. Marshall, Thomas B. Meijknecht, Rob G. E. Timmermans, Anno Touwen, Wim Ubachs, Lorenz Willmann, and Yanning Yin (NL-eEDM Collaboration)
 Phys. Rev. A **105**, 052811 – Published 31 May 2022

State		
$^2\Sigma_{1/2}$	A_{\parallel}	CCSD
	A_{\perp}	"
$^2\Pi_{1/2}$	A_{\parallel}	FSCC
	A_{\perp}	"
$^2\Pi_{3/2}$	A_{\parallel}	FSCC



Malika Denis



Pi Haase



Yuly Chamorro

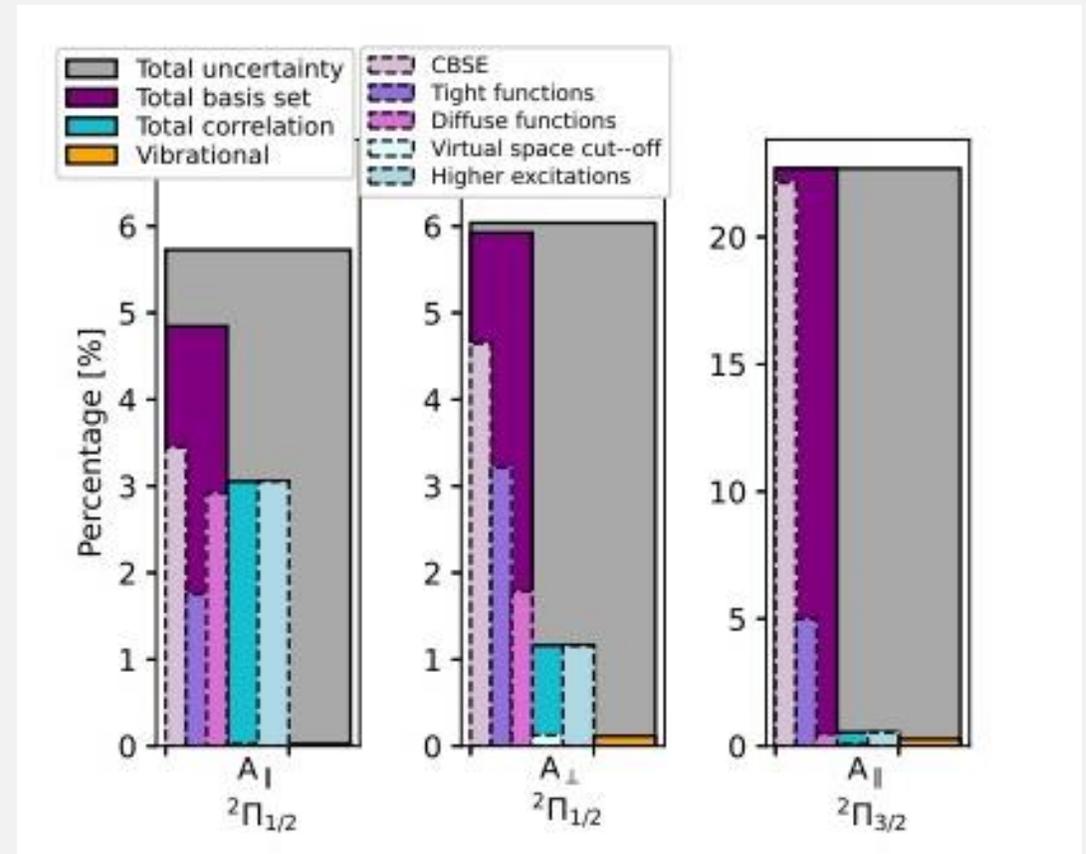
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	$^2\Pi_{1/2}$		$^2\Pi_{3/2}$
	A_{\parallel}	A_{\perp}	A_{\parallel}
Baseline CBS(H)	394.5	242.8	36.6
Corrections			
Diffuse functions	-7.0	-7.7	-1.7
Tight functions	11.6	4.3	-0.2
Vibrational effects	-0.1	-0.3	-0.1
Final results	399.0 ± 22.9	239.1 ± 14.4	34.6 ± 7.9
Experiment	413.2 ± 0.6	254.3 ± 0.5	–



Yuly Chamorro





Tim Langen & Felix Kogel,
TU Vienna

OPEN ACCESS

Magnetic hyperfine structure constants of ^{137}BaF in the $^2\Pi_{1/2}$ and $^2\Pi_{3/2}$ excited states

[Yuly Chamorro](#) ^{1,*}, [Felix Kogel](#) ², [Tim Langen](#) ^{2,3}, and [Anastasia Borschevsky](#) ¹

Show more ▾

Phys. Rev. A **112**, 042808 – Published 9 October, 2025

Exp

Final results	399.0 ± 22.9	239.1 ±
Experiment	413.2 ± 0.6	254.3 ±

OPEN ACCESS

High-resolution spectroscopy of barium monofluoride: Odd isotopologues, hyperfine structure, and isotope shifts

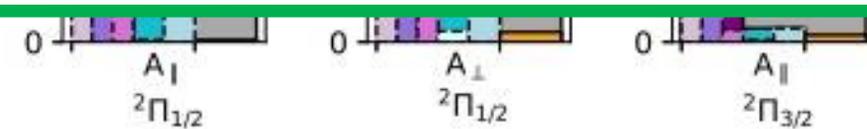
[Felix Kogel](#) ¹, [Yuly Chamorro](#) ², [Mangesh Bhattarai](#) ^{3,4}, [Marian Rockenhäuser](#) ^{1,5}, [Tatsam Garg](#) ^{1,5}, [David DeMille](#) ^{3,4,6}, [Anastasia Borschevsky](#) ², and [Tim Langen](#) ^{1,5,*}

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Phys. Rev. A **112**, 042807 – Published 9 October, 2025



Yuly Chamorro

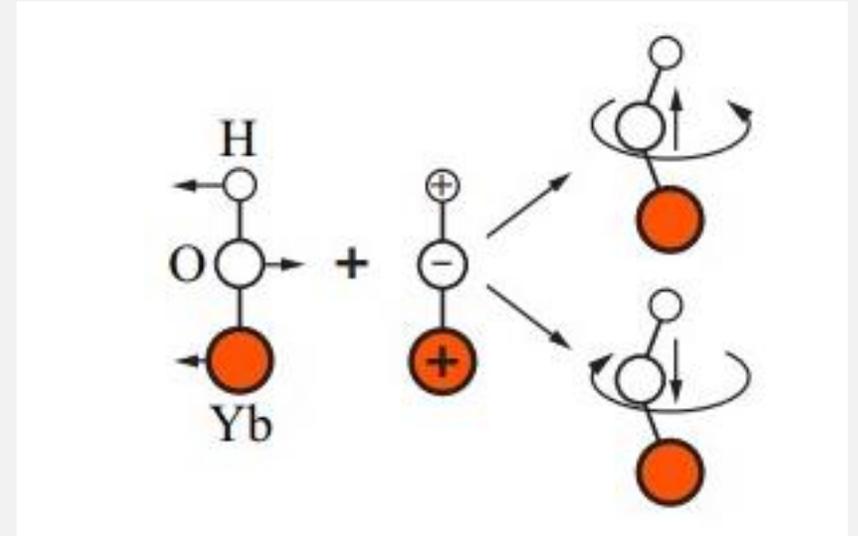


APPLICATIONS

- Ionization potentials of CaF, SrF, BaF, and RaF: meV accuracy, uncertainty estimation, and predictive power
- P,T-violating coupling constants and molecular parameters
 - Search for the electric dipole moment on the electron: the NL-eEDM collaboration
 - **Triatomic molecules: BaOH and YbOH**

TRIATOMIC MOLECULES

- Polyatomic molecules emerge as good candidates for eEDM experiments due to the long-lived close-lying opposite parity eigenstates
- **BaOH** and YbOH are also expected to be laser-coolable
- What about sensitivity to BSM physics?



- Polyatomic molecules emerge as good candidates for eEDM experiments due to the long-lived close-lying opposite parity eigenstates (*I*-doublets)
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- What about sensitivity to BSM physics?
- Sensitivity to eEDM and to *P,T*-violating nuclear magnetic quadrupole moment (NMQM) found to be similar to that of diatomic homologues

	$W_d [10^{24}\text{Hz}/(\text{e}\cdot\text{cm})]$	$W_M [10^{23}\text{Hz}/(\text{e}\cdot\text{cm}^2)]$
BaF	3.13	-0.385
BaOH	3.21	-0.392
YbF	23.37	-1.055
YbOH	23.56	-1.067



Malika Denis

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- Accuracy test: HFS constants:

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TABLE VII. The A_{\parallel} constants of the heavy nucleus calculated using the model optimized for W_M given in (MHz) and comparison with the available experimental data.

System	Calculated A_{\parallel}	Expt. A_{\parallel}	Δ (%)
$^{137}\text{BaOH}$	2194.6	2200.2 ⁴²	0.3
^{171}YbF	7579.0	7429.1 ⁴³	2
^{173}YbF	-2087.6	-2060.0 ⁴⁴	1.3
$^{171}\text{YbOH}$	7174.9		
$^{173}\text{YbOH}$	-1976.3		

- Polyatomic molecules emerge as good candidates for experiments involving lived close-lying opposite parity eigenstates (I -doublets)
- BaOH and YbOH are also expected to be laser-coolable
- What about sensitivity to BSM physics?
- Sensitivity to eEDM and to P, T -violating nuclear moments

Enhanced P, T -violating nuclear magnetic quadrupole moment effects in laser-coolable molecules

Cite as: J. Chem. Phys. 152, 084303 (2020); <https://doi.org/10.1063/1.5141065>

Submitted: 03 December 2019 . Accepted: 06 February 2020 . Published Online: 25 February 2020

Malika Denis, Yongliang Hao, Ephraim Eliav, Nicholas R. Hutzler, Malaya K. Nayak, Rob G. E. Timmermans, and Anastasia Borschevsky

Enhancement factor for the electric dipole moment of the electron in the BaOH and YbOH molecules

Malika Denis^{1,*}, Pi A. B. Haase¹, Rob G. E. Timmermans¹, Ephraim Eliav², Nicholas R. Hutzler³, and Anastasia Borschevsky¹

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Phys. Rev. A 99, 042512 – Published 19 April, 2019

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Fine and hyperfine interactions in $^{171}\text{YbOH}$ and $^{173}\text{YbOH}$

J. Chem. Phys. **154**, 244309 (2021); <https://doi.org/10.1063/5.0055293>

 Nickolas H. Pilgram¹,  Arian Jadbabaie¹,  Yi Zeng¹,  Nicholas R. Hutzler¹, and  Timothy C. Steimle^{2,a)}

- Polyatomic molecules with lived close-lying opposite parity eigenstates (*I*-doublets)

• BaOH and YbOH

• What about YbOH?

• Sensitivity to hyperfine structure calculated a

• Accuracy to

Table II: Comparison of the measured hyperfine parameters of the $\tilde{X}^2\Sigma^+(0,0,0)$ state of $^{171,173}\text{YbOH}$ to calculated values

Isotopologue	Parameter	Measured (MHz)	Theory Ref. ¹⁶ (MHz)	Theory Ref. ⁵¹ cGHF(MHz)	Theory Ref. ⁵¹ cGKS(MHz)	Theory Ref. ⁵² (MHz)
$^{171}\text{YbOH}$	A_{\parallel}^a	6979 (35)	7174.9			
$^{171}\text{YbOH}$	A_{\perp}^b	6745(15)				
$^{173}\text{YbOH}$	A_{\parallel}	-1929(11)	-1976.3	-1600	-1300	
$^{173}\text{YbOH}$	A_{\perp}	-1856 (5)		-1600		
$^{173}\text{YbOH}$	e^2Qq_0	-3319 (48)	-3502			-3492

(MQM)

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- Static and dynamic polarizability in BaOH
- Important in the context of trapping

	α_{\parallel} (a.u.)	α_{\perp} (a.u.)
Static	200.8(24)	297(5)
Dynamic ($\lambda=1064\text{nm}$)	357(31)	714(29)

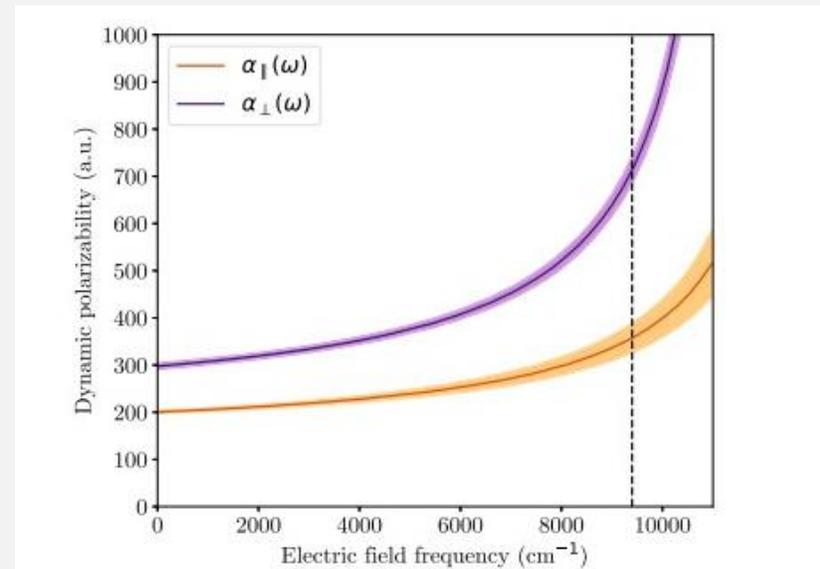


FIG. 2. Parallel and perpendicular components of the dynamic polarizability (solid lines) with uncertainty bounds (shaded regions) as a function of the frequency of the external electric field. The dashed line corresponds to the 1064-nm laser.

- Static and dynamic polarizability in BaOH
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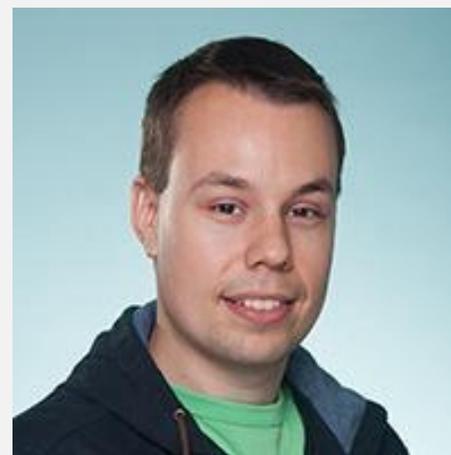
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Eifion Prinsen



Agustin Aucar



Lukas Pasteka

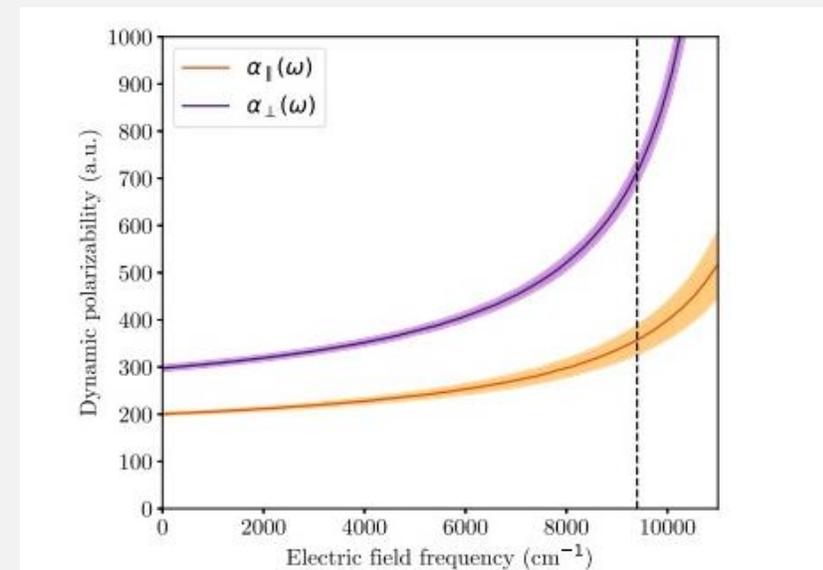


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- Static
- Impo

PHYSICAL REVIEW A **113**, 022809 (2026)

Ab initio calculations of the static and dynamic polarizability of BaOH

Eifion H. Prinsen^{1,2}, Anastasia Borschevsky^{1,2}, Steven Hoekstra^{1,2}, Achintya K. Dutta³, Sudipta Chakraborty,³
Bart J. Schellenberg^{1,2}, Lukáš F. Pašteka^{1,2,4,*} and I. Agustín Aucar^{1,2,5,†}

¹Van Swinderen Institute for Particle Physics and Gravity, University of Groningen, The Netherlands

²Nikhef, National Institute for Subatomic Physics, Amsterdam, The Netherlands

³Department of Chemistry, Indian Institute of Technology Bombay, Mumbai 400076, India

⁴Department of Physical and Theoretical Chemistry, Comenius University, Bratislava, Slovakia

⁵Instituto de Modelado e Innovación Tecnológica, Facultad de Ciencias Exactas y Naturales y Agrimensura, Universidad Nacional del Nordeste, Corrientes, Argentina

Static

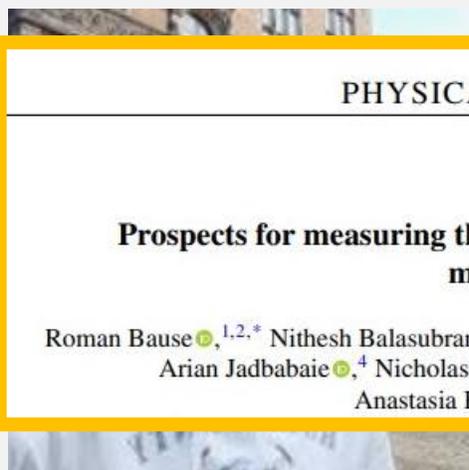
Dynamic ($\lambda=1064\text{nm}$)

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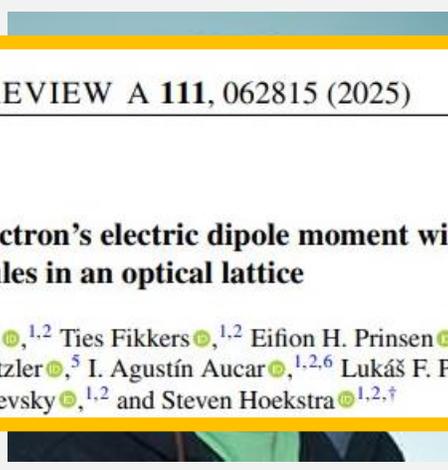
714(29)



Eifion Prinsen



Agustin Aucar



Lukas Pasteka

PHYSICAL REVIEW A **111**, 062815 (2025)

Prospects for measuring the electron's electric dipole moment with polyatomic molecules in an optical lattice

Roman Bause^{1,2,*}, Nithesh Balasubramanian^{1,2}, Ties Fikkers^{1,2}, Eifion H. Prinsen^{1,2}, Kees Steinebach³,
Arian Jadbabaie⁴, Nicholas R. Hutzler⁵, I. Agustín Aucar^{1,2,6}, Lukáš F. Pašteka^{1,2,7},
Anastasia Borschevsky^{1,2} and Steven Hoekstra^{1,2,†}

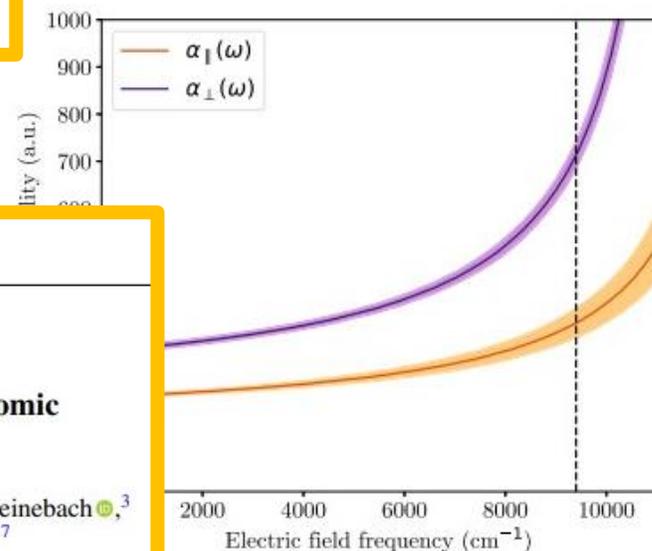


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- Impo

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¹Van Swinderen Institute for Particle Physics and Gravity, University of Groningen, The Netherlands

See talk by Steven Hoekstra, Thursday morning:

09:45

→ 10:15

Precision measurements to test fundamental physics using trapped molecules

Speaker: Steven Hoekstra (University of Groningen, and Nikhef)

PHYSICAL REVIEW A **111**, 062815 (2025)

Prospects for measuring the electron's electric dipole moment with polyatomic molecules in an optical lattice

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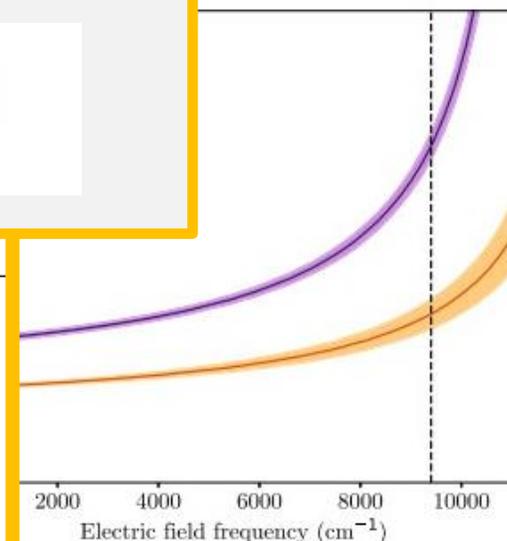
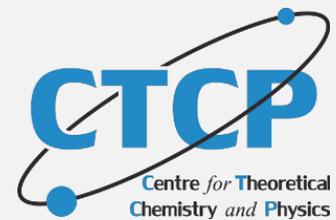


FIG. 2. Parallel and perpendicular components of the dynamic polarizability (solid lines) with uncertainty bounds (shaded regions) as a function of the frequency of the external electric field. The dashed line corresponds to the 1064-nm laser.

CONCLUSIONS

- State of the art high accuracy computational approach
- Versatile methods: many possible applications
- Reliable predictions, uncertainty estimates possible
- Close collaborations with experimental groups



Ephraim Eliav



Miroslav Ilias



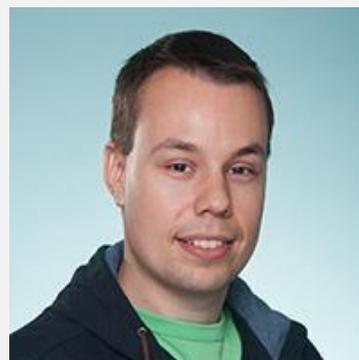
Peter Schwerdtfeger



Jacinda Ginges



Victor Flambaum



Lukas Pasteka



Konstantin Gaul

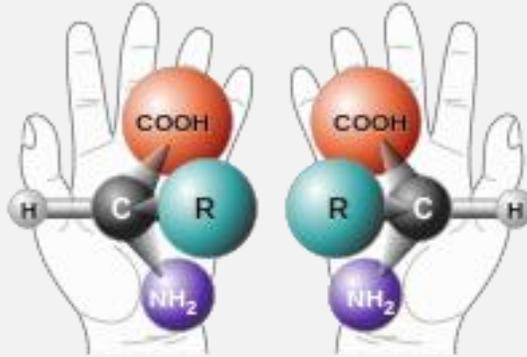


And now
for something
completely different...



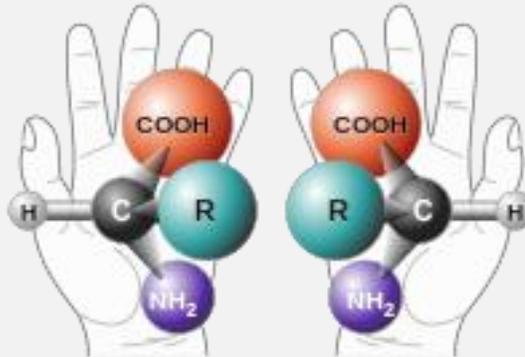
PARITY VIOLATION IN CHIRAL MOLECULES

- **Is there a difference in the properties of the right- and the left handed enantiomers?**



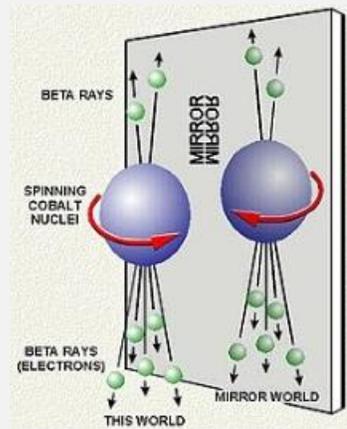
•

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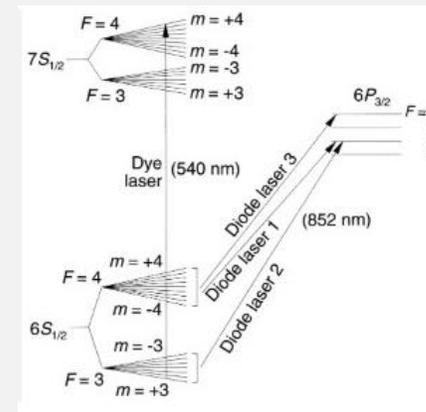


- PV is firmly established in nuclear and atomic physics

-

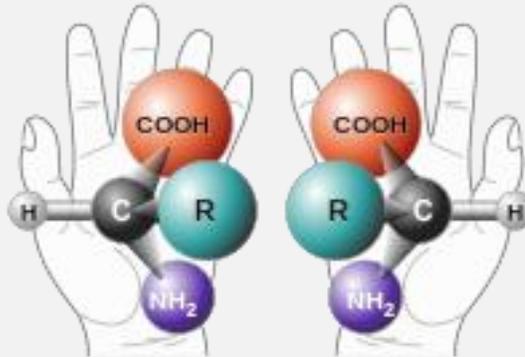


Decay of ⁶⁰Co



Forbidden 6s-7s transition in atomic Cs

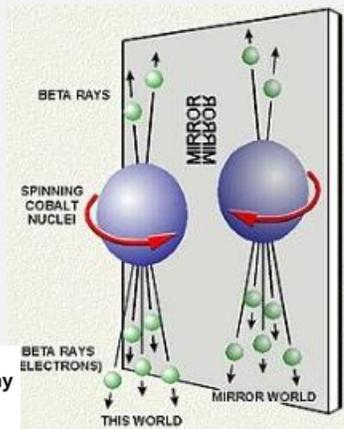
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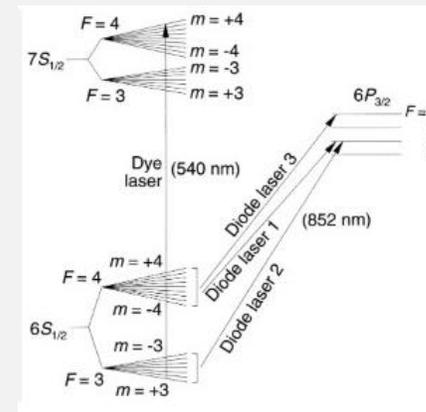
No longer measuring zero!

- PV is firmly established in nuclear and atomic physics

-



Decay of ^{60}Co



Forbidden 6s-7s transition in atomic Cs

Science

HOME > SCIENCE > VOL. 275, NO. 5307 > MEASUREMENT OF PARITY NONCONSERVATION AND AN ANAPOLE MOMENT IN CESIUM

RESEARCH ARTICLES

Measurement of Parity Nonconservation and an Anapole Moment in Cesium

C. S. WOOD, S. C. BENNETT, D. CHO, B. P. MASTERSON, J. L. ROBERTS, C. E. TANNER, AND C. E. WIEMAN [Authors Info & Affiliations](#)

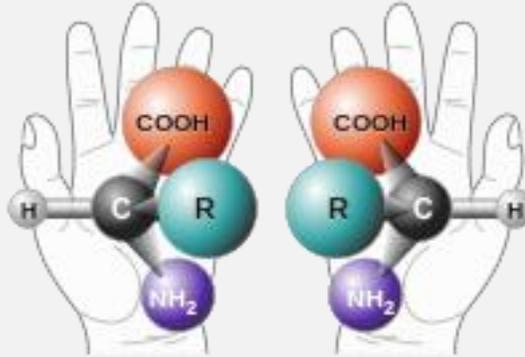
SCIENCE • 21 Mar 1997 • Vol 275, Issue 5307 • pp. 1759-1763 • DOI:10.1126/science.275.5307.1759

Experimental Test of Parity Conservation in Beta Decay

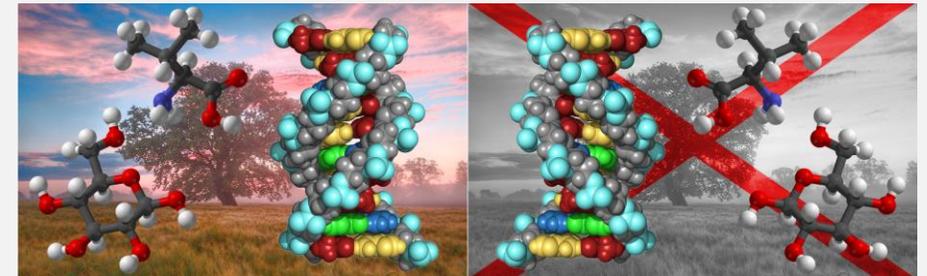
C. S. Wu
F. Ambler, R. W. Hayward, D. D. Hoppes, and R. P. Hudson

Show more
Phys. Rev. 105, 1413 - Published 15 February, 1957

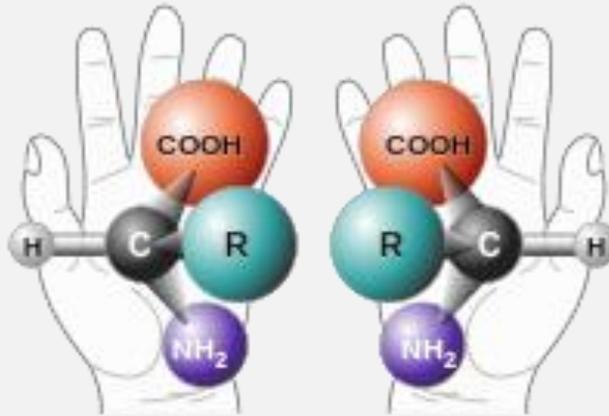
- **Is there a difference in the properties of the right- and the left handed enantiomers?**



- PV is firmly established in nuclear and atomic physics
- In chiral molecules, the weak neutral current between the electrons and the nuclei is predicted to result in a tiny energy difference between the enantiomers.
- If detected, this could
 - Constitute a test of the SM
 - Help explain the origins of biohomochirality
 - Allow deeper understanding of electronic structure effects



- Is there a difference in the properties of the right- and the left handed enantiomers?

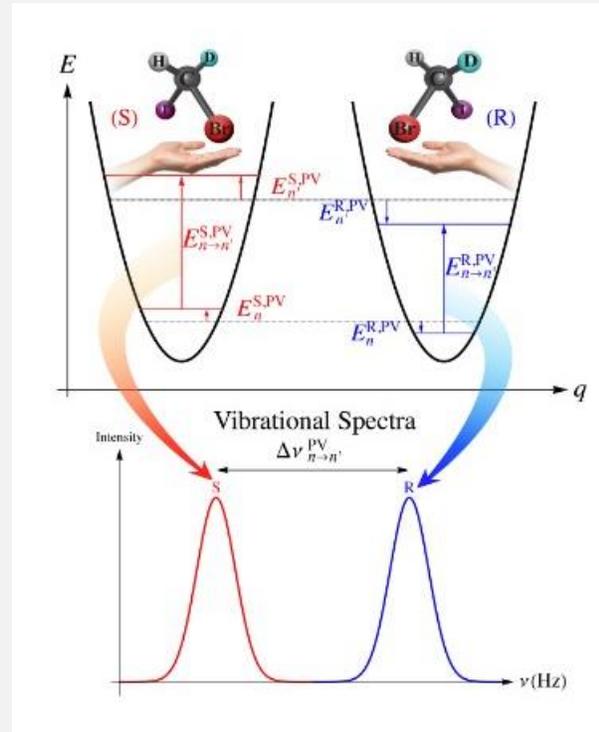


- So far, no detection!
- The search continues: in electronic transitions, NMR spectroscopy, and in **vibrational spectroscopy**

- Search for parity violation in vibrational spectroscopy.

$$E_n^{\text{PV}} = \langle n | V^{\text{PV}}(q) | n \rangle$$

$$\Delta\nu_{m \rightarrow n}^{\text{PV}} = \frac{2}{h}(E_n^{\text{PV}} - E_m^{\text{PV}})$$



- Measure $h\nu_L - h\nu_R$
- Measurements performed at Laboratoire Physique des Lasers (LPL), on the C-F stretch vibration in CHFCIBr

- **Search for parity violation in vibrational spectroscopy.**

- Measure $h\nu_S - h\nu_R$

- Measurements performed at Laboratoire de Physique des Lasers (LPL), on the C-F stretch vibration in CHFCIBr

- Upper limit of

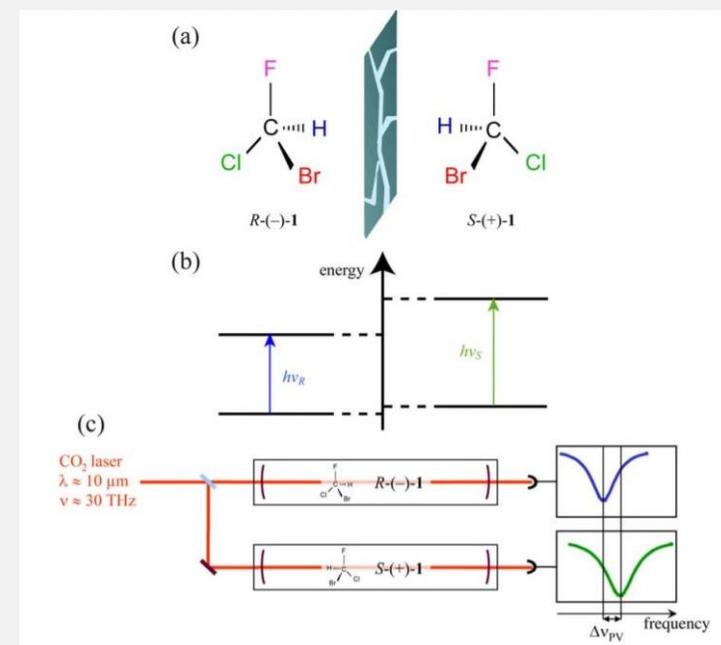
$$\frac{\Delta\nu_{PV}}{\nu} = 2.5 \times 10^{-13}$$

Progress toward the first observation of parity violation in chiral molecules by high-resolution laser spectroscopy†

Benoît Darquié, Clara Stoeffler, Alexander Shelkovnikov, Christophe Daussy, Anne Amy-Klein, Christian Chardonnet, Samia Zrig, Laure Guy, Jeanne Crassous... See all authors

First published: 13 September 2010 | <https://doi.org/10.1002/chir.20911> | Citations: 95

18 C. Daussy, T. Marrel, A. Amy-Klein, C. Nguyen, C. J. Bordé and C. Chardonnet, *Physical Review Letters*, 1999, 83, 1554.



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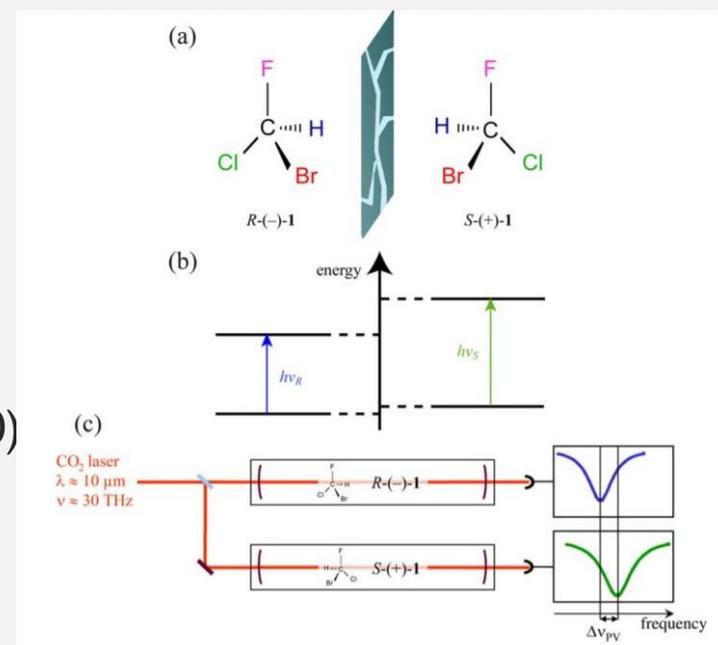
- Upper limit of

$$\frac{\Delta\nu_{PV}}{\nu} = 2.5 \times 10^{-13}$$

- Theoretical estimates of the effect $\sim 10^{-17}$

(Phys. Rev. Lett. **84**, 3807 (2000); Phys. Rev. A **71**, 012103 (2005);

Phys. Rev. A **103**, 042819 (2021); Phys. Rev. Lett. **125**, 123004 (2020)



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- **Search for parity violation in vibrational spectroscopy.**
- Better candidate molecule needed!
- Shopping list:
 - Stable, commercially available
 - Can be separated into pure enantiomers
 - Can be brought into the gas phase
 - Should contain heavy elements (absolute PV energy predicted to scale as Z^5)
 - Should have strong vibrational transitions in the range of the lasers (4-15 μm)
- Relative effects of the order 10^{-15} can now be detected at LPL

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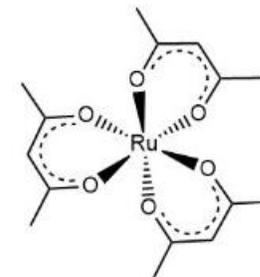
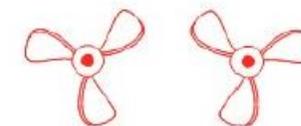


Fig. 1 Chemical structure of Δ -Ru(acac)₃

a) Propeller-like chirality



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- **Relative effects of the order 10^{-15} ?**

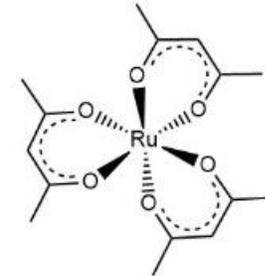
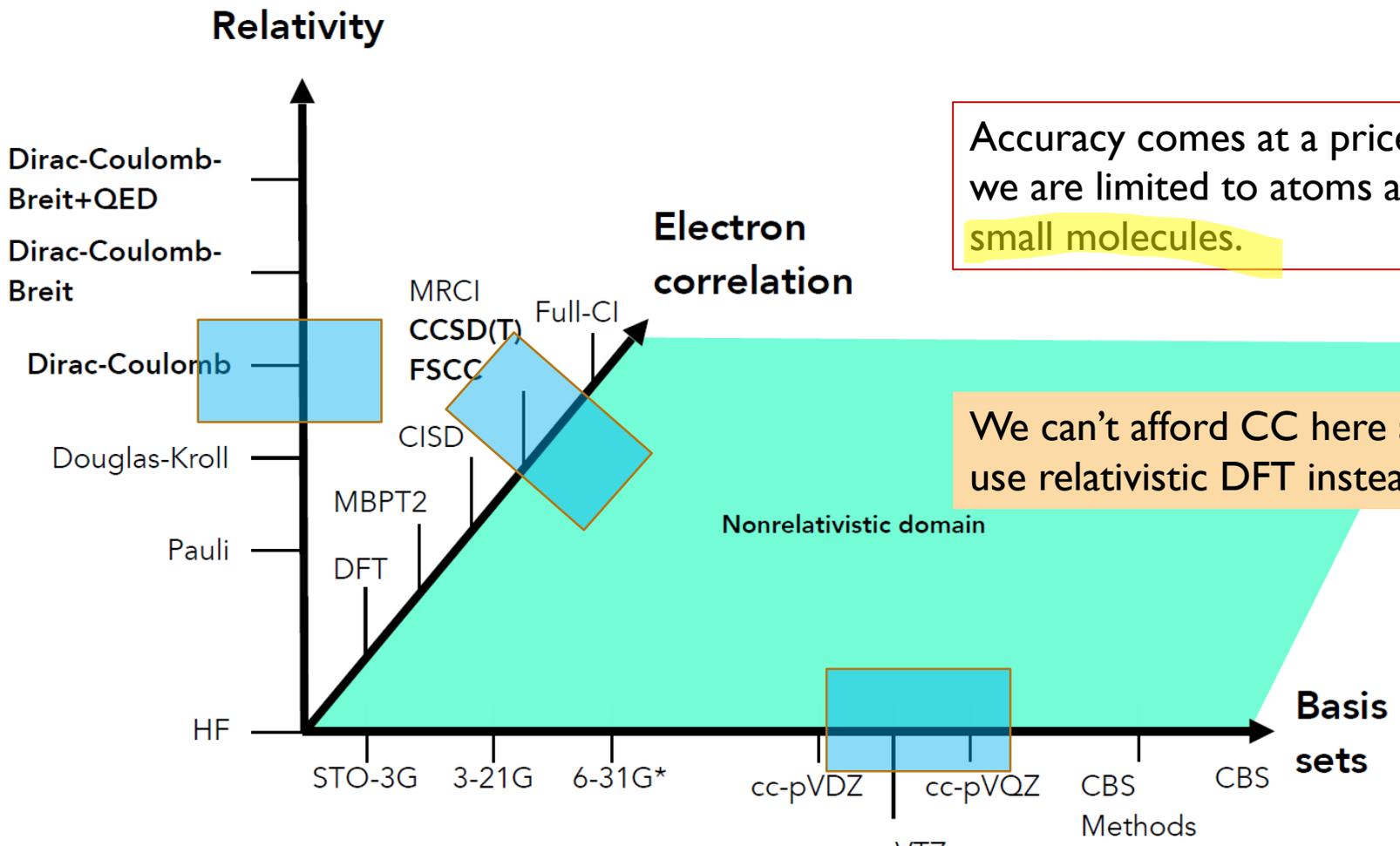


Fig. 1 Chemical structure of Δ -Ru(acac)₃

a) Propeller-like chirality





- Search for parity violation in vibrational spectroscopy.
- Better candidate molecule needed!
- Relative effects of the order 10^{-15} ?
- Relativistic calculations, DFT method

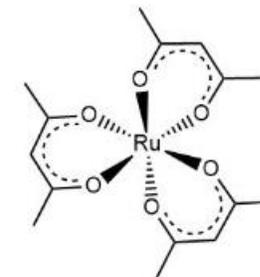


Fig. 1 Chemical structure of Δ -Ru(acac)₃

mode no.	$\nu(\text{cm}^{-1})$	IR int.(KM/mol)	$\Delta\nu_{PV}$ (mHz)	$ \Delta\nu_{PV}/\nu $
17	182.0	0.009	-448.7	7.22E-14
19	200.8	1.718	-297.9	4.57E-14
20	222.9	0.065	279.2	3.80E-14
29	327.2	7.884	325.3	3.40E-14
52	952.6	9.564	-30.4	1.04E-15
53	953.7	1.793	-33.0	1.13E-15
100	1586.0	453.5	-83.0	1.70E-15
102	1612.3	44.22	-110.9	2.25E-15

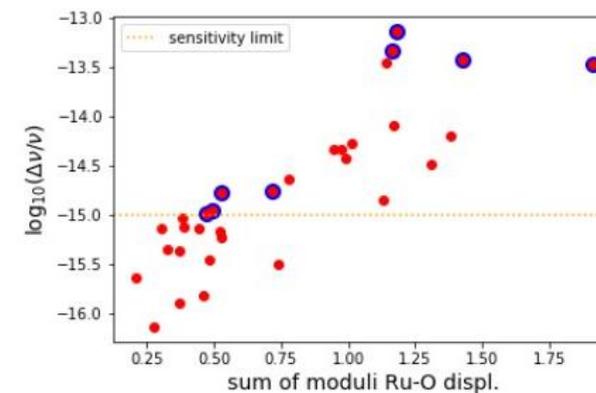


Fig. 3 Plot of all calculated relative parity violating frequency shifts ($\Delta\nu_{PV}/\nu$) of several vibrational transitions in Ru(acac)₃ as a function of the indicator (sum of moduli of Ru-O displacement, see text). The larger dots highlighted in blue correspond to the modes shown in Table 1, and

- Search for parity violation in vibrational spectroscopy.
- Better candidate molecule needed!
- Relative effects of the order 10^{-15} ? Yes!
- Relativistic calculations, DFT method

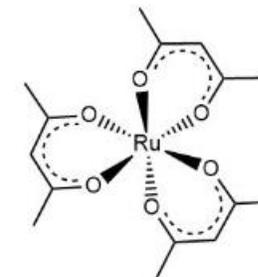


Fig. 1 Chemical structure of Δ -Ru(acac)₃

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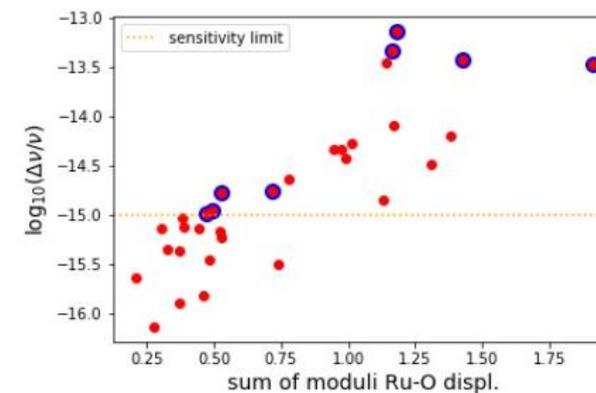


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- Even better: $\text{Os}(\text{acac})_3$, effects of 10^{-13}
- Scaling better than Z^5 predicted for total PV energies
- Practical challenges to be solved

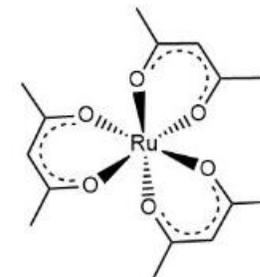


Fig. 1 Chemical structure of $\Delta\text{-Ru}(\text{acac})_3$

mode Os	corresp. mode Ru	overlap	$\nu_{PV}(\text{Os})$ (Hz)	$\nu_{PV}(\text{Os})/\nu_{PV}(\text{Ru})$	ν (cm^{-1})	IR int.(KM/mol)	$ \Delta\nu/\nu $
16	17	0.996	-9.72	21.7	191.4	0.091	1.51E-12
19	19	0.961	-9.59	32.2	210.9	2.356	1.44E-12
20	20	0.948	4.30	15.4	224.2	0.013	5.86E-13
29	29	0.838	3.09	9.5	307.6	2.392	3.39E-13
52	52	0.830	-1.47	48.4	952.2	0.464	5.00E-14
53	53	0.831	-1.32	40.1	954.0	1.272	4.48E-14
100	100	0.955	-1.04	12.7	1562.5	245.6	2.16E-14
102	102	0.985	-0.31	3.6	1589.2	102.2	6.29E-15

Table 2 The PV shift of vibrational modes in $\text{Os}(\text{acac})_3$ and a comparison to $\text{Ru}(\text{acac})_3$. Modes in the two compounds were matched to each other according to their overlap as defined in section 3.

	VIII B	VIII B	VIII B	II			
26	Fe Iron 55.845	27	Co Cobalt 58.933194	28	Ni Nickel 58.6934	29	C Copper 63.546
44	Ru Ruthenium 101.07	45	Rh Rhodium 102.90550	46	Pd Palladium 106.42	47	A Silver 107.865
76	Os Osmium 190.23	77	Ir Iridium 192.222	78	Pt Platinum 195.084	79	A Gold 196.967
108	Hs Hassium (269)	109	Mt Meitnerium (278)	110	Ds Darmstadtium (281)	111	R Roentgenium (282)

- **Search for parity violation in vibrational spectroscopy.**
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Marit Fiechter (now ETH Zurich)



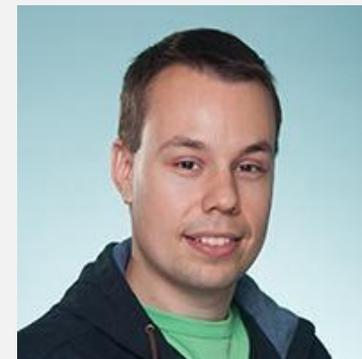
Benoît Darquié, LPL



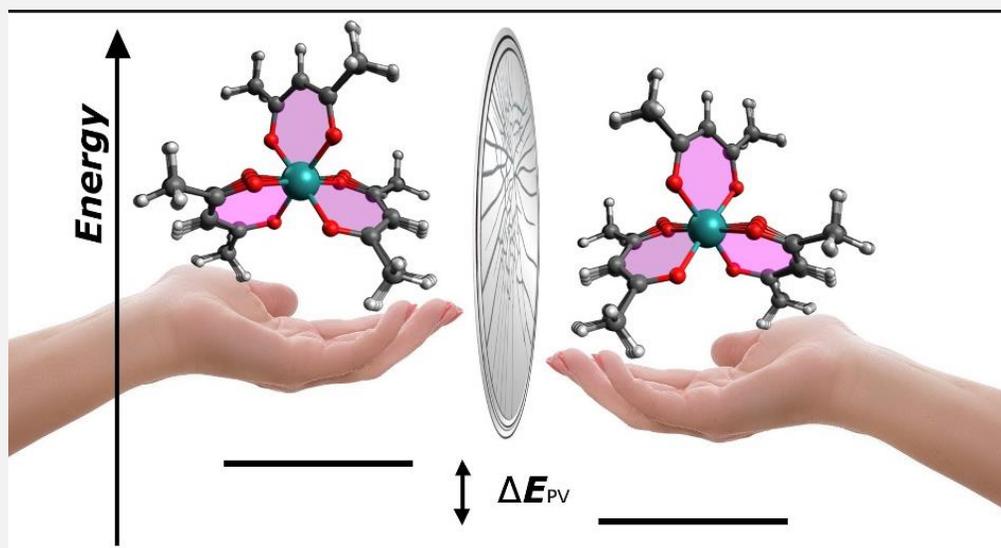
Jeanne Crassous, Rennes



Pi Haase

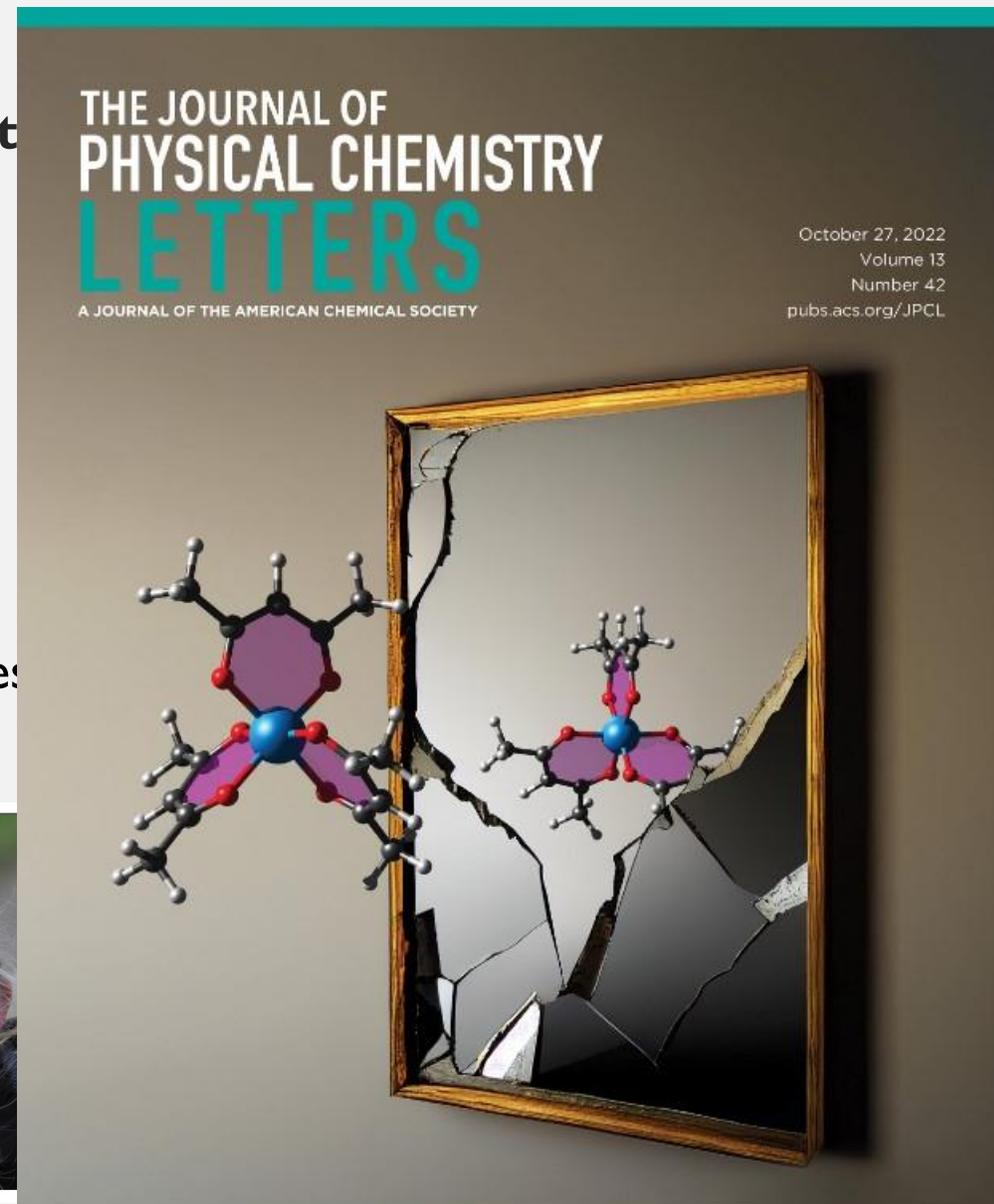


Lukas Pasteka, Bratislava



- Even better: $\text{Os}(\text{acac})_3$, effects of PV
- Scaling better than Z^5 predicted for total PV energies

spect



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Letter

Toward Detection of the Molecular Parity Violation in Chiral $\text{Ru}(\text{acac})_3$ and $\text{Os}(\text{acac})_3$

Marit R. Fiechter, Pi A. B. Haase, Nidal Saleh, Pascale Soulard, Benoît Tremblay, Remco W. A. Havenith, Rob G. E. Timmermans, Peter Schwerdtfeger, Jeanne Crassous, Benoît Darquié, Lukáš F. Pašteka, and Anastasia Borschevsky*

Cite This: *J. Phys. Chem. Lett.* 2022, 13, 10011–10017

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Benoît Darquié, LPL

Jeanne Crassous, Rennes

Pi Haase

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- **Other fun candidates:**

- From simple:

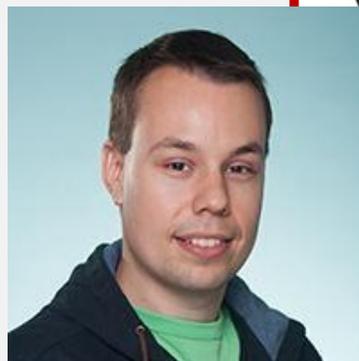
- **CHDBrI⁺**



Eduardus



Yuval Shagam, Technion



Lukas Pasteka, Bratislava

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Large vibrationally induced parity violation effects in CHDBrI⁺

Eduardus,^a Yuval Shagam,^b Arie Landau,^b Shirin Faraji,^c Peter Schwerdtfeger,^d Anastasia Borschevsky^a and Lukáš F. Pašteka^{*ae}

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RESEARCH ARTICLE | SEPTEMBER 19 2023

Chiral molecule candidates for trapped ion spectroscopy by *ab initio* calculations: From state preparation to parity violation

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Arie Landau^a; Eduardus^b; Doron Behar^b; Eliana Ruth Wallach^b; Lukáš F. Pašteka^b; Shirin Faraji^b; Anastasia Borschevsky^b; Yuval Shagam^c

- Other fun candidates:

- To complex:

- Pt(Me-BPCH)

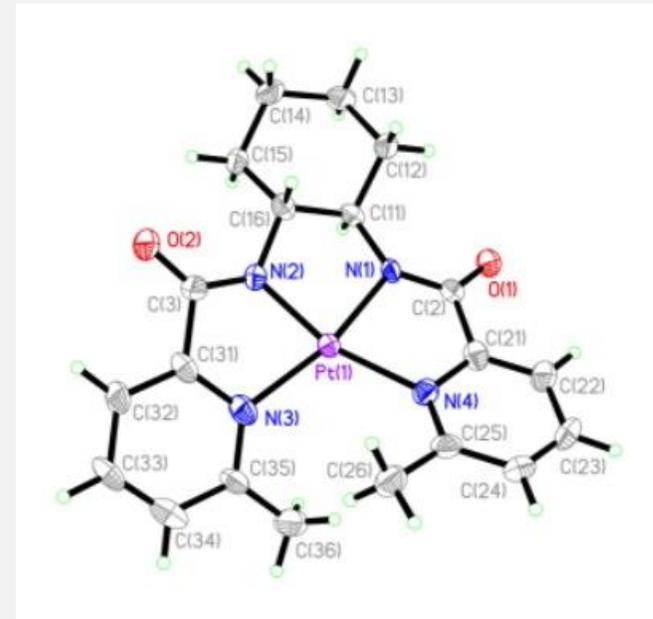
arXiv > physics > arXiv:2509.26407v1

Physics > Chemical Physics

[Submitted on 30 Sep 2025]

Chiral Pt(Me-BPCH): Synthesis and theoretical investigation of parity violation sensitivity

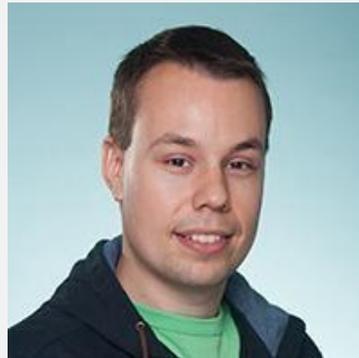
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Eduardus



Scott Bohle, McGill



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